Alma Mater Studiorum – Università di Bologna

DOTTORATO DI RICERCA IN SCIENZE DELLA TERRA

Ciclo XXVII

Settore Concorsuale di afferenza: 04/A1 Settore Scientifico disciplinare: GEO/07

A 300 million year-long history.

The metamorphic evolution of the Northern Apennine Variscan basement

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Esame finale anno 2015

ABSTRACT

Collision zones are the loci on the Earth where large portions of continental crust of different ages and origin are processed at different depths and temperatures. For these reasons, the exhumed slices of continental crust that is commonly found in collisional orogens record the overprinting relations between different stages of metamorphic (re)crystallization and deformation. Metamorphic petrologists investigate chemical and structural variations from the outcrop to the micron scale in an effort to better understand the petrogenetic processes occurring in collision zones.

The Variscan basement of Northern Apennines (Northern Italy) is an example of a continental crust portion affected by a complex polymetamorphic evolution.

The thesis investigates the metamorphic evolution of this basement occurring in the Cerreto Pass, in the bedrock reached by the Pontremoli well, and in the Pisani Mountains. The study comprised fieldwork in the Cerreto Pass, sampling, petrography and microstructural analysis, determination of the bulk rock and mineral composition, thermodynamic modelling, conventional mineral-mineral geothermobarometry, monazite chemical dating and Ar/Ar dating of muscovite.

The selected samples from the Cerreto Pass are mylonitic micaschist with a white mica + chlorite \pm biotite mylonitic foliation wrapping around plagioclase, quartz and garnet porphyroblasts. Mg-rich prekinematic white mica and fish-shaped white mica are also present. The samples from the Pontremoli well are medium grained micaschist with a white mica and chlorite schistose fabric surrounding garnet porphyroblasts. Coronitic microstructures around inherited monazite grains in this rock consist of concentric shells of apatite + Th-silicate, allanite and epidote. Phyllites overlying the micaschist were also studied: they consist of a white mica and chlorite schistosity partially obliterating a previous foliation. The studied samples from the Pisani Mountains are chloritoid-bearing phyllites with abundant hematite bands and disseminated grains.

In late Carboniferous the Cerreto micaschists underwent a high-pressure event at 430-480°C and 11-13.5 kbar, probably due to the Variscan thickening stage. Some similarities occur between the Variscan evolution in Northern Apennines and that described for other fragments of the Southern Variscan belt, as suggested by previous petrological works on the Variscan basement of Northern Apennines.

The Pontremoli samples, by contrast, record a different metamorphic history. Detrital monazite yields U-Th-Pb chemical ages of 294 ± 5 Ma, interpreted as a late-Variscan thermal stage unrelated to the reconstructed metamorphic evolution. The Pontremoli micaschist experienced a prograde evolution at 500-600°C and 5-7 kbar, followed by a pressure peak at 520°C at 8 kbar and then a nearly isothermal cooling to 500°C at 2 kbar. The overall anticlockwise shape of the *P-T-D* path is different from the available metamorphic histories of the pre-Mesozoic basement of the Northern Apennines, and may be due to the post-Variscan shear zone tectonics or to the Alpine orogeny.

A shearing stage affected the Cerreto mylonite at 190.7 ± 1.3 Ma (Ar/Ar isotopic age of Na-rich mylonitic muscovite) and was accompanied by a strong ductile deformation at 450-500°C and 6-7.5 kbar. This age suggests that the exhumation of the Cerreto mylonite took place during the rifting stage of the Adria margin before the opening of the Ligurian-Piedmont ocean.

The high-pressure greenschist-facies conditions (475 °C and 9-10 kbar) preserved in the chloritoid-bearing phyllites from the Pisani Mountains indicate that, from the late Oligocene onwards, the Northern Appenine basement attained similar P-T conditions as the overlying metasediments (Verrucano).

As a whole the results on the Cerreto, Pontremoli and Pisani Mountains samples allowed to reconstruct a long-lasting metamorphic history straddling the Variscan and Alpine orogenesis.

The study of the Variscan basement of Northern Apennines allowed also to contribute to a better understanding of some general petrological issues generally found in low- to medium-grade metapelites from orogenic settings.

(i) Middle-grade mylonitic micaschist usually contains zoned minerals (e.g. garnet and white mica) in different microstructural sites. With suitable samples it is possible to reconstruct a P-T-D path as much complete as possible by combining microstructural analysis and thermodynamic modelling. Prekinematic white mica may preserve Mg-rich cores related to the pre-peak stage. Mn-poor garnet rim records the peak metamorphism. Na-rich mylonitic white mica, the XFe of chlorite and the late mineral assemblage may constrain the retrograde stage.

(ii) Metapelites and particularly orthogneiss may contain coronitic microstructures of apatite + Th-silicate, allanite and epidote around unstable monazite grains. In general, it is still not clear what are the P-T variations responsible for the

development of this kind of corona. In the Pontremoli micaschist, chemistry and microstructure of Th-rich monazite relics surrounded by an apatite-allanite-epidote coronitic microstructure suggest that monazite mineral was inherited and underwent partial dissolution and replacement by REE-accessory minerals at fluid-present peak metamorphic conditions. Element partitioning between garnet and accessory minerals and a detailed microstructural study on garnet inclusion mineralogy (which comprises xenotime in the garnet core, allanite in the outer core and epidote in the rim) allowed to link the crystallization of REE-minerals to the garnet growth in order to reconstruct the reaction history. The replacement of inherited monazite by apatite, allanite and epidote occurred during a prograde evolution at 500-600°C and 5-7 kbar.

(iii) Fish-shaped white mica is not always a (prekinematic) mica-fish. Observed at high-magnification BSE images it may consist of several white mica grains defining a fish-shaped asymmetric foliation boudin formed during a mylonitic stage. Hence, the asymmetric foliation boudin is a suitable microstructure to obtain geochronological information about the shearing stage (190.7 \pm 1.3 Ma in the Cerreto mylonite case study). Differently from the asymmetric foliation boudins, mylonitic foliation may include relics of earlier events as evidenced by the ages of 374 and 315 Ma related to the Variscan orogenic cycle.

(iv) Thermodynamic modelling is an useful tool to constrain the *P*-*T* conditions experienced by a metamorphic rock. However inadequate assumption (e.g. neglecting important components as C, O₂) may result calculation that reproduces neither the observed mineral assemblage nor the measured mineral compositions. The effects of different Fe₂O₃ bulk contents were investigated on the calculated phase equilibria of low-temperature/intermediate-pressure metasedimentary rocks from the Pisani Mountains. Thermodynamic modelling of a chloritoid-bearing hematite-rich metasedimentary rock fails to reproduce the observed mineral compositions when the bulk Fe₂O₃ is neglected or determined through titration. The mismatch between observed and computed mineral compositions and assemblage is resolved by tuning the effective ferric iron content by $P-XFe_2O_3$ diagrams. The introduction of ferric iron affects the stability of the main rock-forming silicates that often yield important thermobaric information.

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A TIME TRAVEL

THROUGH THE EVOLUTION OF THE

VARISCAN BASEMENT OF NORTHERN APENNINES



CHAPTER 1

A TIME TRAVEL THROUGH

THE EVOLUTION OF THE VARISCAN BASEMENT OF NORTHERN APENNINES

1.1 INTRODUCTION

Collision zones are those areas of the Earth where orogenic belts form. They are regions of fundamental petrogenetic processes related to the growth and consumption of lithosphere. The transformations that submersed rocks undergo during the collision are also linked to the long-term biogeochemical cycle of elements (C-O-H-N) that affects life development (Evans, 2011). The study of metamorphic rocks is thus fundamental to investigate the processes occurring throughout the collision. In principle, metamorphic minerals and microstructures may record different stages of the orogenic events: when a crustal slab undergoes below the overhanging plate margin the values of Pressure-Temperature (P-T) increase (prograde metamorphism); then rocks reach peak conditions (maximum temperature); finally the exhumation of the buried rocks is related to a P-T decrease (retrograde metamorphism). In practice, the last stages of metamorphism are better preserved whereas the record of the prograde metamorphism is not commonly recognizable. One challenge is indeed to reconstruct complete metamorphic evolutions comprising also pre-peak stages. Then the task of metamorphic petrologists is to decipher the information and estimate the physical and chemical conditions during the metamorphic evolution. The P-T-t-D-X (Pressure-Temperature-time-Deformation-composition) path experienced by metamorphic rocks provides fundamental constrains for the geodynamic models.

In the study of collisional basement metamorphic petrology has to face some firstorder problems:

- 1. Textural ad chemical modifications occurring during the retrograde evolution hamper the reconstruction of the prograde metamorphic history.
- 2. Difficulties also arise when the petrologic record has to be correlated with age determinations derived from major and accessory minerals. Monazite is one of the main accessory minerals useful for obtaining geochronological information. Therefore much effort is dedicated to connect monazite U-Th-Pb chemical or isotopic ages with *P-T* constraints obtained from mineral assemblages. A

Figure of the previous page: River "Canale dell'Acqua Torbida" in the Cerreto Pass. The Variscan basement outcrops are in river slope on the left.

common approach relies on the Y partitioning between garnet and monazite (Vance *et al.*, 2003; Williams *et al.*, 2007), but it requires that monazite grows during metamorphic evolution. However, clear breakdown microstructures of monazite testify that monazite can also be consumed during metamorphic evolution (Janots *et al.*, 2006; Rasmussen & Muhling, 2007) and the link between monazite and P-T sensitive minerals becomes more complicated. The breakdown reactions of monazite have been observed but their thermobaric conditions are still debated.

- 3. Another key mineral in metamorphic petrology is **potassic white mica**, because it is a fabric-forming major mineral, *P-T* sensitive, and also useful for geochronologic purpose. However potassic white mica occurring in different microstructural positions can provide complex age patterns at thin section scale.
- 4. In calculating the thermobaric conditions by thermodynamic modelling (i.e., using pseudosections), specific problems are: (i) inadequate assumption (neglecting important components as C, CO₂, O₂, considering H₂O in excess, indeterminateness in the fluid activity); (ii) uncertainty related to the solid-solutions models and the thermodynamic dataset; (iii) selection of a reliable representative chemical system; (iv) errors in the effective bulk composition used as input for the calculation of the pseudosection.

The thesis aims to propose innovative approaches to tackle some of the abovementioned petrological questions. They are:

- how to reconstruct a complete polyphase *P-T-D* path comprising pre-peak stages through the combination of thermodynamic modelling and microstructural analysis (chapter 2);
- how to relate geothermobarometry and geochronology when monazite is inherited. The topic deals also with estimating the *P-T* conditions of the reactions involving monazite breakdown in medium-grade metapelites (chapter 3);
- investigating the type of geochronological record provided by potassic white mica in different mylonitic microstructures (e.g. mylonitic foliation and asymmetric foliation boudins; chapter 4);
- 4. exploring the influence of ferric iron on phase assemblages and composition in greenschist-facies metamorphic rocks (**chapter 5**).

In summary, the application of modern petrological methods allowed to **investigate the potentialities and limits of the research advancements in the study of collisional basements.**

1.2 THE NORTHERN APENNINES

The Northern Apennines are an example of collisional orogen that involved both Mesozoic-Tertiary sedimentary sequence and Variscan basement. While the Alpine tectono-metamorphic history of the Triassic Verrucano Group, a metasedimentary sequence tectonically stacked onto the Variscan basement, is well-known (Theye *et al*, 1997; Giorgetti *et al*, 1998; Jolivet *et al*, 1998; Franceschelli & Memmi, 1999; Franceschelli *et al*, 2004), **the state of knowledge on the Variscan basement is poor and modern studies applying advanced petrologic methodologies lack.** The difficulties to study the Variscan basement of Northern Apennines arise from (i) poor exposure of scattered outcrops and (ii) a strong Alpine overprint on Variscan features (Franceschelli *et al.*, 2004). For these reasons, the Variscan *P-T-t* path is less constrained than the Alpine evolution or, when *P-T* data are available (Bertini *et al.*, 1994; Molli *et al.*, 2002), the reconstructed history refers to the late, exhumation-related stages.

This **knowledge gap** attracted our interest and one of the outcome of this research is to provide better constraints on the pre-Alpine P-T-t path and compare the obtained new paths with others retrieved from several Mediterranean Variscan basements.

Since in Southern Tuscany the thermometamorphism related to the emplacement of Pliocene-Quaternary granitoids has obliterated the earlier stages (Elter & Pandeli, 1990; Franceschini, 1998; Musumeci *et al.*, 2002; Pandeli *et al.*, 2005; Musumeci *et al.*, 2011), only the Variscan basement outcropping in Northern Tuscany and Liguria was selected. The occurrences are Punta Bianca, Cerreto Pass, Pontremoli 1 well, Apuane Alps and Pisani Mountains (Fig. 1.1).

In order to obtain useful *P-T* information through thermodynamic modelling, the investigation was limited to medium-grade metasediments, which contain suitable low-variance parageneses. For this reason the greenschist-facies Variscan basement in Apuane Alps and Punta Bianca could not be analysed with the above method. The hematite-rich chloritoid-bearing phyllites from the Monti Pisani, and the garnet micaschist samples from the Cerreto Pass and the Pontremoli well resulted suitable for a study involving phase diagram calculations (see below). The rocks sampled in those

three areas recorded different stages of the metamorphic evolution, so it was possible to combine all the information to reconstruct **a long-lasting metamorphic history**.

The thesis is indeed a time travel through the evolution of the Variscan basements of Northern Apennines, from the first recognizable metamorphic signature of Variscan times to the Alpine metamorphic overprint. The results reveal, for the first time in the study of these basements, a complex history between the Variscan and the Alpine orogenic cycles.

Therefore, this research deals with the following regional goals:

- to unravel the Variscan tectono-metamorphic history including the pre-peak, peak and exhumation conditions and establish possible correlations with other circum-Mediterranean Variscan basements (chapter 2);
- to outline the metamorphic history between the Variscan and the Alpine orogeneses (chapter 3);
- 3. to constrain the age of the shearing event in the Cerreto mylonites (chapter 4);
- 4. to determine the peak conditions of the Alpine metamorphism (chapter 5).

Although the results of this research line contribute to a better understanding of the metamorphic evolution of the Variscan basement of Northern Apennines, open regional questions still remains and are reported in "Conclusions" (**chapter 6**).

1.2.1 Geological framework

In Italy fragments of Variscan basement (Fig. 1.2) occurs in the Alps, in the Corsica-Sardinia block, in the Calabria-Peloritani orogen and in Northern Apennines (Boriani *et al.*, 2003).

The Variscan basement of Northern Apennines is the less studied among these fragments. Differently from the Alps, Corsica-Sardinia and the Calabrian-Peloritan orogen, small extent and scattered exposures characterize the Variscan basement of Northern Apennines. Furthermore only few petrological data are available for this basement (Molli *et al.*, 2002; Franceschelli *et al.*, 2004).

The Northern Apennines (Figg. 1.3a, b) are a thrust and fold belt formed during the Eward collision between the Corsica-Sardinia block (Europe) and the Adria microplate (Africa) during the last 30 Ma (Vai & Martini, 2001; Molli, 2008). They consist of oceanic (Liguride Units) and continental (Tuscan Units and Tuscan Metamorphic Units) domains. Each unit is overlain by neogenic successions, outcropping especially in South Tuscany.



Figure 1.1 Localization of Pontremoli, Cerreto Pass and Pisani Mountains (in red) in the relief map of Northern Apennines (Google Maps). The main thrusts in Italy are from Patacca & Scandone (1989), while the tectonic setting of Northern Apennines is from Bigi *et al.* (1983) and Molli (2008).

The tectonic units are (from top to bottom; Elter, 1975; Fig. 1.4a):

- the Epiligurian wedge-top sediments
- Liguride Units: ophiolites from the Jurassic Ligurian-Piedmontese ocean;
- Tuscan Units: low-grade metamorphic or sedimentary sequences deposited in late Triassic-Tertiary derived above the Adria crust;
- Tuscan Metamorphic Units: metamorphic sequence representing the Adria continental crust, consisting of a Variscan pre-Carboniferous basement and Late Carboniferous-Oligocene metasedimentary rocks;
- basement and sedimentary rocks deriving from the Umbro-Marchigian and the Padan domains.

Chapter 1



Figure 1.2 Distribution of the Variscan Massifs in Italy (from Padovano *et al.*, 2012, modified from Vai, 2001). See the following page for further details. Alps (white areas): CA-SV-AZ, Calizzano-Savona-Arenzano M.; Ag, Argentera M.; PE, Pelvoux M.; BE, Belledonne M.; GP, Gran Paradiso M.; AR-MB, Aiguilles Rouges-Mont Blanc M.; SL, Sesia Lanzo Zone; IZ, Ivrea Zone; AA-GO, Aar-Gotthard M.; MMZ, Monte Muggio Zone; NEOB, North-East Orobic Basement; SVR, Silvretta Thrust Sheet; UZ, Ulten Zone; CM, Cima Mezzana; OTZ, Oetzal Zone; TW, Tauern Window; PO, Pohorje M.; Northern Apennines (grey areas): Pf, Neogenic Portofino Conglomerates; Ce, Cerreto area; Pb, Punta Bianca; AAp, Apuan Alps; Mp, Monti Pisani; Ld, Larderello / Travale geothermal fields; MRR, Monticiano-Roccastrada Ridge; Amt, Amiata geothermal field; Mro, Monti Romani; El, Eastern Elba Island. A areas filled with oblique lines: MTM, Maures–Tanneron M.; CO, Corsica M.; SA, Sardinia M. Areas filled with triangles: SI, Sicily (Peloritani M.); CAL, Calabria (Aspromonte, Sila, Serre M.). Thick black line: Insubric Line.



Figure 1.3 Northern Apennine setting (modified from Clemenzi *et al.*, 2014 and Molli, 2008): (a) tectonic map of the Northern Apennines; (b) cross-section.

The Variscan basement outcrops in Punta Bianca, Apuan Alps, Pisani Mountains, the Monticiano-Roccastrada Ridge, eastern Elba Island, Romani Mountains, and was also found in the subsurface of the Larderello-Travale geothermal region and of the Pontremoli area.

Regional correlations among the different Variscan basement occurrences are difficult, because of the scarcity and bad-quality of the outcrops and the strong Alpine overprint (Pandeli *et al.*, 1994).

This basement shows evidence of a pre-Alpine metamorphism, which has been ascribed to the collisional stages of the Variscan Orogeny (Conti *et al.* 1993), preserved as relics because of the strong Alpine metamorphism superposition (Franceschelli *et al.*, 2004).

In the Larderello subsurface (Southern Tuscany), the Variscan basement consists of three different units (from top to bottom; Bertini *et al.*, 2006; Fig. 1.3b):

- Phyllitic and quartzitic complex (greenschist-facies Variscan metamorphism and Alpine overprinting; Pandeli *et al.*, 2005);
- Micaschist complex (Variscan metamorphism and Alpine overprinting; Pandeli et al., 2005);
- Gneiss complex (Variscan metamorphism at P = 7 kbar at T=500-600°C; Bertini *et al.*, 1994).

Pandeli *et al.* (2005) correlate the rocks from the Cerreto Pass and the Pontremoli to the Micaschist Complex, and the metasediments from the Pisani Mountains to the Phyllitic and quartzitic complex.

Chapter 2, 3, 4 contains specific geological frameworks and tectonic maps of the selected study areas: the readers are referred to the corresponding chapters.



Figure 1.4 Tectonic setting of Northern Apennines: (a) relationships among the tectono-stratigraphic units in Northern Apennines (from Vai & Martini, 2001); (b) metamorphic units in the Variscan basement beneath the Larderello geothermal field (from Bertini *et al.*, 2006).

1.2.2 History of research on the Northern Apennine Variscan basement

In the '60s-70s Variscan metamorphism was envisaged in the Pre-Triassic sequences of Northern Apennines (Borsi et al., 1966; Rau & Tongiorgi, 1974; Cocozza et al., 1974; Vai, 1978). Rau & Tongiorgi (1966, 1974) studied the field relationships and the stratigraphy of the Paleozoic succession in the Pisani Mountains: they recognized both Variscan ("Formazione delle Filladi e Quarziti listate di Buti") and post-Variscan units ("Formazioni degli Scisti di San Lorenzo" and "Conglomerati e Brecce d'Asciano"). The Variscan unit was overlain by a discordance formed in Late Carboniferous. The Rb-Sr age determination on bulk-rock from the Monti Pisani phyllites of Borsi et al. (1967) revealed 275 ± 12 Ma. Only few years later, Del Moro et al. (1982) dated with the Ar-Ar method a muscovite associated with andalusite in the Micaschist Complex from the Larderello subsurface. The obtained age of 285 ± 11 Ma has been interpreted as a late-Variscan thermal stage. The first reconstructions related to the Variscan basement were published in late '70-early '80 (Vai, 1978; Bagnoli et al., 1979; Tongiorgi & Bagnoli, 1981). The studies adopted basically a biostratigraphic approach and correlation with the metamorphic evolution of Sardinia. The paper by Kligfield et al. (1986) is a milestone in the research history of Northern Apennines. They performed K-Ar and ⁴⁰Ar-³⁹Ar on potassic white mica separates of Paleozoic to Oligocene metasedimentary rocks present from the Alpi Apuane. The mixed ages obtained in the Variscan basement suggested the overprinting of the Alpine events on the Variscan metamorphism. The age interval of 27-10 Ma was attributed to the Alpine evolution. Rau (1990) for the first time attempted to reconstruct the evolution of the Variscan basement from Late Paleozoic to Mid Triassic on the basis of the analysis of post-Variscan sedimentary basins. Conti et al. (1991a) proposed the first complete framework of the Variscan and Alpine events. They assert that all the Paleozoic units in Northern Apennines show a low-grade Alpine metamorphism and only some of them contain greenschist- or amphibolites-facies Variscan relics. In particular, Conti et al. (1991b, 1993) outlined the superposition of the Alpine events on the Variscan structures in the Apuane Alps. A new reappraisal of the geodynamic evolution affecting the Variscan basement is provided by Rau (1993). After the Variscan collision (Late Carboniferous) the Variscan chain underwent transcurrent dextral megashear zones accompanied by alternating transpessive and trastensive tectonics. These shear zones caused the fragmentation of the Variscan belt and the formation of shallow marine basins. In Permian and Lower-Triassic times the transcurrent tectonics continued. In Middle- and Upper-Triassic the

extension prevailed, leading to the definitive separation of the Tuscan Variscan basement from the "stable" Europe. The Tuscan Variscan basement became indeed a passive margin of the Adria promontory (Rau, 1993). Bertini et al. (1994) proposed some P-T estimates for the Variscan metamorphism for the Gneiss Complex in the Larderello subsurface. Pandeli et al. (1994) proposed a revised stratigraphy of the Paleozoic stratigraphic sequences in Northern Apennines. A study on the ore deposits in the Variscan basement from the Apuane Alps is given by Dini et al. (1995, 2001). Franceschini (1998) attributed the low-pressure high-temperature metamorphism in the Larderello subsurface to the Pliocene-Quaternary thermometamorphism. Molli et al. (2002) analysed the major and trace elements composition of the amphibolites from the Cerreto Pass. Their data support an Enriched-type MORB derivation for the amphibolites protolith. Classical thermobarometry on the same samples indicate an initial medium-pressure amphibolites stage followed by a greenschist-facies metamorphism. They also performed Ar-Ar dating on horneblende separates, obtaining a Variscan age (312-328 Ma) for the metamorphism affecting the amphibolites. The geological features of the Larderello geothermal field are reported in Batini et al. (2003) and Bertini et al. (2006). A review of the knowledge on the Variscan and Alpine events in Northern Apennines was published by Franceschelli et al. (2004). Elter & Pandeli (2005) assessed a common evolution among the Maures Massif, Corsica-Sardinia Block, Northern Apennines from Early Carboniferous to the Late Carboniferous-Early Permian. The Authors suggested that the Posada Valley Zone of Sardinia might continue in the Tuscan Variscan basement. Pandeli et al. (2005) focussed on the Micaschist Complex and the Gneiss Complex of the Larderello area, proposing correlations with the Cerreto and Pontremoli Variscan basement. The high temperature-low pressure metamorphism in the Larderello subsurface was outlined by Rossetti et al. (2008). Leoni et al. (2009) estimated the P-T conditions of the metamorphism affecting the tectonic units outcropping in the Pisani Mountains. **Musumeci** et al. (2011) performed U-Pb dating on detrital zircon and ⁴⁰Ar-³⁹Ar dating metamorphic muscovite from the Variscan basement in the Elba Island. A on correlation of the exhumation history of the Variscan basement outcropping in the Alps, the Corsica-Sardinia-Maures-Tanneron Massif, the Calabria-Peloritani Orogen, and the Northern Apennines is also provided by Padovano et al. (2012). They found similarities corroborating the thesis that those massifs belonged to the same basement until Late Carboniferous time, when a regional dextral strike–slip shear zone (the East Variscan Shear Zone) led to their segmentation.

1.3 METHODS

The study is primary petrological, carried out through fieldwork in key outcrops, detailed petrography, determination of the mineral and rock compositions, geothermobarometry and geochronology.

A brief description of the used methods is presented below. In-depth accounts of the methods are reported in the following chapters dealing with the different case studies.

1.3.1. Fieldwork

A bibliographic research anticipated the fieldwork in order to select the most appropriate sites. The fieldwork regarded the Variscan basement in the Cerreto Pass, the only site with significant rock exposures. A geological survey allowed to determine the main structural and petrographic features of the mylonitic micaschist and amphibolite, the geometric relationships between the Variscan basement and the enclosing Triassic evaporites and the sampling of representative micaschist. One of the results of the fieldwork activity was the creation of a geological map reporting the existing outcrops of the Variscan basement. 34 samples were collected. Sampling was also carried on the hematite-rich phyllites from the Pisani Mountains (5 samples). Professor Molli (University of Pisa) and Professor Alessandra Montanini (University of Parma) kindly provided the micaschist recovered from the cuttings of the Pontremoli commercial well (8 samples) and some samples from the Cerreto Pass.

1.3.2 Petrography

Petrographic observations were carried out with a polarising optical microscope and the scanning electron microscope Philips 515B at the Dipartimento di Scienze Biologiche, Geologiche ed Ambientali (Università di Bologna). Additional observations with a Jeol JSM-6500F thermal Field emission scanning microscopy at INGV (Rome) were achieved.

Petrographic characterization and microstructural analysis were always performed on each sample to reconstruct the deformation and blastesis sequences and the metamorphic-deformation relationships. The porphyroblast-matrix geometric relations were particularly useful for this purpose.

1.3.3 Mineral and rock compositions

Mineral chemistry was determined through a CAMECA SX100 electron microprobe equipped with five wavelength dispersive spectrometers at the Institut für Mineralogie und Kristallchemie (Universität Stuttgart). X-ray maps of minerals were acquired to inspect the spatial distribution of selected major and trace elements. Point analyses were accurately chosen on the basis of their microstructural position. Compositional diagrams were then constructed to recognize compositional trends and variations. Rock compositions (major elements) were obtained by XRF analyses on pressed powder pellets using the Philips PW1480 and the Axios-Panalytical spectrometers at the Department of Biological, Geological and Environmental Sciences, Bologna University or by calculating the bulk-rock using mineral modes and compositions.

1.3.4 Geothermobarometry

Geothermobarometry is the branch of metamorphic petrology that permits the estimation of the *P*-*T* conditions of formation of metamorphic assemblages. The so-called conventional methods (e.g. mineral-mineral elemental exchange and/or net transfer reactions) have been recently replaced by thermodynamic modelling for calculating phase diagrams. *P*-*T* pseudosection (Connolly and Petrini, 2002; Powell and Holland, 2008) are isochemical phase diagram showing the stable paragenesis at different thermobaric condition for a fixed rock composition. Thermodynamic modelling allows also to calculate isothermal Pressure-Composition (*P*-*X*) diagrams or isobaric Temperature-Composition (*T*-*X* diagram), where the chemistry of the rock is a variable. This can be useful when dealing with metasomatic rocks and the composition of the fluid has to be known.

Another variable considered in metamorphic petrology is the time (t) of metamorphism. Through geochronological methods we can obtain the age of different stages in the metamorphic evolution and reconstruct P-T-X-t path. If the tectono-metamorphic sequence has been established through microstructural analysis, then the thermobaric and chemical constrains can be connected with deformation (D), thus defining P-T-X-t-D path. The usage of pseudosections (Williams & Jercinovic, 2012) has benefited from the improvement of the thermodynamic dataset (Berman, 1988; Holland & Powell, 1998), from the implementation of software that easily calculate phase diagrams (Connolly, 2005) and from the development of new activity-models for minerals (Holland & Powell, 1998; 2011). With pseudosections it is possible to predict the

evolution of the mineral assemblages and compositions during *P-T-X* variations (forward modelling). Another potentiality of the pseudosections is the possibility to quantify the variations in the *P-T-X* space of many rock properties, such as the mineral modes and compositions (isopleths), thermodynamic parameters, sound- and seismic-wave velocities. This PhD work often employs compositional isopleths, that are lines in a phase diagram indicating the calculated stable composition of a solid solution mineral, eventually changing with pressure and temperatures. By comparing the observed and computed mineralogy we are able to identify some thermobaric and chemical constraints and thus define a *P-T-X* path. Differently from the classical geotermobarometry that gives only one point (\pm uncertainties) within the *P-T* space, pseudosections report all the possible parageneses that can be generated with a given composition in a specific *P-T* window. Mineral assemblages that were stable in the earlier phases of the metamorphic evolution can be detected, even if they are no more present in the observed rock sample.

Medium-grade metapelites are suitable for the thermodynamic modelling approach because of their reactivity in response to relatively small changes in pressure and temperature or fluid composition and because reliable solid solution models are available (Spear, 1993; Massonne, in press).

In order to obtain *P*-*T* estimates during the PhD activity thermodynamic modelling was preferentially performed, accompanied by classical geothermobarometry when necessary.

1.3.5 Geochronology

The absolute dating of tectono-metamorphic events in crystalline basements relies on the measurement of elements and/or isotopes originated from the natural decay of U, Th and K included in minerals lattice. We used two geochronological techniques in order to achieve the ages of different stages of the metamorphic evolution: *in situ* U-Th-Pb chemical dating of monazite and Ar-Ar isotopic dating of potassic white mica.

Monazite (lanthanum-cerium light rare earth elements phosphate) is a mineral of geochronological interest for understanding metamorphic processes (Kelly *et al.*, 2006). It contains Th and U at a level detectable by the electron microprobe (Montel *et al.*, 1996), so it is possible to analyze tiny monazite microdomains directly on polished thin sections, thus preserving their textural position (William & Jercinovic, 2002). Furthermore monazite contains virtually no common Pb, so all the Pb analysed derives

from the Th and U decay. Before dating, the monazite has to be chemically and microstructurally characterized in order to establish if it is metamorphic or inherited and to link monazite to the reconstructed metamorphic history.

Potassic white mica can be dated by the 40 Ar- 39 Ar method that derives from the K-Ar dating. It is based on the decay of K to Ar (McDougall & Harrison, 1999). Through some extraction techniques (e.g. laser) it is possible to analyse the 40 Ar, that is generated by the K decay, and the 39 Ar, induced by the irradiation of K-bearing minerals in a nuclear reactor. This method permits to obtain age mapping of minerals in rock chips at high spatial resolution (50-100 µm; Di Vincenzo, 2004). Potassic white mica is not only a suitable mineral for 40 Ar- 39 Ar dating. Since the composition of potassic white mica changes in response to *P*-*T* variations, this mineral can provide also petrological information. Thus, potassic white mica is one of the main minerals that allows to link petrology and geochronology.

1.3.6 Linking geothermobarometry and geochronology

The connection between petrological and textural evolution with the age dating is not a trivial problem (Vance *et al.*, 2003). Even if zircon and monazite (tipical minerals targeted for age dating work) can supply very precise ages, it is not easy to interpret when these minerals (re)crystallize in relation to the evolution of major minerals that define *P-T-D* history. This is because understanding when accessory phases grow with respect to changing pressure and temperature and melt/rock composition is still limited (Kelsey *et al.*, 2008). If our capabilities to perform forward modelling of the main rock-forming minerals (garnet, micas, feldspar, $Al_2SiO_5...$) has been increasing since the evolution of more realistic solid solutions models (Holland & Powell, 1998), the forward modeling of dateable accessory minerals is hampered by the lack of thermodynamic data both for accessory minerals as well as for trace-element bearing end-members of major minerals (YAG garnet, for example; Pyle & Spear, 2000).

Vance *et al.* (2003) suggest potential solutions to solve the problematic link between petrological and chronometric data. The use of "the-inclusion-is-older-than-the-host" textural criteria still constitutes the favorite approach to resolve the relative chronology between major and accessory minerals. This approach, however, is hampered by a number of factors reviewed by Langone *et al.* (2011). Another method to infer a relative chronology between dateable and major minerals is the analysis of trace element distribution between major and accessory minerals. The Y partitioning between garnet

and monazite or the Heavy Rare Earth Elements (HREE) distribution between garnet and zircon/monazite are common tools to asses when these accessory phases grew with respect to *P*-*T* sensitive equilibria involving garnet. Another approach is linking an accessory phase to a specific reaction involving major rock-forming minerals, so a metamorphic mineral growth event can be dated. All approaches mentioned above require textural and/or chemical connection between geochronometers and geothermobaromers. Only few thermodynamic modelling involving phosphates have been performed (Janots *et al*, 2007; Spear & Pyle, 2010). However, the poor availability of activity models for accessory minerals doesn't always permit to use thermodynamic modelling to obtain the *P*-*T* conditions of the reactions between accessory minerals. An alternative way is to link the growth of accessory minerals to the crystallization of *P*-*T* sensitive major minerals through chemical an microstructural constraints (Gieré *et al*, 2011; see chapter 3).

1.4 CHAPTER OVERVIEW

The chapters are chronologically ordered according to the age of the investigated metamorphic stage.

Chapter 2 proposes an integration between classical microstructural analyses and thermodynamic modelling in the Cerreto micaschist. Themodynamic modelling comprises phase diagrams calculation and garnet isopleths thermobarometry. This approach aims to reconstruct P-T-D path as much complete as possible for the Variscan collision and exhumation. A comparison between the obtained P-T path and the thermobaric trajectories proposed for other Variscan basement of the circum-Mediterranean area is presented.

Chapter 3 focuses on the post-Variscan evolution in the Pontremoli micaschist recovered from the cuttings of the Pontremoli 1 commercial well (Agip; Anelli *et al*, 1994). Coronitic microstructures around monazite grains in this rock consist of apatite surrounded by allanite and an epidote rim. The chemistry and microstructure of the 294 \pm 8 Ma old Th-rich monazite grains suggest that the monazite was inherited and underwent partial dissolution at peak metamorphic conditions. Element partitioning between garnet and accessory minerals and a detailed microstructural study on garnet inclusion mineralogy allowed us to link the crystallization of REE-minerals to the garnet growth. An anticlockwise *P*-*T* path was reconstructed taking into account the effect of garnet fractionation on bulk-rock composition. The obtained *P*-*T* path

postdates 296 ± 6 Ma and results different from the available metamorphic histories of the pre-Mesozoic basement of the Northern Apennines.

Chapter 4 reports a combined microstructural-geochronological study on the shearing fabric in the Cerreto mylonites. This is the first attempt of determining the age of the shearing stage after the Variscan peak in the in the Variscan basement of Northern Apennines. The issue of linking petrology and geochronology has been faced by dating a fabric-forming mineral, which is potassic white mica, through *in situ* and step-heating laserprobe Ar-Ar method. The study provides a comparison of the age distribution in mylonitc foliation and in asymmetric foliation boudins and a possible interpretation of the mylonitic ages.

Chapter 5 concerns the Alpine peak metamorphism in the chloritoid-bearing hematiterich phyllites from the Pisani Mountains. This study investigates also the effects of different Fe₂O₃ bulk contents on the calculated phase equilibria of lowtemperature/intermediate-pressure metasedimentary rocks. Thermodynamic modelling within the MnO–Na₂O–K₂O–FeO–MgO–Al₂O₃–SiO₂–H₂O–TiO₂–O (MnNKFMASHTO) chemical system fails to reproduce the observed mineral compositions when the bulk Fe₂O₃ is determined through titration. The mismatch between observed and computed mineral compositions and assemblage is resolved by tuning the effective ferric iron content by $P-XFe_2O_3$ diagrams. The introduction of ferric iron affects the stability of the main rock-forming silicates that often yield important thermobaric information. The use of $P-T-XFe_2O_3$ phase diagrams provided chemical and thermobaric constrains about the Alpine peak conditions.

Chapter 6 presents the general conclusions. The main findings and the open questions are reported.

1.5 PUBLISHED PARTS

This thesis comprises three papers prepared for the submission to international journals. A fourth paper (chapter 4) has been already published and it has been reported unchanged into the thesis. Only the lay-out and the figure positions has been rearranged. The pubblication is:

Lo Pò D. & Braga R. (2014). Influence of ferric iron on phase equilibria in greenschist-facies assemblages: the hematite-rich metasedimentary rocks from the Monti Pisani (Northern Apennines). *Journal of Metamorphic Geology*, 32, 371-387.

Participation at international and national conferences was accompanied by oral or poster presentations of some results of my PhD research. The corresponding abstract are:

Lo Pò D., Braga R., Theye T., Massonne H.-J., Montanini A., and Molli G. (2014). Linking the formation of coronitic microstructures around monazite to the growth of garnet in the Pontremoli well metapelites (Northern Apennines, Italy). *Rend. Online Soc. Geol. It.*, Suppl. n. 1 Vol. 31., p. 474. [Oral presentation at the SGI-SIMP Conference 2014]

Lo Pò D. & Braga R., (2013). Quantifying Effective Ferric Iron Content in Hematite-Rich Metapelites Through Phase Equilibria Modelling. *Mineralogical Magazine*, 77(5) 1640. [Oral presentation at the Goldschimdt 2013]

Lo Pò D., Braga R., Montanini A., and Molli G. (2012). Tectono-metamorphic evolution of medium-grade metapelites of the Northern Apennine Paleozoic Basement inferred from quantitative modeling of microstructures. *European Mineralogical Conference* Vol. 1, EMC2012-394-2. [Poster at the European Mineralogical Conference 2012]

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Chapter 2

INSIGHTS INTO THE VARISCAN OROGENY:



THE TECTONO-METAMORPHIC EVOLUTION

OF THE CERRETO MICASCHIST

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2.1 INTRODUCTION

Medium-grade mylonitic metapelite can preserve several textural domains related to different stages of the tectono-metamorphic evolution. Sometimes even mineral assemblages formed during prograde-metamorphism are not totally obliterated. A detailed microstructural analysis is the essential basis to link the mineral assemblage (effect of the metamorphism) with the fabric (effect of the deformation) and thus to reconstruct the tectono-metamorphic evolution (Passchier & Trouw, 2005). Then thermodynamic modelling allows to associate Pressure and Temperature constrains for each microstructure and metamorphic stage by comparing the observed mineral assemblages and compositions in distinct microstructural positions with the computed ones in P-T phase diagrams (Powell & Holland, 2008). Therefore thermodynamic modelling coupled with a detailed microstructural analysis is a powerful tool to reconstruct Pressure-Temperature-Deformation paths (Williams & Jercinovic, 2012).

A combined microstructural-petrological study on the micaschist from the Cerreto Pass provides new *P-T* constraints for the recognized metamorphic-deformational phases. The studied samples belong to the Northern Apennines Variscan basement, a remnant of the southern segment of the Variscan chain now reworked within the northern Apennines nappe stack. Classical geothermobarometry has already been applied on this crustal sector (Molli *et al.*, 2002). However, pseudosections are a better approach for quantifying the Pressure-Temperature variations during the metamorphic evolution since they report all the possible paragenesis along a path (Powell and Holland, 2008). Even if the Cerreto micaschist exhibit a strong mylonitic fabric, relics of the earlier stages were recognized and used to constrain the prekinematic history. In particular the pre-peak stage, never constrained before, points to a thickening process at middle crustal depths during the Variscan collision with the burial of this crustal sector up to 45 km depth. The peak stage, attained at 600°C, was followed by exhumation along

Figure of the previous page: Variscan garnet porphyroclast in a Cerreto micaschist. **Collaboration**: This study is in collaboration with Prof. H.-J. Massonne and Dr. T. Theye from the University of Stuttgart (guidance on microprobe analyses), Dr. A. Montanini and Dr. G. Molli respectively from the University of Parma and University of Pisa (providing the CER8 sample and contributing to the discussion of data).

ductile shear zones. The strong deformation in this stage is responsible for the well developed mylonitic fabric.

2.2 GEOLOGICAL FRAMEWORK

The Northern Apennines are an Alpine orogenetic belt generated after the collision between Europe (the Sardinia-Corsica block) and Africa (the Adria microplate) from Oligocene times (Vai & Martini, 2001).

The Northern Apennine nappe-pile edifice consists of oceanic and continental units (from top to bottom): (1) Liguride Units: ophiolitic rocks derived from a branch of the Western Tethys ocean; (2) Tuscan Unit: a continental Triassic-Tertiary sedimentary sequence unaffected by the Alpine metamorphism; (3) Tuscan Metamorphic Units: Variscan basement with its Triassic-Oligocene cover, belonging to the Adria-derived continental sequence and affected by the Alpine metamorphism.

The Variscan basement (Franceschelli *et al.*, 2004) is present in scattered outcrops in the Cerreto Pass (Fig. 2.1a), in the Tuscan Metamorphic Ridge in the subsurface of the Larderello and Pontremoli areas (Fig. 2.1b). The Cerreto metamorphic rocks represent

the northernmost of the occurrence Apennines Northern Variscan basement. They consist of mylonitic micaschist with minor lenses of mafic amphibolite. Some micaschist contain porphyroclasts garnet while others exhibit a strong mylonitic fabric, accompanied by retrogression of garnet to chlorite. The main foliation in amphibolite was dated at 312-328 Ma by Ar-Ar on hornblende



Figure 2.1 Regional geological framework of the study area (Molli et al., 2002): (a) location of the Cerreto outcrop; (b) distribution of the Paleozoic basement in the Northern Apennines with the sampling site.

separates (Molli *et al.*, 2002) and formed at 630°C and 8.5 kbar. This confirms that the Cerreto metamorphic rocks were involved in the collisional stage of the Southern Europe Variscides (Molli *et al.*, 2002). The mylonitic retrograde stage likely occurred in a post-collision setting at 530°C and 4-5 kbar (Molli *et al.*, 2002).

2.3 ANALYTICAL METHODS

Petrographic observation and microstructural analysis were performed with a polarising optical microscope and a scanning electron microscope Philips 515B at the Dipartimento di Scienze Biologiche, Geologiche ed Ambientali (Università di Bologna). Mineral chemistry was acquired using a CAMECA SX100 electron microprobe (EMP) equipped with five wavelength dispersive (WDS) spectrometers at the Institut für Mineralogie und Kristallchemie (Universität Stuttgart). Na, Mg, Al, Si, K, Ca, Ti, Mn, Fe, Ba were analysed in potassic white mica, chlorite, biotite, garnet, plagioclase at acceleration voltage 15 kV and primary electric current 10-15 nA. Counting times for each element element were 20 s at the peak and the background. Mg, Al, Si, Ca, Ti, Mn, Fe, Cr were analysed in garnet at acceleration voltage 15 kV and primary electric current 15 nA. Synthetic and natural minerals and pure oxides were used as standards. The PaP correction procedure was applied. X-ray maps for Fe, Mn, Mg, Ca in garnet were acquired with a counting time of 50 ms and a step width of 6 µm. Acceleration voltage and electric current are 15 kV and 70 nA, respectively. Fe, Mg, Ti, Na X-ray maps in potassic white mica were acquired using a counting time of 100 ms, a step width of 1 µm, an acceleration voltage of 15 kV and a current of 30 nA. Structural formulae for potassic white mica, chlorite, garnet, plagioclase, biotite were calculated with CalcMin program (Brandelik, 2009), while X-ray map images were elaborated with XMap software (Bernhardt et al., 1995).

The bulk-rock composition was determined through WD-XRF analysis on pressed powder pellets using a Philips PW1480 and an Axios-Panalytical spectrometers at the Department of Biological, Geological and Environmental Sciences, Bologna University. Loss on ignition (L.O.I.) was obtained with standard gravimetric procedures, after heating the powder at 950 °C for 12 h.

2.4 PETROGRAPHY

The Micaschist Complex from the Cerreto Pass consists of mylonitic micaschist with pre-kinematic microstructures (Fig 2.2a, b, c, d) and a shearing fabric (Fig. 2.2e, f, g, h). Ultramylonitic levels cut occasionally the mylonitic foliation.

Two samples were studied in detail: PC1 which is a garnet-free micaschist showing a well developed mylonitic foliation, and CER8 which is a garnet-bearing micaschist already described in Molli *et al.* (2002).

PC1 shows quartz porphyroclasts and mica-fishes wrapped by a fine-grained potassic white mica-chlorite mylonitic foliation. The quartz porphyroclasts host chlorite and potassic white mica inclusions.

Large polygonized quartz crystals surrounded by quartz subgrains with suturated grain boundaries form bands (Fig. 2g). The mylonitic foliation is locally crenulated (Fig. 2.2f) or associated with C'-shear bands. Asymmetric foliation boudins consisting of potassic white mica aggregates are also present. Garnet is totally pseudomorphosed by chlorite. Plagioclase grains are poorly present in the mylonitic foliation. Ilmenite, rutile, titanite, allanite, epidote, pyrite, tourmaline, apatite and zircon are the accessory minerals. Ilmenite is xenoblastic and forms also fish-shaped crystals and boudins (Fig. 2.2d). It contains inclusions of biotite. Titanite forms crystals parallel to the mylonitic foliation. An ilmenite crystal, with a rutile core and titanite rim is observed (Fig. 2.2h). Allanite cores are surrounded by epidote rims. In this section an epidote-albite vein is also present.

Figure 2.2 (**next page**) Petrographic features of the Cerreto micashist: (a) σ -type quartz mantled-porphyroclast (CER8; crossed polars photomicrograph); (b) biotite inclusions in a plagioclase porphyroclast (CER8; crossed polars photomicrograph); (c) fractured and partially retrogressed garnet porphyroclast (CER8; crossed polars photomicrograph); (d) mica-fish and ilmenite-fish (PC1; BSE image); (e) C'-type shear bands (CER8; photomicrograph with plane polarized light); (f) hinge of a fold in the mylonitic foliation (PC1; BSE image); (g) oblique foliation constitutes by aligned quartz subgrains with suturated grain boundaries (CER8; crossed polars photomicrograph); (h) rutile core in an ilmenite crystal surrounded by titanite rim (PC1; BSE image). Abbreviations of minerals: Pl = plagioclase, Bt = biotite, Ms = K white mica, Ilm = ilmenite, Rt = rutile, Ttn = titanite.



CER8 contains a fine-grained potassic white mica-biotite-chlorite mylonitic foliation (Fig. 2.2e) and pre-kinematic crystals, consisting of mica fishes and quartz, plagioclase and garnet porphyroclasts (Fig. 2.2a, c). Ilmenite, rutile, epidote, allanite, zircon, tourmaline, titanite, chalcopyrite, pyrite, microcline and apatite are accessory minerals. Biotite fishes host plagioclase and zircon inclusions (Fig. 2.2b). Plagioclase porphyroclasts contain inclusions of biotite, apatite and rutile. Garnet porphyroclasts are fractured and partly retrogressed to chlorite (Fig. 2.2c). Ilmenite, quartz and zircon are enclosed in garnet. Recrystallized quartz grains define an oblique foliation (Fig. 2.2g). Rutile cores with ilmenite rims and ilmenite boudins have been observed in both garnet rim and matrix. Epidote and allanite are present both in the mylonitic foliation and in ultramylonitic levels associated with fine-grained chlorite.

2.5 MINERAL AND ROCK COMPOSITION

PC1

Prekinematic potassic white mica has Mg-rich cores and Na-rich rims (Fig. 2.3a, b). The Mg-rich prekinematic potassic white mica has Si=3.17-3.30 apfu, Mg=0.14-0.22 apfu and Na=0.04-0.06 apfu. Na/(Na+K) is 0.04-0.06. The Na-rich prekinematic potassic white mica contains Si=3.03-3.13 apfu, Mg=0.05-0.08 apfu, Na=0.15.0.27 apfu and Na/(Na+K)=0.05-0.27. Potassic white mica lying along the mylonitic foliation has a more variable composition: Si ranges from 3.06 to 3.19 apfu, Mg is 0.06-0.23 apfu, Na is 0.02-0.10 apfu and Na/(Na+K) is 0.02-0.10 (Fig. 2.4a, b, c, d; Tab. 2.1). Chlorite has Si=5.15-5.93 apfu, Al^{IV} =2.07-2.85 apfu, Al^{VI} =2.21-3.34 apfu and Fe/(Fe+Mg)=0.50-0.60 (Tab. 2.1). Feldspar grains are albite (Ab₉₄₋₉₉An₀₋₅Or₀₋₄) and contain K- feldspar exsolutions (Ab₀₋₁An₂₋₅Or₉₅₋₉₇). The bulk-composition (Tab. 2.3) is similar to PAAS (Post-Archean Australian Shale; Taylor & McLennan, 1985) and UCC (Upper Continental Crust; McLennan, 2001), excepting for CaO, which is depleted in PC1. In the Fe₂O₃/K₂O *vs* SiO₂/Al₂O₃ diagram (Herron, 1988) the composition of PC1 plots in the wacke-shale boundary.

Table 2.1 Mineral compositions of potassic white mica, chlorite and feldspar in PC1. The formula of potassic white mica is calculated assuming 42- (Ca+ Ba) valencies and 4 H. Fe³⁺ is stoichiometric-estimated considering $Fe^{3+} = 42 - sum$ of the valencies (Si, Ti, Cr, Al, Mn, Mg, Fe)-Ba-Ca. The structural formula of chlorite is calculated on the basis of 28 oxygen and neglecting Na and Ca. Chlorite is assumed as completely idratated, with 16 H. The structural formula of plagioclase is based on 8 oxygen.

K-white mica									
	Prekine	matic Mg-r	ich core	Prekine	matic Na-r	ich rim	Ν	Iylonitic foliat	tion
Point analyses	Ms11_4	Ms11_5	Ms11_6	Ms08_1	Ms9_2	Ms9_3	Ms10_01	Ms10_02	Ms10_03
SiO ₂	50.13	49.69	50.62	46.40	47.05	46.38	46.39	45.55	45.10
TiO_2	0.30	0.33	0.38	0.14	0.46	0.40	0.37	0.36	0.31
Al_2O_3	31.45	31.33	30.95	34.78	35.45	36.15	33.85	34.50	34.12
FeO	1.64	1.43	1.66	2.15	1.09	0.98	1.87	1.33	1.69
Fe ₂ O ₃	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.
MnO	0.05	n.d.	n.d.	n.d.	n.d.	n.d.	0.04	n.d.	n.d.
MgO	2.04	1.94	2.30	1.10	0.87	0.72	1.19	0.78	0.95
CaO	0.04	0.03	0.01	n.d.	0.03	0.01	0.02	0.01	0.01
Na ₂ O	0.42	0.39	0.39	0.35	0.92	0.91	0.33	0.43	0.33
K ₂ O	10.56	10.76	10.46	10.94	10.56	10.35	10.65	10.93	10.60
BaO	0.19	0.20	0.15	0.35	0.24	0.27	0.34	0.25	0.32
Total	96.82	96.11	96.91	96.20	96.67	96.18	95.04	94.13	93.44
Si	3.274	3.274	3.295	3.085	3.094	3.060	3.112	3.088	3.079
Al ^{IV}	0.726	0.726	0.705	0.915	0.906	0.940	0.888	0.912	0.921
Ti	0.015	0.016	0.019	0.007	0.023	0.020	0.019	0.018	0.016
$\mathrm{Al}^{\mathrm{VI}}$	1.695	1.707	1.668	1.810	1.841	1.870	1.787	1.844	1.823
Fe ²⁺	0.089	0.079	0.090	0.120	0.060	0.054	0.105	0.075	0.096
Fe ³⁺	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.
Mn	0.003	n.c.	n.c.	n.c.	n.c.	n.c.	0.002	n.c.	n.c.
Mg	0.199	0.191	0.223	0.109	0.085	0.071	0.119	0.079	0.097
Ca	0.003	0.002	0.001	n.c.	0.002	0.001	0.001	0.001	0.001
Ва	0.005	0.005	0.004	0.009	0.006	0.007	0.009	0.007	0.009
Na	0.054	0.050	0.049	0.045	0.117	0.117	0.043	0.056	0.044
Κ	0.880	0.905	0.869	0.927	0.886	0.871	0.912	0.945	0.923

Table 2.1 continued

		Chl	orite		Feldspar					
		Mylonitic	foliation		My	lonitic folia	tion	Within pla	agioclase	
Point analyses	Ch106_02	Ch106_03	Ch106_04	Chl06_05	P101	Pl06_10	P103	Kfs03	Kfs04	
SiO ₂	24.20	24.41	23.93	25.67	68.03	68.27	67.94	64.39	65.00	
TiO ₂	0.06	0.06	0.07	0.07	0.07	0.00	0.10	0.00	0.00	
Al_2O_3	22.05	22.06	22.07	21.85	20.71	20.05	20.97	18.52	18.30	
FeO ^{tot}	28.85	28.88	28.87	28.49						
Fe ₂ O ₃					0.45	0.09	0.31	0.27	0.28	
MnO	0.21	0.23	0.23	0.26						
MgO	12.55	12.89	12.52	12.20	n.d.	n.d	0.03	n.d	0.01	
CaO					0.84	0.67	1.11	0.02	n.d	
Na ₂ O					11.67	11.59	11.55	0.54	0.31	
K ₂ O					0.10	0.07	0.11	16.54	16.39	
BaO					0.00	0.00	0.00	0.33	0.17	
Total	87.91	88.53	87.69	88.54	101.88	100.73	102.12	100.61	100.45	
Si	5.193	5.199	5.154	5.432	2.933	2.967	2.923	2.977	2.997	
Ti	0.010	0.010	0.011	0.011	0.002	0.000	0.003	0.000	0.000	
Al	5.576	5.536	5.603	5.449	1.052	1.027	1.063	1.009	0.994	
Fe ³⁺					0.014	0.003	0.010	0.010	0.010	
Fe ^{tot}	5.178	5.143	5.201	5.042						
Mn	0.037	0.041	0.042	0.047						
Mg	4.015	4.093	4.022	3.850						
Ca					0.039	0.031	0.051	0.001	n.c.	
Na					0.976	0.977	0.963	0.048	0.027	
K					0.006	0.004	0.006	0.975	0.964	
Ba					n.c.	n.c.	n.c.	n.c.	n.c.	
An					0.038	0.031	0.050	0.007	0.003	
Ab					0.956	0.966	0.944	0.047	0.027	
Or					0.006	0.004	0.006	0.947	0.969	

CER8

Pre-kinematic potassic white mica shows compositional variations: the core is characterised by higher contents of Ti, Fe and Mg, while the rim is richer in Na (Tab. 2.2). Specifically, the core contains Si=3.12-3.22 apfu, Fe=0.11-0.14 apfu, Mg=0.18-0.24 apfu, Ti=0.04-0.06 apfu, Na=0.09-0.12 apfu, while the rims have Si=3.05-3.12 apfu, Fe=0.06-0.10 apfu, Mg=0.07-0.13 apfu, Ti=0.03-0.04 apfu, Na=0.03-0.13 apfu. Potassic white mica along the mylonitic foliation is characterized by scattered compositions, with Si=3.04-3.13 apfu, Fe=0.03-0.14 apfu, Mg=0.11-0.25 apfu, Ti=0.02-0.06 apfu, Na=0.03-0.06 apfu. Pre-kinematic biotite has XFe=0.48-0.52. XFe in mylonitic biotite is 0.52-0.57. Biotite deriving from garnet retrocession has Si=2.79-2.92 apfu and XFe=0.51-0.55. Mylonitic chlorite contains Si=5.30-5.35 apfu, Al^{IV} =2.65-2.70 apfu, Al^{VI} =2.61-2.75 apfu and XFe=0.47-0.48. Plagioclase included into mica-fishes (Ab₇₇₋₈₃An₁₆₋₂₂Or₁) and porphyroclasts (Ab₇₂₋₈₆An₁₄₋₂₇Or₀₋₃) are oligoclase. Microcline porphyroclasts in the matrix (Ab₁₋₉An₀₋₂Or₉₀₋₉₈) are also observed. Garnet shows a prograde growth zoning (Fig. 2.3c, d), ranging from Alm₆₂Grs₂₀Prp₉Sps₈ in the core to Alm₆₈Grs₁₈Prp₁₂Sps₂ in the rim (Fig. 2.4d; Tab. 2.2). The bulk-composition of

CER8 (Tab. 2.3) normalized to PAAS and UCC shows a depletion in MnO and CaO and an enrichment in MgO. The CER8 micaschist derives from a wacke-shale protolith according to the Fe_2O_3/K_2O vs SiO_2/Al_2O_3 diagram (Herron, 1988).

Table 2.2 Structural formulae of potassic white mica and garnet in CER8. The structural formula of potassic white mica is calculated assuming 42- (Ca+ Ba) valencies and 4 H. Fe³⁺ is stoichiometric-estimated considering $Fe^{3+} = 42 - sum$ of the valencies (Si, Ti, Cr, Al, Mn, Mg, Fe)-Ba-Ca. The garnet formula is normalized on 8 cations. Fe³⁺ is equal to 4 – Al.

K-white mica									
	Prekin	ematic co	ore		Prekinematic	rim	Mylonitic foliation		
Point analyses	Ms13_08	Ms16	Ms17	Ms04	Ms01_09	Ms01_10	Ms20	Ms90fm	Ms97fm
SiO ₂	47.82	47.83	48.19	46.58	46.43	46.71	46.04	45.97	46.08
TiO ₂	0.92	1.04	0.98	0.56	0.53	0.67	0.56	0.49	0.58
Al_2O_3	32.43	33.00	33.07	35.08	35.44	34.94	34.34	33.45	32.38
FeO	1.90	1.71	1.58	1.39	1.38	1.43	2.02	2.38	2.19
Fe ₂ O ₃	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.
MnO	0.01	n.d.	n.d.	0.01	n.d.	0.02	0.01	0.05	0.02
MgO	1.75	1.63	1.48	1.08	0.94	1.02	1.23	1.55	1.62
CaO	0.03	0.03	0.00	n.d.	n.d.	n.d.	n.d.	0.05	0.13
Na ₂ O	0.72	0.88	0.71	0.98	0.98	0.98	0.36	0.27	0.33
K ₂ O	9.94	9.84	10.19	9.83	10.05	10.08	10.62	10.35	10.49
BaO	0.86	0.83	0.83	1.00	0.97	0.93	1.20	1.26	1.35
Total	96.37	96.79	97.03	96.50	96.72	96.75	96.38	95.81	95.18
Si	3.167	3.150	3.166	3.080	3.069	3.086	3.076	3.091	3.131
Al ^{IV}	0.833	0.850	0.834	0.920	0.931	0.914	0.924	0.909	0.869
Ti	0.046	0.052	0.048	0.028	0.027	0.033	0.028	0.025	0.030
Al ^{VI}	1.698	1.712	1.726	1.814	1.829	1.807	1.780	1.742	1.724
Fe ²⁺	0.105	0.094	0.087	0.077	0.076	0.079	0.113	0.134	0.124
Fe ³⁺	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.
Mn	0.001	n.c.	n.c.	0.001	n.c.	0.001	0.001	0.003	0.001
Mg	0.173	0.160	0.145	0.106	0.092	0.100	0.123	0.156	0.164
Ca	0.002	0.002	0.000	0.000	n.c.	n.c.	n.c.	0.004	0.009
Ba	0.022	0.021	0.021	0.026	0.025	0.024	0.031	0.033	0.036
Na	0.093	0.113	0.090	0.125	0.126	0.125	0.047	0.035	0.044
Κ	0.839	0.827	0.854	0.829	0.847	0.849	0.905	0.888	0.910

Table 2.2 continued

Garnet										
		inner core				outer core	e		rim	
Point analyses	Grt10	Grt11	Grt16		Grt21	Grt24	Grt25	Grt27	Grt28	Grt29
SiO ₂	36.92	37.17	36.42	-	36.17	35.53	36.18	36.40	36.83	36.85
TiO ₂	0.27	0.27	0.28		0.11	0.09	0.10	0.13	0.13	0.12
Al_2O_3	21.40	21.27	20.99		21.21	21.06	21.19	21.29	21.42	21.32
Cr_2O_3	0.02	0.03	0.02		0.02	0.02	n.d.	0.02	0.01	0.02
Fe ₂ O ₃	0.23	0.68	0.92		0.55	0.71	0.65	0.28	0.49	0.35
FeO	28.43	28.72	28.60		30.24	30.55	30.56	31.35	31.28	31.12
MnO	3.77	3.83	3.28		2.24	1.48	1.21	0.87	0.88	0.86
MgO	2.39	2.45	2.63		2.96	2.92	2.99	2.74	2.99	2.99
CaO	7.00	6.89	6.97		6.08	6.39	6.63	6.41	6.52	6.38
Na ₂ O	0.04	0.05	0.04		0.02	0.02	0.03	0.04	0.04	0.04
Total	100.46	101.35	100.15		99.59	98.79	99.54	99.52	100.58	100.05
Si	5.812	5.809	5.725		5.691	5.603	5.682	5.750	5.752	5.800
Ti	0.032	0.031	0.034		0.013	0.011	0.012	0.015	0.015	0.014
Al	3.971	3.917	3.889		3.932	3.914	3.923	3.965	3.942	3.956
Cr	0.002	0.003	0.003		0.003	0.002	n.c.	0.002	0.001	0.003
Fe ³⁺	0.027	0.080	0.109		0.065	0.084	0.077	0.033	0.058	0.042
Fe ²⁺	3.743	3.754	3.760		3.979	4.028	4.014	4.141	4.084	4.097
Mn	0.503	0.507	0.437		0.298	0.198	0.161	0.116	0.116	0.115
Mg	0.560	0.570	0.616		0.693	0.687	0.700	0.645	0.697	0.701
Ca	1.180	1.154	1.175		1.024	1.080	1.115	1.085	1.091	1.076
Na	0.013	0.015	0.012		0.005	0.007	0.010	0.013	0.012	0.011
Grs	0.20	0.19	0.20		0.17	0.18	0.19	0.18	0.18	0.18
Prp	0.09	0.10	0.10		0.12	0.11	0.12	0.11	0.12	0.12
Alm	0.62	0.63	0.63		0.66	0.67	0.67	0.69	0.68	0.68
Sps	0.08	0.08	0.07		0.05	0.03	0.03	0.02	0.02	0.02

Table 2.3 Bulk-rock composition of PC1 and CER8 determined through WD-XRF analysis and modified for thermodynamic modelling.

	X	RF	Thermodyna	mic modelling
	PC1	CER8	PC1	CER8
	wt. %	wt. %	wt. %	wt. %
SiO ₂	67.74	63.29	70.53	65.48
TiO_2	0.85	0.90	0.89	0.93
Al_2O_3	16.90	14.47	17.60	14.97
Fe_2O_3	4.61	7.44	-	-
FeO	-	-	4.32	6.92
MnO	0.05	0.10	0.06	0.10
MgO	1.53	3.97	1.59	4.11
CaO	0.75	1.61	0.53	1.39
Na ₂ O	1.59	3.12	1.66	3.23
K ₂ O	2.71	2.77	2.83	2.87
P_2O_5	0.18	0.20	-	-
L.O.I.	3.08	2.13	in excess	in excess
Total	100.00	100.00	100.00	100.00



Figure 2.3 X-ray maps of potassic white mica in PC1 and garnet in CER8: (a) Mg distribution in potassic white mica; (b) Na distribution in potassic white mica; (c) Fe distribution in garnet; (d) Mn distribution in garnet. The colour code scales is related to the counts X-ray radiation per time unit.



Figure 2.4 Compositional diagrams for potassic white mica in PC1 and garnet in CER8: (a) $Fe^{2+}+Mg \ vs.$ Si in potassic white mica; (b) Mg vs. Al in potassic white mica; (c) Mg vs. Si in potassic white mica; (d) Na vs. Si in potassic white mica; (e) compositional profile along garnet.

2.6 GEOTHERMOBAROMETRY

2.6.1 Phase diagram calculation

Thermodynamic modelling was performed in the MnCKNFMASHT system with Perple_X package (Connolly, 1990; Connolly & Petrini, 2002). Mn was considered in the system because it has the effect of shifting garnet-in curve to lower pressures and temperatures (Thompson, 1957; Albee, 1965; Spear & Cheney, 1989; Mahar et al., 1997; Tinkham et al., 2001). Ti was also considered to include the Ti-bearing minerals in the parageneses. Iron was assumed as totally ferrous because of the lack of ferric-iron minerals and of the negligible stochiometric amount in major minerals. The

thermodynamic dataset of Holland & Powell (1998, with the 2002 upgrade) and the compensated Redlich Kwong (CORK) fluid equation of state by Holland & Powell (1991, 1998) were used. SiO₂ and H₂O were considered in excess. We used the following solid solution models: Pheng(HP) for K-white mica (Holland & Powell, 1998; the maximum paragonite content has been set to 50 mol.%), Mica(M) for Na-Cawhite mica (Massonne, 2010; the maximum muscovite content has been set to 50 mol.%), Gt(HP) for garnet (Holland & Powell, 1998), TiBio(HP) for biotite (Powell & Holland, 1999), Chl(HP) for chlorite (Holland et al., 1998), feldspar for plagioclase (Fuhrman & Lindsley, 1988), Kf for microcline-albite solid solutions (Waldbaum & Thompson, 1968), St(HP) for staurolite (Holland & Powell, 1998), hCrd for cordierite and **IlGkPy** for ilmenite (ideal models; http://www.perplex.ethz.ch/PerpleX_solution_model_glossary.html). Pure phases are titanite, rutile, zoisite, lawsonite, sillimanite, kyanite and andalusite. We performed thermodynamic modelling for the garnet-free micaschist (PC1; P-T pseudosection) and the garnet-bearing micaschist (CER8; garnet isopleths thermobarometry).

2.6.2 P-T pseudosection

The *P*-*T* pseudosection was calculated using the XRF bulk-rock composition of PC1 (Fig. 2.5). CaO was calculated excluding the Ca contained in apatite (Tab. 2.3). The considered P-T range is 350-600°C and 2-15 kbar. K-white mica is stable in the entire P-T grid. Na-Ca-white mica out-curve is above 400°C-2 kbar and 600°C-5 kbar. Chlorite breaks down at temperatures exceeding 600°C. Biotite appear at temperatures higher than 500°C. Garnet stability is above 400°C at P>12 kbar and above 550°C at P>4 kbar. Staurolite is stable in a narrow PT range around 600°C and 6 kbar. Plagioclase-in curve is at temperatures higher than 400°C at 2 kbar and 650°C at 15 kbar. K-feldspar occurs below 9 kbar and 500°C. Lawsonite appears in the lowtemperature high-pressure corner. Zoisite is present at 500°C at 7-9 kbar and 450°C at14 kbar. Rutile is not stable at pressures below 9 kbar for temperatures higher than 450°C. Ilmenite coexists with rutile at temperature equal to 450°C for pressures below 9 kbar and at pressures equal to 9 kbar for temperatures higher than 450 °C. Ilmenite then replaces rutile at temperatures higher than 450°C for pressures below 9 kbar. Titaniteout curve is at 400°C-2 kbar and 500°C-12 kbar. The Si content in K-white mica increases with pressure reaching 3.39 apfu. XFe=Fe/(Fe+Mg) in chlorite decreases from 0.62 with increasing temperature (Fig. 2.5c). The maximum measured Si content (3.30 apfu) and the Na/(Na+K) ratio (0.04) of the prekinematic Mg-rich potassic white mica plots at 455°C and 12 kbar in the paragenetic field chlorite-K white mica-garnet-Na white mica-titanite-rutile-quartz (Fig. 2.6). Considering 2σ -errors in the range of 5% for the temperature estimate and 10% in the pressure value (Massonne, 2013), the previous isopleth intersection plot at about 430-480°C and 11.0-13.5 kbar. The peak stage cannot be constrained with PC1, because garnet is completely retrogressed to chlorite. The retrograde evolution was constrained in pseudosection1. The intersection among the Si content in mylonitic potassic white mica (3.06-3.19), the XFe of mylonitic chlorite (0.50-0.60) and the titanite-bearing assemblages plot at 450-500°C and 6.0-7.5 kbar in the paragenetic field chlorite + potassic white mica + sodic white mica + plagioclase + ilmenite + titanite ± rutile (Fig. 2.6), in agreement with the observed paragenesis. Garnet-out curve is crossed, in agreement with the petrographic observations.



Figure 2.5 *P*-*T* pseudosection for PC1 with potassic white mica and chlorite isopleths. Abbreviations: Ms = potassic white mica (muscovite/phengite), Pg = sodic white mica, Chl = chlorite, Kfs = alkali-feldspar, Pl = plagioclase, Grt = garnet, Bt = biotite, St = staurolite, And = andalusite, Crd = cordierite, Ilm = ilmenite, Rt = rutile, Ttn = titanite, Lws = lawsonite, Zo = zoisite.



Figure 2.6 Potassic white mica isopleths constraining the pre-peak stage (grey circle) and potassic white mica and chlorite isopleths constraining the mylonitic stage in PC1 (grey triangle). The ilmenite-in curve, garnet-in curve and titanite-in curve are also reported.

2.6.3 Garnet isopleths geothermobarometry

garnet isopleths In order constrain the peak conditions, we applied to geothermobarometry for the garnet-bearing micaschist (CER8). XRF-bulk composition, corrected excluding the Ca and P in apatite (Tab. 2.3), and the same parameters of the pseudosection1 were used. The considered P-T range is 350-600°C and 2-10 kbar (Fig. 2.7). Garnet-in curve is at around 8 kbar in the range 350-550°C, while garnet is stable at pressures superior than 3 kbar for temperature higher than 550°C. Almandine increases reaching 70 mol. % with increasing temperature. Grossular increases from 70 mol. % towards low-temperature and high-pressures. Spessartine has its peak (24 mol. %) at 500°C and 7.5 kbar, then it decreases either down-temperature and uptemperature. Pyrope variation is also mainly controlled by temperature reaching the maximum value of 15 mol. % at 600°C. Garnet mode increases up to 4 vol. % with growing temperatures and pressures. The observed garnet core composition (Alm₆₂Grs₂₀Prp₉Sps₈) plots in a narrow area at 550-575°C and 8-9 kbar. The almandine, grossular and pyrope isopleths ($Alm_{68}Grs_{18}Prp_{12}$) of the garnet rim constrain the peak metamorphism at 580-600°C and 9-10 kbar. Considering the error associated with the isopleths geothermobarometry (Massonne, 2013), garnet core and garnet rim *P-T* estimations overlap. No fractionation effect could be considered because the garnet mode in CER8 is low (< 1 %).





Figure 2.7 Calculated almandine, grossular, spessartine and pyrope isopleths and the mode variations in CER8. The garnet core isopleths intersection and the garnet rim isopleths intersection is also reported.

2.7 DISCUSSION

2.7.1 *P-T-D* path

Thermodynamic modelling combined with microstructural analysis allowed us to reconstruct a clockwise P-T path for the micaschist belonging to the Variscan basement in the Cerreto Pass (Fig. 2.8). High-Si Mg-rich potassic white mica relics in deformed potassic white mica grains constrains the pre-peak stage at 430-480°C and 11.0-13.5 kbar. The rim of garnet porphyroclasts crystallized during the metamorphism peak, consisting of 580-600°C and 9-10 kbar. The mylonitic stage is represented by potassic white mica and chlorite along the mylonitic foliation and by the titanite-ilmenite transition in PC1. This stage occurred at 450-500°C and 6.0-7.5 kbar. The shearing event is also responsible for the development of the well-developed mylonitic fabric (mylonitic foliation, shear bands, asymmetric foliations boudins, quartz oblique foliation, deformed prekinematic crystals). The evolution of the calculated mineral assemblages in PC1 along the *P-T-D* path is shown in Fig. 2.9. The diagram displays the variation of the cumulative mineral modes with temperature and pressure: from 430°C-13-5 kbar (pre-peak stage), passing through 600°C-10 kbar (peak stage), to 450°C-6 kbar (retrograde stage). The pre-peak stage generated a quartz-potassic white mica-sodic white mica-chlorite paragenesis. Garnet porphyroclast may have started to crystallize in this stage. Accessory minerals were rutile and titanite. Increasing temperature and decreasing the pressure towards the peak conditions, titanite breaks down and rutile became the only Ti-bearing phase. Garnet abundance grows and when biotite appears, chlorite and potassic white mica amount fall down. At peak conditions chlorite disappears and the paragenesis is quartz-sodic white mica-potassic white micabiotite-garnet-rutile. Then the retrograde stage starts: rutile is replaced by ilmenite and garnet begins to be replaced by chlorite and potassic white mica. At 520°C garnet breaks down and plagioclase crystallizes. The modal amount of chlorite and potassic white mica increases by 5 and 10 vol. %, respectively. On the contrary, paragonite looses 10 vol. %. Ilmenite is partly replaced by titanite, in agreement with the observed titanite rims around ilmenite crystals. At 475°C Ca-poor plagioclase disappears in favour of K-feldspar and titanite. The calculated retrograde assemblage potassic white mica-chlorite-quartz-titanite-K-feldspar assemblage at 500-450°C is concordant with the observed mylonitic paragenesis. The only apparent discrepancy between observed and calculated mineral assemblages is sodic white mica. Sodic white mica was never observed in the selected sample with optical and scanning electronic microscopes. However, high-magnification observations with a field emission scanning microscopy on another micaschist from the Cerreto Pass reveal the existence of fine-grained sodic white mica flakes (wide 10-50 μ m and thick few μ m) intergrown with potassic white mica (see Chapter 4). It is indeed reasonable that sodic white mica is present also in PC1.



Figure 2.8 *P*-*T*-*D* path for the Cerreto micaschist. The table contains a summary of the tectonometamorphic evolution, with the microstructures, the constraining phases and the corresponding P-T estimates.



Figure 2.9 Calculated evolution of the mineral modes along the *P*-*T* path.

2.7.2 Tectonic implications

The obtained *P-T-D* path suggests that the Cerreto micaschist experienced a polyphasic metamorphism during the Variscan orogeny. The initial high-pressure stage may testify the burial at middle-crustal depths during the collision-related thickening of the crust. The temperature peak followed the pressure peak, and this may suggest that the heating subsequent to the thermal equilibration occurred when the exhumation had already begun. The retrograde stage, accompanied by a strong mylonitic deformation, is probably related to an exhumation process along ductile shear zones.

Only two *P-T* paths are available for the Variscan basement of Northern Apennine (Bertini *et al.*, 1994). The Cerreto metamorphic rocks have been correlated with the Micaschist Complex of the Larderello subsurface (Pandeli *et al.*, 2005). The Variscan basement in the Larderello subsurface has been divided into three tectonic units (Batini *et al.*, 2003; Bertini *et al.*, 2006):

- Phyllitic-Quartzitic Complex: low-grade metasedimentse and metabasite levels, overprinted by the Alpine metamorphism and correlated with the

Variscan succession in the Apuane Alps, in the Monti Pisani and in Punta Bianca;

- Micaschist Complex: two-mica garnet-bearing albite micaschist with minor amphibolite deriving from OFB (Ocean Floor Basalts), affected by a polymetamorphism during the Variscan and Alpine orogenesis;

- Gneiss Complex: muscovite-biotite gneisses with orthogneiss, OFB amphibolite layers and calc-silicate rocks, affected only by the Variscan metamorphism.

Bertini *et al.* (1994) reconstructed the thermobaric history of the Gneiss Complex in the Larderello subsurface through classical geothermobarometry and the stability fields of staurolite, muscovite, andalusite and cordierite. The pre-Alpine history shows a peak metamorphism at 500-600°C and 7 kbar, followed by an isothermal decompression at 500-600°C and 2-3.5 kbar. The Authors interpreted the obtained *P*-*T* path as related to a fast tectonic exhumation after the Variscan collision. The retrograde stage is different from our reconstruction that proposes a gradient of 40°C kbar⁻¹. The Larderello Gneiss Complex and the Cerreto micaschist metamorphic evolutions result remarkably different. This difference can be explained by considering the Cerreto micaschist and the Larderello Gneiss Complex as basement slices belonging to different paleogeographic domains.

Molli *et al.* (2002) investigated the metamorphic evolution of the Micaschist Complex in the Cerreto Pass. They didn't constrain the pre-peak stage with geothermobarometry, however they envisaged an early metamorphic event in garnet-amphibolite or eclogitic facies Using classical thermobarometry the peak conditions were constrained at 650° C and 8 kbar for the amphibolite and 630° C and 8.5 kbar for the micaschist. Ar-Ar dating on amphibole separates from undeformed and unretrogressed amphibolite gives Variscan ages (328-312 Ma) for the pre-mylonitic stage(s). The micaschist are associated with amphibolite and they show the same microstructural features (e.g. prekinematic relics and mylonitic fabric). Since they share a common metamorphic evolution from at least the peak conditions, the age determination of the amphibolite regards also the garnet-forming stage in the micaschist. According to Molli *et al.* (2002), the retrograde stage occurred at 530°C and 4-5 kbar for the amphibolite. The trajectory by Molli *et al.* (2002) is shifted by 50°C up in the temperature for the peak conditions and by 1 kbar down in the pressure for the retrograde stage with respect to the new *P-T* path. Nevertheless the overall shape of the two *P-T* trajectories is similar and point to a common peak and exhumation history for the amphibolite and metapelites. The differences in the *P*-*T* estimates may be due to the use of different geothermobarometric methods (classical geothermobarometry *vs.* pseudosections), the improvements of the thermodynamic data and solutions models for minerals during the last decade (Holland & Powell, 1998), and a lack of uncertainties estimations in the *P*-*T* calculations of Molli *et al.* (2002). Besides, the peak conditions in the micaschist have been estimated using the garnet rim composition with the higher almandine value. However, since garnet breakdown is intensive, the highest values of almandine could have been lost. As a consequence, higher temperatures in peak conditions cannot be ruled out. The application of thermodynamic modelling on amphibolite can resolve the question.

After the collision stage, the Variscan chain was dismembered in several segments in the interval 330-300 Ma by strike-slip shear zones (e.g. East Variscan Shear Zone; Padovano et al., 2012). The pre-Carboniferous rocks of Northern Apennines have been ascribed to the Southern part of the Variscan belt, as like the Austroalpine basement of the Eastern Alps, Corsica-Sardinia, the Calabrian-Peloritan orogen and the Kabylies (Bagnoli et al., 1979; Vai & Cocozza, 1986, Vai & Martini, 2001; Elter et al., 2004; Elter & Pandeli, 2005). The correlation among these basements relies on stratigraphic and structural similarities. In particular the greenschist-facies metamorphic rocks outcropping in the Apuane Alps show stratigraphic and metamorphic similarities with the External Nappe units of Sardinia (Conti et al., 1991; Pandeli et al., 1994). A correlation based on the metamorphic evolution has been discussed in Elter & Pandeli (2005) and Padovano et al. (2012). We selected the P-T paths reported by those Authors with the following criteria: (i) metapelitic rocks; (ii) medium-grade peak conditions; (iii) peak conditions occurring at about 330 Ma. We could compared only our retrograde P-T path to those trajectories (Fig. 2.10), because only our reconstruction and that of Angì et al. (2010) are the only ones providing information about pre-peak conditions (pressures higher than 10 kbar). For pressures below 10 kbar, analogies arise for the peak conditions and exhumation histories among the Cerreto micaschist, the Giuncana area (northwest sector of the Sardinia Massif; Ricci, 1992; Franceschelli et al., 2005; Cruciani et al., 2008), and the Posada Valley Area mylonites (Sardinia, Elter et al., 1999; Elter & Pandeli, 2005). On the contrary, the sillimanite-bearing gneisses of the Pohorje Massif (Austroalpine basement of the Eastern Alps; Thöni, 2002), the Mammola Paragneiss Complex (Serre Massif; Angì et al., 2010), and the ilmenitegarnet-bearing metapelite of the Mandanici Unit (Peloritani Mountains; Fiannacca *et al.,* 2012) show quite different peak temperature estimates. The peak conditions occurred for all the considered *P*-*T* paths at 9-10 kbar.

Thus, our *P-T-D* path show similarities with the metamorphic evolution of other segments of the Southern Europe Variscides, especially those outcropping in Sardinia. Some microstructures in the Cerreto Pass micaschist (e.g. book shelf sliding plagioclase and boudins) and the regional framework suggest that the mylonitic event could develop in a strike-slip tectonics with extensional kynematics (Elter & Pandeli, 2005).



Figure 2.10 Comparison of the obtained P-T path (yellow to blue arrow) with the thermobaric trajectories proposed for some Variscan segments in the circum-Mediterranean area: violet arrows and boxes for the Larderello Gneiss Complex (Northern Apennines; Bertini et al., 1994); green arrows and boxes for the Cerreto amphibolite (Northern Apennines; Molli et al., 2002); 1. Pohorje Massif (Eastern Alps); 2. Giuncana area (Sardinia); 3. Internal Nappe Zone (Sardinia); 4. Posada Valley Area (Sardinia); 5. Mammola Paragneiss Complex (Serre Massif); 6. Mandanici Unit (Fiannacca et al., 2012). See the text for references.

2.8 CONCLUSIONS

This contribution proposes a new reconstruction of the metamorphic evolution for the micaschist outcropping in the Cerreto Pass. These rocks represent a remnant of the Variscan basement, reworked into the Northern Apennine belt by the Alpine tectonics. For the first time thermodynamic modelling (P-T pseudosections and garnet isopleths geothermobarometry) was performed on the Cerreto metamorphic rocks in order to

constrain the pressure-temperature evolution. Although a strong retrogression affected the micaschist, prekinematic relics were preserved, thus even the prograde path was defined.

The Cerreto micaschist record a clockwise Pressure-Temperature path consisting of

- a prograde stage, occurring at high-pressure conditions (T=430-480°C; P=11-13.5 kbar) during the Variscan thickening;
- > a temperature peak (600 °C at 9-10 kbar) that postdates the pressure peak;
- > a retrograde stage at T = 450-500 °C and P = 6-7.5 kbar, likely related with the exhumation of Variscan crust along ductile shear zones in a strike-slip tectonic framework.

Comparison with the P-T paths of the Variscan basement in Easter Alps, Sardinia and the Calabrian-Peloritan orogen reveals similarities in the metamorphic evolution. On the contrary, the Gneiss Complex in Larderello shows a different P-T path.

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Chapter 3

A NEW P-T-t PATH FOR THE POST-VARISCAN HISTORY: LINKING MONAZITE AND GARNET IN THE PONTREMOLI METAPELITES
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A NEW *P-T-t* PATH FOR THE POST-VARISCAN HISTORY: LINKING MONAZITE AND GARNET IN THE PONTREMOLI METAPELITES

3.1 INTRODUCTION

The linkage between accessory minerals rich in rare-earth elements (REE) and major rock-forming phases plays a crucial role in deciphering metamorphic and tectonic processes (Vance et al., 2003). Monazite is one of the main REE-accessory minerals useful for obtaining geochronological information on a wide range of Ca-poor crustal rocks. Therefore much effort is dedicated to connect monazite U-Th-Pb chemical or isotopic ages with pressure (P)-temperature (T) constraints obtained from mineral assemblages, which can be modelled by thermodynamically-based phase diagrams. In low-medium grade pelitic and felsic compositions, monazite usually occurs as inherited grains and/or authigenic metamorphic crystals. Along a prograde metamorphic path monazite can be progressively consumed by the formation of other REE-accessory minerals like allanite (Wing et al., 2003; Janots et al., 2008). However, Williams et al. (2007 and references therein) reported the resumption of monazite growth at the staurolite-in/garnet-out isograd (~550°C). The phase relations described above are complicated by the textural and compositional evidence. During low-grade regional metamorphism (< 400°C) metamorphic monazite may form after the dissolution of precursor (inherited) monazite (Janots et al., 2006; Rasmussen & Muhling, 2007). Inherited and (new) metamorphic monazite may therefore potentially coexist under greenschist-lower amphibolite facies conditions. In these cases, efforts are required to discriminate between inherited monazite, unrelated to the metamorphic assemblages, and metamorphic monazite that can actually provide meaningful age constraints. Even when low-grade monazite and inherited monazite can be distinguished, it is not simple to relate a monazite age to P-T conditions obtained from rock-forming minerals. A common approach relies on the Y partitioning between garnet and monazite (Vance et al., 2003; Williams et al., 2007), but it requires that monazite grew together with garnet during metamorphic evolution. However, clear breakdown microstructures of monazite

Figure of the previous page: Variscan garnet porphyroclast in a Cerreto micaschist. **Collaboration**: This study is in collaboration with Prof. H.-J. Massonne and Dr. T. Theye from the University of Stuttgart (guidance on microprobe analyses), Dr. A. Montanini and Dr. G. Molli respectively from the University of Parma and University of Pisa (providing the CER8 sample and contributing to the discussion of data).

testify that monazite can be consumed during prograde metamorphic evolution (Janots *et al.*, 2006; Rasmussen & Muhling, 2007) and the linkage between monazite and *P-T* sensitive minerals becomes more complicated. In such a case a possible approach is the connection of the products of the monazite breakdown to the crystallization of *P-T* sensitive major minerals (e.g. garnet) through chemical and microstructural constraints (Gieré *et al.*, 2011).

Here we present a study where inherited, and partially decomposed, monazite has been recognized in garnet-micaschist from the basement beneath Pontremoli, Northern Apennines (Fig. 3.1). The early Permian monazite age presented here provides a lower age limit for the metamorphic history of the Northern Apennine basement. Careful considerations of garnet chemical zoning (major and selected trace elements), garnet mineral inclusions (including xenotime), monazite breakdown products and phase diagram modelling permit to link the reaction history among accessory minerals with the P-T evolution of garnet. We propose a previously unrecognized P-T path for the post-Variscan history of pre-Mesozoic basement reworked within the Apennines orogenic structures.

3.2 GEOLOGICAL FRAMEWORK AND SAMPLE PROVENANCE

The Northern Apennines are an orogenic belt formed during the collision between the Corsica-Sardinia block and the Adria microplate in Oligocene-Miocene times. In the Northern Apennines, slices of Variscan basement belonging to the distal Adria (i.e. palaeo-Africa) margin can be observed in tectonic windows below the Mesozoic-Tertiary sedimentary sequences (Tuscan Units) that are tectonically overlain by the ophiolite-bearing Ligurian units (Vai & Martini, 2001 and reference therein).

The Variscan basement of the Northern Apennines occurs in few outcrops (Cerreto Pass, Punta Bianca and in the so-called Tuscan Metamorphic Ridge) and in the Larderello and Pontremoli subsurface (Figs. 3.1a, b). The upper 1.5 km of the Variscan basement underlying the sediments of the Larderello geothermal system (Pandeli *et al.*, 2005) istraditionally subdivided into three tectonic units (from top to bottom): (i) Phyllitic-Quartzitic Complex; (ii) Micaschist Complex; (iii) Gneiss Complex (Elter & Pandeli, 1990; Bertini *et al.*, 1994). The Gneiss Complex and the Micaschist Complex were also affected by a thermometamorphism related to the emplacement of Pliocene-Quaternary granites.

The Variscan basement in the Pontremoli subsurface has been correlated to the Micaschist Complex of the Larderello geothermal field (Pandeli *et al.*, 2005). This Complex underwent a low-medium grade regional Variscan metamorphism, followed by a thermo-metamorphic stage at 285 Ma. An Alpine greenschist-facies polyphase metamorphism overprinted the earlier stages (Franceschini, 1998; Pandeli *et al.*, 2005; Musumeci *et al.*, 2011). Differently from the Micaschist Complex of Larderello, the Variscan basement under Pontremoli was not involved in a contact metamorphism induced by the Pliocene-Quaternary igneous activity (Pandeli *et al.*, 2005).

The Pontremoli 1 well intercepted the Variscan basement within the 3059-3520 m depth range (Anelli *et al.*, 1994). SamplesAG6, AG6a, and 3118 come from 3117-3118 m depth (core 6) and are fine-grained rocks with a single planar schistosity consisting of alternating white mica-chlorite quartz-albite-calcite layers. Samples AG7 and Pontremoli 7 were taken from 3422 m depth (core 7) whereas the samples AG8 and Pontremoli8 belong to the stratigraphically deepest core 8 and are representative of the basement at 3519-3520 m depth. Samples AG7, Pontremoli7, AG8 and Pontremoli 8 were studied in detail because of the presence of garnet.



Figure 3.1 Geological framework.(a) location of the Pontremoli well;(b) distribution of the Variscan basement in Northern Apennines.

3.3 PETROGRAPHY

The investigated samples (AG7, Pontremoli7, AG8 and Pontremoli 8) show a medium grain size and a schistose fabric with white mica and chlorite wrapping around garnet porphyroblasts (Fig. 3.2a), which contain inclusions of a variety of minerals such as ilmenite, xenotime, allanite, epidote and pyrite. Chlorite and white mica aggregates with a decussate fabric were also observed in late microfolds (Fig. 3.2b). Quartz and poikilitic plagioclase are the other main minerals. Plagioclase crystals host K-feldspar and white mica inclusions. Chlorite, white mica, plagioclase and quartz occur in pressure shadows around the garnet porphyroblasts. Fractures in garnet are filled with chlorite and white mica. The accessory minerals are ilmenite, rutile, zircon, pyrite, tourmaline, monazite, apatite, thorite, allanite, epidote, xenotime, and zircon.

Ilmenite inclusions within garnet define an internal schistosity, which is discontinuous to the external one. Euhedral ilmenite crystals in the matrix are oriented parallel to the schistosity. Rutile cores in matrix ilmenite and K-feldspar inclusions in plagioclase occur. Allanite is also present in the matrix and forms aggregates of anhedral grains. Irregular xenotime grains with corroded rims occur only in the garnet inner core.

The microstructural analysis permits to reconstruct the following metamorphic evolution. Rutile and K-feldspar may represent the remnants of an earlier assemblage. The transition from rutile to ilmenite predates the garnet forming stage, because no rutile but ilmenite has been observed in the garnet inner core. The internal schistosity of ilmenite in garnet represents the relics of a former schistosity (Fig. 3.2c). The crystallisation of garnet porphyroblasts overprints and entraps this schistosity. During garnet growth, white mica-chlorite-ilmenite-external schistosity and quartz and plagioclase re-crystallized. The growth of chlorite and white mica with decussate structure postdates the external schistosity formation. Potassic white mica and chlorite crystallization as replacement products of garnet rims and plagioclase growth in the matrix may be related to a late metamorphic stage.

Pontremoli 7 was studied in more detail because it contains reaction microstructures due to the breakdown of monazite grains (Fig. 3.2d; see "Coronitic microstructure around monazite"). This sample was thus considered suitable to infer the reaction history among REE-minerals.



Figure 3.2 Main petrographic features of the Pontremoli metamorphic rocks. (a) potassic white mica and chlorite foliation surrounding a garnet porphyroblast (sample AG7, BSE image); (b) potassic white mica with decussate structure (sample Pontremoli 7, crossed polars photomicrograph); (c) relationships between S_1 and S_2 (sample Pontremoli 8, BSE image); (d) microstructure representing the replacement of monazite in the matrix (sample Pontremoli 7, BSE image). Mineral abbreviations: Grt = garnet, Qtz = quartz, Chl = chlorite, Ms = K white mica, Ilm = ilmenite.

3.3.1 Coronitic microstructure around monazite

A detailed microstructural study on REE-accessory minerals was carried on with a scanning electron microscope Philips 515B with an EDAX energy-dispersive system (EDS) at the Dipartimento di Scienze Biologiche, Geologiche ed Ambientali (Università di Bologna). Analytical conditions were 15 kV and 30 nA for the qualitative analyses of minerals.

Monazite in Pontremoli 7 is surrounded by a complex, polymineralic corona microstructure consisting of apatite + Th-silicate, allanite and epidote (Fig. 3.3). This microstructure, which was previously described from amphibolite-facies orthogneisses by Finger *et al.* (1998) and interpreted as the result of monazite-consuming reactions, is associated with every monazite grain observed within the Pontremoli 7 micaschist. The microstructure has a maximum width of 200 μ m and may occur at two microstructural

sites: (1) in the rock matrix surrounded by potassic white mica lamellae (Figs. 3.3a, c), or (2) included in the white mica + chlorite aggregate partially replacing garnet (Fig. 3.3 d, e). When the microstructure is elongated, the elongation is parallel to the schistosity (Fig. 3.3c).

Monazite crystals show lobate rims with a maximum size of 60 μ m. The surrounding **apatite** is 10 μ m thick. It contains tiny inclusions of Th-silicate, characterised by its brightness in back-scattered electron (BSE) images and confirmed by qualitative EDS analyses. This corona has sharp protrusions into the allanite corona. In places apatite grains grade into the allanite corona and apatite and Th-silicate can be found also in the allanite coronas. The **allanite** corona has a width of c. 10 μ m. It shows an irregular internal structure discernable by different grey tones in BSE images. The allanite corona has a greater width (100 μ m) when the coronitic microstructures occur within the replaced garnet rims: in this case only relics of monazite and apatite are present. The thickness of the **epidote** corona ranges from few μ m adjacent to ilmenite crystals to 30 μ m in white-mica matrix. Single epidote crystals close to the corona were also observed.



Microstructural site: matrix

Figure 3.3 BSE images and schematic representations of the coronitic microstructure involving monazite. Abbreviations: Grt = garnet, Chl = chlorite, Ms = K white mica, Ilm = ilmenite, Zrc = zircon, Ep = epidote, Aln = allanite, Py = pyrite, Ap = apatite, Mnz = monazite, Pl = plagioclase.

3.4 MINERAL CHEMISTRY

The chemical compositions of white mica, chlorite, garnet, plagioclase, monazite were determined at the Institut für Mineralogie und Kristallchemie (Universität Stuttgart) using a CAMECA SX100 electron microprobe (EMP) equipped with five wavelength dispersive (WD) spectrometers. Sodium, Mg, Al, Si, K, Ca, Ti, Mn, Fe, Ba were analysed in white mica, chlorite, garnet, plagioclase employing an acceleration voltage of 15 kV and 10 or 15 nA beam current. Counting times for each element element were 20 s at the peak and the background. Analytical errors of this standard method were reported by Massonne (2012). Magnesium, Al, Si, Ca, Ti, Mn, Fe, P, Y, Dy, Er, Yb were also analysed in garnet at an acceleration voltage of 20 kV and a beam current of 100 nA. Detection limits for trace elements were 400 ppm for Yb, 400 ppm for Er, 900 ppm for Dy, 120 ppm for Y, 45 ppm for P. Sulphur, U, Th, Pb, Ca, Y, La, Ce, P, Si, Nd, Gd, Dy, Sm, and Pr were determined to analyse and date monazite, using an acceleration voltage of 20 kV and a beam current of 150 nA. Standards for all EMP analyses were synthetic and natural minerals and pure oxides. The PaP correction procedure was applied. X-ray maps for Fe, Mn, Mg, Ca in garnet were acquired with a counting time of 50 ms per step, a step width of 15 µm, and 15 kV and 70 nA as acceleration voltage and electric current, respectively. Yttrium maps were also acquired using a counting time of 180 ms and a beam current of 150 nA. A counting time of 100 ms, a step width of 1 µm, an acceleration voltage of 15 kV and a current of 30 nA were the conditions for the Fe, Mg, Ti, Na X-ray maps of potassic white mica. Structural formulae for potassic white mica, chlorite, garnet, plagioclase, were calculated with the CalcMin program (Brandelik, 2009). Monazite analyses were recalculated with the MinCalc program (Bernhardt, 2010).

Garnet porphyroblasts in Pontremoli7 show a growth zoning with an increase of the almandine (Alm) and pyrope (Prp) components and a decrease of the spessartine (Sps) component from the inner core to the outermost rim (Figs. 3.4a and 3.5a-c). The grossular (Grs) component only decreases from the outer core towards the outermost rim. Typical compositions are Alm₇₂Sps₁₂Grs₁₁Prp₄ in the inner core and Alm₈₅Prp₁₁Grs₃Sps₀ at the outermost rim (Tab. 3.1). The P content in the inner core is low (below the detection limit). In the outer core and rimwards, P increases reaching a maximum of 180 ppm (Fig. 3.4b). Garnet also shows a clear zoning profile of Y, Dy, Er, Yb: these elements are concentrated only in the inner core of the porphyroblasts, whereas their amounts drop below the detection limit of our EMP analyses (see above)

in both the outer core and the rim. In the inner core average values for Y, Dy, Er, and Yb are 5520 ppm, 1050 ppm, 900 ppm, and 1020 ppm, respectively.

Potassic white mica occurring along the schistosity and as late-stage aggregates with decussate structure consists of Na-rich cores (average Na=0.57 atoms per formula unit (apfu), which is based on 22 oxygen; average Mg=0.15 apfu) and Mg-rich rims (average Na=0.29 apfu; average Mg=0.23 apfu; Figs. 3.4c, e). Small lamellae of Mg-rich potassic white mica also occur in the recrystallized areas (maximal bending areas of late microfolds). The Na-rich potassic white mica has a Si content ranging from 6.04 to 6.20 apfu and Fe²⁺ + Mg = 0.19-0.40 apfu; the Mg-rich potassic white mica is characterized by Si = 6.04-6.38 apfu and Fe²⁺ + Mg = 0.19-0.60 apfu; potassic white mica in the garnet rim has Si = 6.06-6.39 apfu and Fe²⁺ + Mg = 0.27-0.64 apfu (Tab. 3.2). X-ray maps of potassic white mica are displayed in Fig. 3.5.

Chlorite in different microstructural sites shows limited chemical variations. $XFe=Fe^{2+}/(Fe^{2+}+Mg)$ is 0.60-0.63 for chlorite along the schistosity, 0.56-0.62 for crystals with decussate structure, 0.57-0.63 for chlorite surrounding garnet (Tab. 3.3).

Plagioclase compositions range from $Ab_{89-92}An_{8-10}Or_0$ in the core to $Ab_{98-99}An_{1-2}Or_{0-2}$ in the rim (Tab. 3.4). Plagioclase aggregates located in the garnet pressure shadows consist of homogeneous albite crystals ($Ab_{99}An_1Or_0$).

Monazite grains show no systematic chemical zoning and no chemical variations between core and rim. Nevertheless, ThO₂ contents vary between 3.2 and 10.9 wt.%, most analyses clustering between 6-8 wt.% ThO₂. UO₂ contents are in the range 0.0-0.5 wt.%, clustering around 0.3-0.5 wt.%. PbO amounts to 0.01-0.15 wt.% clustering at 0.09-0.15 wt.%. Y₂O₃ contents are low (0.01-0.4 wt.%) and the majority of the analyses gave Y < 1300 ppm. Chemical dating of monazite was performed following the approach of Montel *et al.* (1996). Detailed analytical conditions as well as comparisons of the applied EMP dating method with more precise SHRIMP and TIMS dating results on the same samples were presented by Massonne (2014). We analysed 29 point analyses in 4 monazite grains. The derived ages of 25 analyses cluster between 250 and 361 Ma with a weighted average of 294.2 \pm 4.6 Ma (average 2 σ error of the single dates; MSWD = 8.6; Tab. 3.5, Fig. 3.6, see Appendix 2 for the complete dataset). We obtained also seven younger ages (from 92 to 250 Ma), which were often obtained at the monazite rims.

Garnet		rim			outer core			inner core	
Point analyses	gt1_01a	gt1_01b	gt1_01c	gt1_04	gt1_05	gt1_05b	gt1_09b	gt1_10a	gt1_11b
wt. %									
SiO_2	35.69	35.82	35.35	35.46	35.44	35.30	35.13	34.77	34.95
Al_2O_3	21.02	21.35	21.05	21.16	21.03	21.02	20.65	20.78	20.77
Fe ₂ O ₃	3.04	3.00	3.39	3.81	3.61	3.88	2.73	2.98	2.71
TiO ₂	0.06	0.05	0.05	0.08	0.08	0.07	0.07	0.06	0.07
FeO	36.00	36.02	35.61	33.37	33.18	31.69	29.90	29.41	29.55
MnO	0.10	0.15	0.28	1.31	1.59	2.44	4.86	4.96	5.16
MgO	2.51	2.13	2.02	1.44	1.36	1.24	1.08	1.06	1.04
CaO	1.73	2.30	2.22	4.10	4.10	4.64	4.15	4.15	4.08
P_2O_5	0.03	0.02	0.01	0.02	0.01	0.02	b.d.l.	b.d.1.	b.d.l.
Y_2O_3	b.d.l.	b.d.l.	0.06	b.d.l.	b.d.l.	b.d.l.	0.77	0.80	0.85
Dy_2O_3	b.d.l.	b.d.l.	0.07	b.d.l.	0.04	b.d.l.	0.09	0.08	0.11
Er_2O_3	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	0.13	0.14	0.12
Yb_2O_3	b.d.l.	b.d.l.	b.d.1.	b.d.l.	b.d.l.	b.d.l.	0.13	0.11	0.15
Total	100.20	100.89	100.13	100.78	100.44	100.29	99.67	99.29	99.58
ppm									
Р	150.13	89.90	59.79	89.90	40.15	69.83	n.c.	n.c.	n.c.
Ti	350.11	320.14	309.95	470.02	490.40	399.87	390.28	369.90	429.85
Y	n.c.	n.c.	449.63	n.c.	n.c.	n.c.	6039.66	6319.99	6720.01
Dy	n.c.	n.c.	589.87	n.c.	n.c.	n.c.	750.19	659.57	980.21
Er	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	1129.85	1190.19	1089.63
Yb	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	1112.32	961.84	1282.72
apfu									
Si	2.895	2.889	2.879	2.868	2.877	2.869	2.889	2.871	2.879
Р	0.002	0.001	0.001	0.001	0.001	0.001	0.000	0.000	n.c.
Al ^{iv}	0.105	0.111	0.121	0.132	0.123	0.131	0.111	0.129	0.121
Al ^{vi}	1.905	1.917	1.900	1.885	1.890	1.882	1.891	1.893	1.896
Fe ³⁺	0.186	0.182	0.208	0.232	0.220	0.238	0.169	0.185	0.168
Fe ²⁺	2.442	2.429	2.426	2.257	2.253	2.154	2.057	2.031	2.036
Mn	0.007	0.010	0.019	0.090	0.110	0.168	0.338	0.347	0.360
Mg	0.304	0.257	0.245	0.174	0.164	0.150	0.132	0.130	0.128
Ca	0.150	0.199	0.194	0.356	0.357	0.404	0.366	0.367	0.360

Table 3.1 Mineral composition of garnet normalized on 8 cations. Fe^{3+} is equal to 4 - Al.

Table 3.2 Structural formulae of potassic white mica calculated on the basis of 22 oxygen. Fe^{3+}
is stoichiometric-estimated considering $Fe^{3+} = 42 - sum$ of the valences (Si, Ti, Cr, Al, Mn, Mg,
Fe)-Ba-Ca.

K-white mica									
		Schi	istosity			Decussate structure			
	Na-ric	h core	Mg-rio	Mg-rich rim Na-		ch core	Mg-ri	Mg-rich rim	
Point analyses	Ms01_03	Ms01_02	Ms24	Ms25	Ms45_05	Ms45_06	Ms44_01	Ms44_02	
SiO ₂	46.35	46.25	47.79	47.44	45.76	45.58	46.76	46.59	
TiO ₂	0.52	0.53	0.45	0.50	0.38	0.40	0.36	0.31	
Al_2O_3	35.91	35.82	32.06	32.12	36.23	36.28	32.93	32.83	
FeO	1.05	1.13	2.22	2.03	0.86	0.82	1.71	1.58	
Fe ₂ O ₃	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	
MnO	0.00	0.01	0.04	0.00	0.00	0.01	0.02	0.00	
MgO	0.87	0.88	1.75	1.54	0.61	0.63	1.29	1.36	
CaO	0.01	0.03	0.04	0.00	0.02	0.03	0.00	0.01	
Na ₂ O	2.21	2.28	0.86	0.89	2.22	2.24	0.87	0.72	
K ₂ O	8.21	8.17	10.01	9.97	8.38	8.19	10.08	10.45	
BaO	0.14	0.14	0.17	0.20	0.22	0.25	0.17	0.18	
Total	95.27	95.23	95.39	94.70	94.68	94.42	94.19	94.04	
Si	6.116	6.113	6.375	6.369	6.090	6.077	6.310	6.310	
Al ^{IV}	1.884	1.887	1.625	1.631	1.910	1.923	1.690	1.690	
Al ^{VI}	3.702	3.693	3.415	3.452	3.774	3.776	3.546	3.551	
Ti	0.052	0.052	0.045	0.050	0.038	0.040	0.036	0.032	
Fe ²⁺	0.115	0.125	0.248	0.228	0.096	0.092	0.193	0.179	
Fe ³⁺	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	
Mn	0.000	0.001	0.005	0.000	0.000	0.001	0.002	0.000	
Mg	0.171	0.172	0.348	0.308	0.121	0.125	0.259	0.275	
Ca	0.002	0.005	0.005	0.000	0.003	0.005	0.000	0.002	
Ba	0.007	0.007	0.009	0.011	0.012	0.013	0.009	0.010	
Na	0.566	0.583	0.223	0.233	0.572	0.578	0.228	0.190	
Κ	1.382	1.378	1.704	1.707	1.423	1.392	1.735	1.806	

Table 3.2 continued

K-white mica					
	Garnet rim				
Point analyses	Ms46_01	Ms46_02			
SiO ₂	46.57	46.81			
TiO ₂	0.24	0.29			
Al ₂ O ₃	35.22	33.94			
FeO	1.55	1.87			
Fe ₂ O ₃	0.00	0.00			
MnO	0.03	0.00			
MgO	0.85	1.19			
CaO	0.02	0.00			
Na ₂ O	0.76	0.81			
K ₂ O	10.07	9.92			
BaO	0.24	0.18			
Total	95.55	95.00			
Si	6.176	6.246			
Al ^{IV}	1.824	1.754			
Al^{VI}	3.681	3.583			
Ti	0.024	0.029			
Fe ²⁺	0.172	0.208			
Fe ³⁺	0.000	0.000			
Mn	0.003	0.000			
Mg	0.167	0.236			
Ca	0.003	0.000			
Ba	0.012	0.009			
Na	0.196	0.210			
K	1.704	1.688			



Figure 3.4 Chemical diagrams of garnet and potassic white mica. (a) compositional profile of almandine, spessartine, pyrope and grossular along a garnet transect; (b) Y and P profile along the same transect; (c) Si *vs.* Al plot for potassic white mica in different microstructural positions; (d) Si *vs.* Mg plot for potassic white mica in different microstructural positions (e) Si *vs.* Na for potassic white mica.



Figure 3.5 X-ray maps of garnet and potassic white mica. (a) Mn in garnet; (b) Ca in garnet; (c) Y in garnet; (d) Mg in potassic white mica along the schistosity; (e) Na in potassic white mica along the schistosity.

Chlorite						
Schistosity			Decussate str	becussate structure Garnet rim		
Point analyses	Chl01	Chl02	Chl10	Chl03	Chl07	Chl08
SiO ₂	23.53	24.22	24.01	24.12	24.05	23.74
TiO ₂	0.06	0.08	0.07	0.07	0.07	0.04
Al ₂ O ₃	20.67	21.72	20.95	20.84	21.03	21.86
FeO	31.76	31.26	31.63	30.08	29.51	31.58
MnO	0.18	0.13	0.14	0.10	0.10	0.09
MgO	10.65	11.30	11.26	11.53	12.41	10.57
Total	86.84	88.71	88.07	86.73	87.17	87.89
Si	5.218	5.215	5.232	5.293	5.235	5.175
Al ^{IV}	2.782	2.785	2.768	2.707	2.765	2.825
Al^{VI}	2.619	2.726	2.611	2.681	2.631	2.793
Ti	0.010	0.013	0.011	0.011	0.011	0.006
Fe	5.890	5.628	5.763	5.520	5.370	5.758
Mn	0.033	0.023	0.026	0.019	0.019	0.017
Mg	3.519	3.626	3.655	3.770	4.026	3.435

Table 3.3 Structural formulae of chlorite calculated on the basis of 28 oxygen and the negligence of Na and Ca. Chlorite is assumed to be completely hydrated (16 H).

Table 3.4 Structural formulae of plagioclase based on 8 oxygen.

Plagioclase									
		Core			Rim			Garnet pressure shadows	
	Pl02_03	Pl02_04	Pl01_04	P101	P104	P105	P107	P108	
SiO ₂	66.27	66.60	66.30	68.59	68.74	68.86	68.74	68.76	
Al_2O_3	21.40	21.52	21.32	19.31	19.28	19.30	19.05	19.29	
Fe_2O_3	0.05	0.09	0.10	0.13	b.d.l.	0.04	0.13	0.15	
CaO	1.98	2.01	2.20	0.33	0.20	0.15	0.12	0.25	
Na ₂ O	10.94	10.91	10.88	11.81	11.86	12.01	12.34	12.08	
K_2O	0.07	0.05	0.05	0.02	0.02	0.00	0.05	0.03	
BaO	b.d.l.	0.02	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	b.d.l.	
Total	100.71	101.20	100.86	100.18	100.11	100.36	100.44	100.55	
Si	2.894	2.894	2.892	2.994	3.000	2.999	2.998	2.993	
Al	1.102	1.102	1.096	0.993	0.992	0.990	0.979	0.990	
Fe ³⁺	0.002	0.003	0.003	0.004	n.c.	0.001	0.004	0.005	
Ba	n.c.	0.000	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	
Ca	0.093	0.094	0.103	0.015	0.010	0.007	0.006	0.012	
Na	0.926	0.919	0.920	1.000	1.004	1.014	1.044	1.019	
K	0.004	0.003	0.003	0.001	0.001	0.000	0.003	0.001	

Monazite								
Point analyses	Mnz04_01	Mnz01_02a	Mnz01_08a	Mnz02_04a	Mnz02_05a	Mnz04_07a	Mnz02_01a	Mnz01_01a
SiO	0.58	0.81	0.80	0.91	0.58	0.63	0.49	0.50
P ₂ O ₂	29.34	27.30	26.28	29.80	29.98	30.33	30.51	30.77
SO ₂	0.03	0.02	0.04	0.06	0.03	0.02	0.02	0.01
CaO	1.54	1.58	1.78	1.88	1.47	1.51	1.40	1.58
Y_2O_2	0.05	0.09	0.08	0.05	0.01	0.03	0.09	0.10
	14.56	13.42	13.72	13.84	15.39	14.86	14.69	14.67
Ce_2O_3	28.64	27.53	26.80	26.70	29.16	28.70	29.02	29.10
Pr ₂ O ₃	3.05	2.65	2.79	2.93	3.04	3.06	3.09	3.20
Nd ₂ O ₃	11.95	11.48	11.12	11.33	11.63	11.87	12.19	12.26
Sm_2O_3	2.11	2.05	1.97	2.04	1.54	1.90	2.15	2.14
Gd_2O_3	1.54	1.57	1.50	1.46	0.79	1.18	1.59	1.55
Dy ₂ O ₃	0.20	0.27	0.21	0.21	0.11	0.17	0.27	0.27
PbO	0.10	0.13	0.14	0.15	0.11	0.12	0.09	0.10
ThO_2	6.89	9.08	9.62	10.17	7.44	7.69	5.87	5.76
UO ₂	0.42	0.48	0.50	0.52	0.42	0.47	0.43	0.43
Total	100.99	98.46	97.35	102.05	101.70	102.55	101.88	102.43
Si	0.023	0.033	0.035	0.022	0.024	0.019	0.031	0.032
Р	0.972	0.946	0.972	0.981	0.982	0.991	0.981	0.945
S	0.001	0.001	0.002	0.001	0.001	0.000	0.001	0.001
Ca	0.065	0.069	0.077	0.061	0.062	0.064	0.056	0.066
Y	0.001	0.002	0.001	0.000	0.001	0.002	0.001	0.002
La	0.210	0.203	0.197	0.219	0.210	0.206	0.211	0.210
Ce	0.410	0.413	0.376	0.412	0.402	0.405	0.397	0.415
Pr	0.043	0.040	0.041	0.043	0.043	0.044	0.044	0.043
Nd	0.167	0.168	0.156	0.161	0.162	0.167	0.166	0.172
Sm	0.028	0.029	0.027	0.021	0.025	0.028	0.027	0.028
Gd	0.020	0.021	0.019	0.010	0.015	0.020	0.019	0.021
Dy	0.003	0.004	0.003	0.001	0.002	0.003	0.002	0.003
Pb	0.001	0.001	0.002	0.001	0.001	0.001	0.001	0.001
Th	0.061	0.085	0.089	0.065	0.067	0.050	0.059	0.076
U	0.004	0.004	0.004	0.004	0.004	0.004	0.003	0.004
AGE	289.58	294.37	294.07	293.1	312.09	314.01	294.84	292.09
statistical error	7.21	5.88	5.49	6.81	6.69	8.18	7.68	6.24

Table 3.5 EMP age and mineral chemistry of monazite on the basis of 4 oxygen.



Figure 3.6 Monazite composition and EMP age. (a) UO₂ vs. ThO₂ diagram of the monazite analysed in the Pontremoli sample and other monazite occurrences (Rasmussen & Muhling, 2007; Janots *et al.*, 2008; Langone *et al.*, 2011); (b) relative probability plot and histogram of the monazite EMP ages.

3.5 GEOTHERMOBAROMETRY

3.5.1 Calculation of the bulk-rock chemistry

We performed thermodynamic modelling in order to derive the P-T path of the Pontremoli sample (Pontremoli7) and, in particular, the thermobaric conditions during the growth of chemically different garnet domains. One of the input parameters to calculate isochemical phase diagrams, so-called pseudosections, is the bulk-rock composition. Because of the limited availability of the cuttings, it was not possible to obtain sufficient material to prepare powders for conventional XRF analyses. Moreover, the presence of zoned garnet required the calculation of different bulk compositions to take into account possible modifications of the effective bulk compositions (Stüwe, 1997) due, for instance, to the shielding of inner garnet zones at late stage metamorphic equilibration. For these reasons, we calculated three different bulk compositions to

model different stages of the metamorphic evolution. The initial, unfractionated bulkrock chemistry was calculated by combining mineral modes and average mineral compositions (bulk1). We also calculated the bulk composition after the fractionation of the entire garnet core (bulk2) and the whole garnet (bulk3).

As the composition bulk1 includes the entire garnet and the Pontremoli7 thin section represents a garnet-rich area, we averaged the mode of Pontremoli7 and AG7 samples to avoid garnet overestimation. The Pontremoli 7 and AG7 modal compositions were obtained by counting 1000 points for each thin section, using a polarizing optical microscope equipped with a mechanical point-counter. The calculated bulk composition of the selected sample is Ca-poor and Mn-rich compared to the average composition of pelites given by Shaw (1956). Whole-rock Ca and Al amounts normalized to Shaw's (1956) data plot along the reaction line of the monazite to allanite reaction inferred by Wing *et al.* (2003). This agrees with the occurrence of allanite replacing monazite in the studied sample.

The garnet core composition was calculated using mineral chemistry and the garnet core mode. The composition of the garnet core was subtracted from the unfractionated XRF bulk-rock composition (in mol units), obtaining the "bulk2" composition.

We determined the volumes of the garnet core and rim in Pontremoli 7 through image analysis of X-ray maps. To apply this approach the required parameters are composition and density of the unfractionated rock, and composition, volume and density of the fractionated phase. For the digital image analysis of two garnets we applied the JMicroVision 1.2.7 software (http://www.jmicrovision.com/). The garnet core volume resulted in 9 % vol. of the whole rock. The garnet composition taken into account is an average of the inner core and outer core analyses. The density of the garnet inner core (4.15 g cm⁻¹) and density of the bulk-rock (2.91 g cm⁻¹) were obtained from the thermodynamic output in the inner core intersection area. The composition bulk3 was finally calculated excluding the whole garnet by the method outlined above. The resultant compositions are reported in Table 3.6.

	unfractioned composition (bulk1)	fractioned compositions		
		garnet inner and outer	whole garnet fract.	
		core fract. (bulk2)	(bulk3)	
	wt. %	wt. %	wt. %	
SiO ₂	53.31	57.03	57.85	
TiO_2	1.17	1.31	1.36	
Al_2O_3	24.22	24.18	24.32	
FeO	12.00	8.26	7.13	
MnO	0.44	0.03	0.00	
MgO	2.56	2.71	2.72	
CaO	0.62	0.09	0.00	
Na ₂ O	1.19	1.33	1.38	
K_2O	4.49	5.06	5.25	

Table 3.6 Calculated bulk-rock compositions with garnet (bulk1) and without garnet core (bulk2) and the whole garnet (bulk3).

Footnote: To convert vol. % in wt. % we used the densities of minerals reported in Anthony *et al.* (2011). The densities of the minerals that show significant solid solution were calculated through a linear interpolation between the densities of the corresponding end-members.

3.5.2 Phase diagram calculations

Thermodynamic modelling in the MnO-K₂O-Na₂O-CaO-FeO-MgO-Al₂O₃-SiO₂-H₂O-TiO₂ system was performed with the PERPLE_X computer package (Connolly 1990; Connolly & Petrini, 2002). Ti was considered to take into account the relationships among Ti-rich minerals (rutile, ilmenite and titanite). Ferric iron was not introduced in the chemical system because of the lack of Fe³⁺-rich oxides and the negligible stoichiometrically calculated Fe³⁺ content in the major silicates (Lo Pò & Braga, 2014). Neglecting ferric iron implies that the Fe³⁺-rich minerals (allanite and epidote) cannot be thermodynamically properly modelled. They are represented by Fe-free zoisite. Nevertheless, we think that this simplification is adequate also because of the low modal amounts (< 1 vol. %) of allanite-epidote. The following input parameters were chosen: (a) the thermodynamic dataset of Holland & Powell (1998, with the 2002 upgrade); (b) the compensated Redlich Kwong (CORK) fluid equation of state by Holland & Powell (1991, 1998); (c) SiO_2 and H_2O in excess. The solid solution models used in the thermodynamic modelling were: K-white mica (Holland & Powell, 1998; the maximum paragonite content has been set to 50 mol.%), Na-Ca-white mica (Massonne, 2010; the maximum muscovite content has been set to 50 mol.%), garnet (Holland & Powell, 1998), biotite (Powell & Holland, 1999), chlorite (Holland et al., 1998), plagioclase (Fuhrman & Lindsley, 1988), K-feldspar (Waldbaum & Thompson, 1968), staurolite (Holland & Powell, 1998) and ilmenite (ideal ilmenite-geikielite-pyrophanite solid solution; http://www.perplex.ethz.ch/PerpleX_solution_model_glossary.html). Pure phases in the calculations were titanite, rutile, zoisite, sillimanite and andalusite.

3.5.3 Prograde path before peak conditions (bulk1)

An isochemical phase diagram was calculated for bulk1 and the P-T range 2-15 kbar and 300-600 °C (Fig. 3.7). Potassic white mica is stable throughout the entire *P*-*T* range. Chlorite and Na-Ca white-mica disappear at temperature higher than 550°C at 3 kbar and 600°C at 6-7 kbar, replaced by staurolite and plagioclase. The garnet-in curve is located at 400°C at 5 kbar to 450°C at 3 kbar. Biotite occurs above 530 °C. The plagioclase-in curve is located below 7 kbar. The P-T fields with zoisite terminate between 450°C and 530°C. Ilmenite is replaced by rutile only at relatively low temperatures and high pressures. Titanite-bearing P-T fields occur below 450 °C. The paragenetic field containing chlorite + potassic white mica + sodic white mica + ilmenite + rutile + titanite + quartz + K-feldspar plots at 350-400°C and P < 3.5 kbar. Isopleths for molar fractions of garnet related to the garnet inner core Alm₇₂Sps₁₂Grs₁₁Prp₄ intersect in a small area at 530 °C and 5.5 kbar within this paragenetic field albeit without sodic white mica. The resulting prograde path runs from 350-400°C and P < 5 kbar to 530 °C and 5.5 kbar, crossing the rutile/ilmenite transition and the garnet-in curve, in agreement with the observed petrographic and microstructural features. The plagioclase-in and zoisite-out curves are also crossed. Corresponding isopleths for the garnet outer core intersect outside the investigated P-Trange, but this can be due to the negligence of a fractionation effect. The calculated Si content in potassic white mica is 6.14-6.16 apfu in the early paragenesis field, and 6.12-6.14 apfu in the garnet inner core intersection area. Such values are within the observed Si range of the Na-rich potassic white mica (Si=6.04-6.20 apfu). The calculated XFe of chlorite is 0.73 apfu in the early paragenesis field and 0.56-0.69 apfu in the garnet inner core intersection area. The observed XFe of chlorite along the schistosity is 0.56-0.62 apfu.

3.5.4 Peak conditions after early garnet fractionation (bulk2)

The composition bulk2 excludes the garnet core. Therefore, only the rim of the garnet was considered for P-T estimations. In the P-T pseudosection for bulk2 (Fig. 3.8) the

topology is different compared to that of bulk1. Potassic white mica is again stable in the entire *P*-*T* range, but chlorite occurs only at temperatures below 500°C. Na-Cawhite-mica disappears at temperature higher than 525°C at 2 kbar and 600°C at 5 kbar in the *P*-*T* pseudosection for bulk2. The staurolite-in curve runs from 530°C at 3 kbar to 600°C at 7 kbar. Ilmenite replaces rutile above 450°C. The *P*-*T* field of garnet is reduced compared to that of bulk1. The garnet isopleths geothermobarometry defines *P*-*T* conditions around 575°C and 7 kbar for the rim compositions (Alm₈₃₋₈₅Prp₇₋₁₁Grs₃₋₈). These conditions are compatible with the paragenetic fields containing potassic white mica, sodic white mica, chlorite, garnet, ilmenite, biotite, staurolite, and quartz. The *P*-*T* isopleth for the maximum Si content in Na-rich potassic white mica (6.20 apfu) intersects the garnet compositional isopleths.



Figure 3.7 *P*-*T* pseudosection calculated with the unfractionated bulk-composition (bulk1). The dotted area outlines the field corresponding to an early paragenesis and the small triangle is the inner core garnet isopleths area. A *P*-*T* uncertainty ellipse is also displayed. Abbreviations: Ms = potassic white mica (muscovite/phengite), Pg = sodic white mica, Chl = chlorite, Kfs = alkalifeldspar, Pl = plagioclase, Grt = garnet, Bt = biotite, St = staurolite, And = andalusite, Crd = cordierite, Ilm = ilmenite, Rt = rutile, Ttn = titanite, Lws = lawsonite, Zo = zoisite, Sil = sillimanite, Cld = chloritoid, Ab = albite.



Figure 3.8 *P*-*T* pseudosection calculated for an effective composition after the fractionation of the garnet core (bulk2). The circle corresponds to the garnet rim composition with a *P*-*T* uncertainty ellipse. The maximum Si content in the potassic white mica core (Na-rich) is also displayed.

3.5.5 Retrograde path (bulk3)

Since the garnet rim was in equilibrium with the Na-rich core of the potassic white mica crystals, the Mg-rich rims of the potassic white mica in aggregates along the schistosity and with decussate microstructure should have crystallised later together with chlorite in the same microstructural position. Thermodynamic modelling of this stage is based on composition bulk3 excluding the whole garnet (Fig. 3.9). Because no significant Mn and Ca was left, the KNFMASHT chemical subsystem was selected for corresponding calculations. Potassic white mica is again stable in the entire *P-T* range of the pseudosection for bulk3. Na-white mica breaks down above $525^{\circ}C$ at 2 kbar and $600^{\circ}C$

at 5 kbar. Chlorite occurs below 550°C. Biotite forms at temperatures of 500°C and more. Staurolite occurs around 550-600°C and above 3 kbar. Albite appears in the range 300-500°C at pressures below 5 and 8 kbar, respectively. At higher temperatures (550-600°C) the maximum pressure for albite occurrence is 5 kbar. The maximum Si content in Mg-rich potassic white mica (6.38 apfu) and the minimum XFe ratio in chlorite (0.56) with decussate structure indicate *P*-*T* conditions of 520°C and 8 kbar by intersecting isopleths. These conditions are in the paragenetic field potassic white mica + chlorite + ilmenite + sodic white mica + quartz. The lower values of Si in Mg-rich potassic white mica and the observed paragenesis potassic white mica + chlorite + albite + ilmenite + quartz allow to constrain the retrograde evolution passing through T=500°C and P=2 kbar.



Figure 3.9 *P*-*T* pseudosection calculated for an effective composition after the fractionation of the entire garnet (bulk3). Maximum Si content in the potassic white mica rim (Mg-rich) and the minimum XFe in chlorite allow the estimation of the pressure peak. The ellipse represents the *P*-*T* uncertainty. Retrograde paragenesis is delimitated by the green line.

3.6 DISCUSSION

3.6.1 The identification of inherited monazite in the Pontremoli micaschist

Most of the monazite analyses in the Pontremoli micaschist yielded ThO₂ and UO₂ contents in the range of 6-8 and 0.3-0.5 wt.%, respectively. The ThO₂ and UO₂ contents in low-grade, fluid-derived monazite are generally < 2 wt. % and 0-0.1 wt.%, respectively. These contents are distinct from igneous/high-grade monazite that shows ThO₂ = 3-6 wt.% and UO₂ ~ 0.2-0.3 wt.% (Schandl & Gorton, 2004; Rasmussen & Muhling, 2007; Langone *et al.*, 2011). Thus, high ThO₂ and UO₂ contents identify igneous or medium- to high-grade monazite. Thus, the monazite from Pontremoli micaschist resembles monazite formed at medium-high rather than low-medium grade conditions inferred for the prograde path of the Pontremoli micaschist.

The partitioning of Y between a Y-rich mineral and coexisting garnet is a useful tool to assess the mutual relationships between the two phases. The Y content in the garnet falls from 4550-6800 ppm (inner core) to values close to the detection limit in the outer core and in the rim. The observed Y zoning can be explained with the consumption of a Y-rich phase during the crystallization of the garnet inner core. The low (< 1300 ppm) Y concentration in monazite excludes monazite from being the source of Y for the garnet inner core. The Y and the HREE concentrations in garnet are instead controlled by xenotime consumption because this mineral with resorption features was found only in the garnet inner core. When xenotime was almost completely dissolved and all Y was introduced in the inner garnet core, no significant amounts of Y were available during the growth of the garnet outer core and rim. This is in agreement with the results of Spear & Pyle (2010), who demonstrated that xenotime becomes instable at pressures above 5-6 kbar. Moreover, these authors suggest, through theoretical modelling, that during garnet growth xenotime commonly breaks down and this explain why xenotime is found only in the garnet inner core and not in the matrix. Most of the monazite analyses in the Pontremoli micaschist yielded ThO2 and UO2 contents in the range of 6-8 and 0.3-0.5 wt.%, respectively. The ThO₂ and UO₂ contents in low-grade, fluidderived monazite are generally < 2 wt. % and 0-0.1 wt.%, respectively (Rasmussen & Muhling, 2009). Such low contents are distinct from igneous and high-grade metamorphic monazite that usually shows ThO₂ contents of 3-6 wt.% and UO₂ contents of 0.2-0.3 wt.% (Schandl & Gorton, 2004; Rasmussen & Muhling, 2007; Langone et al., 2011). Thus, the observed high ThO₂ and UO₂ contents point to igneous or mediumto high-grade monazite in the Pontremoli micaschist and not to the low-medium grade conditions inferred for the prograde path of the Pontremoli micaschist by other methods. The partitioning of Y between a Y-rich mineral and coexisting garnet is a useful tool to assess the mutual relationships between the two phases. The Y content in the garnet falls from 4550-6800 ppm (inner core) to values close to the detection limit in the outer core and in the rim. The observed Y zoning can be explained by the consumption of a Y-rich phase during the crystallization of the garnet inner core. The low (< 1300 ppm) Y concentrations in monazite excludes this mineral from being the source of Y for the garnet inner core. The Y and the HREE concentrations in garnet are instead controlled by consumption of xenotime which shows resorption features and occurs exclusively in the garnet inner core. After xenotime was almost completely dissolved and nearly all Y was introduced in the inner garnet core, no significant amounts of Y were anymore available during the growth of the garnet outer core and rim. This is in agreement with the results of Spear & Pyle (2010), who demonstrated that xenotime becomes instable at pressures above 5-6 kbar. Moreover, these authors suggested, through theoretical modelling, that during garnet growth xenotime commonly breaks down and this explain why xenotime is found only in the garnet inner core and not in the matrix.

3.6.2 P-T condition and garnet response to REE-accessory mineral reactions

Textural evidence indicates that the Th-rich monazite became unstable and underwent partial or total dissolution. This process led to the formation of a complex corona made of different REE-minerals. The sequence of these REE-minerals forming the multiple corona microstructure around monazite can be correlated with a change of the kind of mineral inclusions in the garnet porphyroblasts where single allanite crystals are hosted only in the garnet outer core whereas the garnet rim hosts epidote (Figs. 3.3 e, f). The garnet zoning also reflects the order of growth of the different REE-minerals. While the usual Fe, Mn and Mg growth zoning in garnet can be observed, its Ca profile is more complex, with lower values in the inner core than in the outer core (Figs. 3.4a, 3.5b). From the garnet outer core to the rim the Ca content decreases gradually, following the common grossular prograde pattern. Similarly to Ca, also P is low in the garnet inner core and then increases towards the garnet rim. The humps in the Ca profile and the low P content in the garnet inner core may be explained by the contemporary crystallization of apatite in the matrix, replacing the pre-existing monazite.

As inferred by Finger et al. (1998) for a similar occurrence in amphibolite-facies

metagranitoids, the original monazite may provide the phosphorous necessary for the apatite growth, since apatite always surrounds monazite (Fig. 3.3). Thorium, previously hosted in monazite, partitions into Th-silicates and apatite in the corresponding corona. The anorthite component in plagioclase and biotite may provide Ca and Fe required for allanite and epidote growth (Finger *et al.*, 1998).

Based on the above considerations, we infer the following correspondence between garnet zoning and the crystallization of the REE-minerals produced during the consumption of the inherited monazite (Fig. 3.10): 1) garnet inner core - crystallization of apatite and Th-silicate, 2) garnet outer core - crystallization of allanite, 3) garnet rim - crystallization of epidote.

If this reconstruction is correct, we can correlate the thermobaric conditions of the garnet growth to the accessory mineral reactions. Since uncertainties involving thermodynamic modelling must be considered, we follow the suggestion by Massonne (2013) who proposed 2σ -errors in the range of 10% and 5% related to the corresponding pressure and temperature estimate, respectively. For this reason we infer that the partial destruction of monazite and the development of the apatite, allanite and epidote coronae occurred presumably at 500-600 °C and 5-7 kbar during the prograde evolution. This assumption is compatible with the Y-Al-garnet (YAG) thermometry (Pyle & Spear, 2000) that yields, for the garnet inner core, an average temperature of 562°C (\pm 30°C). Interestingly, the *P-T* estimates of several amphibolite-facies metagranitoids with apatite-allanite-epidote coronas around monazite reported in Finger *et al.* (1998) were also metamorphosed in the range of 500–600 °C and 4–7 kbar.

Since monazite predates the garnet, the monazite age of 294 ± 5 Ma must be older than the garnet forming metamorphic stage.



Figure 3.10 *P*-*T* evolution of the sequence of the REE-minerals related to garnet growth.

3.6.3 The monazite-allanite transition

Previous studies based on natural assemblages (summarised by Tomkins & Pattinson, 2007) indicate that allanite forms from a monazite precursor at the biotite-in isograd. (~ 300-400°C; Spear, 1993). Moreover, the thermodynamic calculations by Janots et al. (2007) in the system SiO_2 -Al₂O₃-FeO-Fe₂O₃-MgO-CaO-Na₂O-K₂O-P₂O₅-La₂O₃-CO₂-H₂O predict that the reaction of monazite to form allanite starts at temperatures as low as 300°C. The results of our study, however, show that the monazite-allanite transition falls in the 500-600 °C (at 5-7 kbar) range. This discrepancy may be due to (i) the uncertainties regarding the calorimetric data for synthetic monazite(-La), LaPO₄, and dissakisite, CaLaMgAl₂(SiO₄)₃OH, (ii) the simplification that these end-member phases are good analogues for natural monazite(-Ce) and Fe-allanite(-Ce) (Janots et al., 2007), (iii) the differences in the bulk-rock compositions, especially in the Ca content, considered here (CaO = 0.62 wt. %) and used (0.88 wt. %) by Janots *et al.* (2007). A lower CaO content might favour the stability of monazite over allanite, thus, probably shifting the reaction monazite = allanite to higher temperatures (Foster & Parrish, 2003; Wing et al., 2003). In the diagram Ca_{sample}/Ca_{Shaw's average pelite} vs. Al_{sample}/Al_{Shaw's average} pelite for samples from the biotite and garnet zone (Wing et al., 2003), the Pontremoli metapelite plots along the line in which metamorphic allanite and low-grade monazite coexist.

3.6.4 *P*-*T* path

The derived *P*-*T* path (Fig. 3.11) starts at 350°C and P < 5 kbar (rutile and K-feldspar bearing paragenesis). The chlorite + potassic white mica + sodic white mica + ilmenite + rutile + titanite + quartz + K-feldspar paragenesis ($350-400^{\circ}$ C and P < 3.5 kbar) is in agreement with the K-feldspar inclusions in plagioclase and the rutile/ilmenite transition and may represent an early metamorphic stage. At increasing temperature, rutile breaks down and the calculated field chlorite + potassic white mica + ilmenite + sodic white mica + titanite + quartz could correspond to the assemblage that formed the schistosity now represented only by the ilmenite S_i in garnet. With a further increment of temperature the garnet-in curve was crossed, and conditions of 530 °C and 5.5 kbar, obtained by garnet inner core isopleths, were reached. Zoisite and titanite of the prograde path have not been observed in the studied rock sample, however the calculated modes of these phases are less than 2 %. The temperature derived from the garnet inner core isopleths is 130°C higher than the temperatures of the garnet-in curve. This may be due to reaction overstepping or to an imperfect correspondence of the maximum-Mn analysis of garnet and the true Mn-content of the garnet innermost core. The peak conditions of 575°C and 7 kbar (garnet rim and potassic white mica cores) were followed by a stage at higher pressures (520°C and 8 kbar) that stabilized the Mgrich potassic white mica and chlorite with decussate structure. The retrograde evolution points to $T=500^{\circ}C$ and P=2 kbar. The lack of biotite at the peak conditions may be due to its complete retrocession to chlorite and potassic white mica during the retrograde evolution. Sodic white mica is always present in the P-T pseudosection along the derived path, but distinct crystals have not been observed. Thus, we think that sodic white mica was replaced by potassic white mica and albite during the retrograde evolution. In Fig. 3.12 the main observed and computed parageneses of the pre-peak stage, temperature and pressure peak stages and retrograde stage are displayed.



Figure 3.11 Obtained *P-T* path for the selected sample. The anticlockwise trajectory passes through a metamorphic peak at 575°C and 7 kbar, followed by the pressure peak at 8 kbar during the retrograde evolution.

3.6.5 Tectonic consequences

The pre-Oligocene metamorphic record in the late Variscan basement of the Northern Apennines is evident for the Larderello and Pontremoli subsurface basement (Bertini *et al.*, 1994; Pandeli *et al.*, 2005) and for the Cerreto micaschist outcrops (Molli *et al.*, 2002). In the Larderello and Cerreto basements, only retrograde segments of Variscan *P*-*T* paths (Bertini *et al.*, 1994; Molli *et al.*, 2002) are available with no information on the prograde metamorphic stages. The reconstruction of a complete *P*-*T* loop for the Pontremoli hidden basement provides important insights into the late- to post-Variscan geodynamic evolution in this sector of the Mediterranean area. In the Pontremoli micaschist, the monazite chemical age of 294 ± 5 Ma obtained here from monazite relics represents a lower age limit for the metamorphic evolution that took place from Early Permian times onwards. The Early Permian monazite age can refer to a medium-high grade metamorphic stage unrelated to the Pontremoli metamorphic history. This age is nearly coeval with the 275-285 Ma Rb/Sr bulk rock age obtained for the Variscan

1985) and with the 285 ± 11 Ma Rb/Sr age for a muscovite associated with andalusite in a Larderello micaschist (Del Moro *et al.*, 1982), interpreted as a late Variscan thermal stage. The emplacement of plutons, which accompanied the collapse of the Variscan orogen, occurred in late Carboniferous-early Permian times (Mohn *et al.*, 2010 and references therein). The Early Permian magmatism is characterised by intrusions at different crustal levels, from the crust-mantle boundary (e.g. the Braccia Gabbro; Hermann *et al.*, 2001) to low- and mid-crustal (the Ivrea Mafic complex at 288 Ma: Demarchi *et al.*, 1998; Peressini *et al.*, 2007; the Sondalo gabbroic complex at 290 Ma: Tribuzio *et al.*, 1999; Braga *et al.*, 2001; the Collon Gabbro at 284 Ma: Monjoie *et al.*, 2005) and shallow depths (Bocca di Tenda Gabbro at 280-283 Ma, Corsica: Tribuzio *et al.*, 2009; Renna *et al.*, 2013).

Late-Variscan magmatic activity has been recognized also in the Penninic domain and Internal Massif of the Western Alps (Tilton et *al.*, 1997; Bertrand *et al.*, 2000; Liati *et al.*, 2001; Bertrand *et al.*, 2005; Ring *et al.*, 2005), in the Malenco Unit of the Central Alps (Hermann & Rubatto, 2003), in Sardinia (307-299 Ma; Del Moro *et al.*, 1975; Bralia *et al.*, 1981; Poli *et al.*, 1989), in the Serre Massif in Calabria (Schenk, 1980; Graessner *et al.*, 2000), and in the Betic Cordillera (Zeck & Whitehouse, 1999; Esteban *et al.*, 2011). In the Tuscan Metamophic Units Late-Variscan magmatic bodies have not been found. However, the Liguride Units contain gabbro-derived granulites with a 290 Ma age related to the emplacement of the protoliths: Meli *et al.*, 1996; Montanini & Tribuzio, 2001). The here obtained younger chemical ages in the Pontremoli micaschist (19-250 Ma) may represent mixed ages between late Variscan and Alpine events. As a consequence, more age constraints (e.g. ⁴⁰Ar-³⁹Ar dating of potassic white mica) are needed to better reconstruct the timing of tectonic events occurred from late-Variscan times onward. Nevertheless, we are able to infer two different sequences of events:

A first scenario considers the reconstructed *P*-*T* path for the Pontremoli micaschist (counterclockwise segment followed by a nearly isothermal decompression) and its relation to the activity of a regional scale shear zone accommodating transpressive (conterclockwise - T_{max} before P_{max}) to transtensive (near isothermal decompression) deformation (McCann *et al.*, 2006). Transpressive tectonics has been documented in Mid-Permian times (Cadel, 1986; Cadel *et al.*, 1996; Deroin & Bonin, 2003), while several Variscan basement complexes record extensional deformation from very late Carboniferous to Mid Triassic times (e.g. 280-210 Ma in the Ivrea lower crust; Brodie *et al.*, 1989). This frame is in line with regional scale considerations, suggesting the

importance of intracontinental strike-slip tectonics (Arthaud & Matte, 1977; Doglioni 1984; Massari, 1986; Ziegler, 1986; Handy & Zingg, 1991; Schmid, 1993; Matte, 2001), and with the depositional history of Upper Carboniferous-Mid Triassic sediments of the Tuscan Domain (Rau, 1990; Molli, 2002).

The second scenario considers the role of the post-Oligocene evolution of the Northern Apennine chain during the collision of the Sardinia-Corsica block and the Adria margin. Most of the youngest monazite ages obtained here are within the 27-11 Ma age range defined by Ar-Ar dating of white micas from the Alpi Apuane phyllites (Kligfield *et al.*, 1986). These ages were related to underplating of the Adria microplate below the Corsica-Sardinia block and subsequent early exhumation of the Adria continental margin (Fellin *et al.*, 2007; Molli, 2008). In this frame the tectonic loading during the thickening stages may explain the anticlockwise segment of the *P*-*T* path for the Pontremoli micaschist. This loading was followed by rapid exhumation due to crustal extension connected to the opening of the Tyrrhenian Sea. The uncertainties connected with the analysis of young monazite ages with the EMP and the significant difference between the *P*-*T* paths of the Pontremoli micaschist and adjacent basement complexes make it difficult to relate the entire metamorphic evolution of the Pontremoli micaschist to late Alpine events.

As a whole, although we cannot definitively rule out the involvement of the Pontremoli micaschist in the Oligocene-Miocene Apennines-related deformation and metamorphism, our favourite scenario relates the Pontremoli basement evolution to a regional scale strike-slip shear zone as part of the South Europe post-Variscan shear belts accommodating lateral displacement between Africa and Europe (Arthaud & Matte, 1977; Doglioni 1984; Massari, 1986; Ziegler, 1986; Handy & Zingg, 1991; Schmid, 1993; Matte, 2001). In this view, the correlation of Pontremoli basement rocks with those of the Micaschist Complex at Larderello (Pandeli *et al.*, 2005) should be reconsidered.

	PRE-PEAK STAGE	-PEAK AGE TEMPERATURE STAGE		PEAK PRESSURE STAGE	RETROGRADE STAGE	
MICROSTRUCTURES	Rutile inclusions within ilmenite; K-feldspar inclusions within plagioclase	Rim In Court Na-rich Ms Garnet porr and Na-ric potassic wh	Grt mer er core ohyroblasts h cores in nite mica	Chl Mg-rich Ms rim Ms rim Na-rich Mg-rich rims around Na-rich-cores in potassic white mica and chlorite with decussate structure. Mg-rich potassic white mica occurs also in the recrystallized areas	Grt Chl Ms Ms Qtz Ab Late observed paragenesis	
CONSTRAINING PHASES	Garnet-free and rutile- and K-feldspar bearing paragenesis	Garnet control $INNER$ CORE Alm_{72} Sps_{12} Grs_{11} Prp_4	RIM Alm_{83-85} Grs_{3-8} Prp_{7-11} Si max in potassic white mica cores=6.20	Si max=6.28 in potassic white mica rims and XFe=0.56 in chlorite	Late paragenesis and low Si content in potassic white mica	
COMPUTED ASSEMBLAGE	Ms Chl Pg Ilm Rt Ttn Kfs Qtz	Ms Chl Pg Pl Ilm Grt Qtz	Ms Chl Pg Ilm Grt Bt Qtz	Ms Chl Pg Ilm Qtz	Ms Chl Ab Ilm Qtz	
OBSERVED ASSEMBLAGE	Ms Chl Ilm Rt Kfs Qtz	Ms Chl Pl Ilm Grt Qtz	Ms Chl Ilm Grt Qtz	Ms Chl Ilm Qtz	Ms Chl Ab Ilm Qtz	
T P	350 °C < 3.5 kbar	530 °C 5.5 kbar	575 °C 7 kbar	520 °C 8 kbar	500 °C 2-3 kbar	

Figure 3.12 Scheme with the main petrographic and chemical constrains and the computed <i>P</i> - <i>T</i>
conditions.

3.7 CONCLUSIONS

- Discontinuous variations of Ca and Y in chemical profiles of garnet can be attributed to its growth during apatite crystallization and xenotime breakdown. Thus, garnet compositions is able to record modifications in accessory minerals of mineral assemblages.
- Coronitic microstructures of accessory minerals and changes of the inclusion mineralogy of garnet provide microstructural evidences of the reaction history. Through garnet isopleths thermobarometry the *P*-*T* conditions of the reactions among REE-minerals were inferred: the sequence monazite → apatite + Th-silicate → allanite →epidote occurred during prograde metamorphism at 500-600°C and 5-7 kbar.
- 3. The chemistry and microstructure of the monazite grains suggest that the monazite was inherited and became unstable along the *P*-*T* path towards peak metamorphic conditions, undergoing partial to total dissolution. Thus the EMP age of 294 ± 5 Ma should be older than the garnet forming metamorphic stage, arguing for a post-Lower Permian metamorphic evolution.
- 4. The Pontremoli metasediments experienced an anticlockwise *P*-*T* path: the prograde evolution started at 350°C and P < 5 kbar (rutile and K-feldspar bearing paragenesis and reached the metamorphic peak at 575°C and 7 kbar (garnet rim and potassic white mica cores); the retrograde evolution passed through a higher pressure stage (520°C and 8 kbar), that stabilized the Mg-rich potassic white mica and chlorite with decussate structure, and a late stage at T=500°C and P=2 kbar. The high temperature at medium pressures (~3-7 kbar) of the prograde path was followed by a pressure increase after garnet growth and by a fast tectonic exhumation responsible for the nearly isothermal decompression from 8 to 2 kbar. As a whole this metamorphic history may be related to Permian–Mid Triassic strike-slip zones. In this framework, an early transpression stage was followed by a later transtension predating the Late Triassic to Jurassic extension related to the opening of the Western Tethys (Piccardo *et al.*,1993; Marroni *et al.*, 1998).

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Chapter 4

DATING THE MYLONITIC STAGE

IN THE CERRETO VARISCAN BASEMENT:



Ar/Ar Laser Analysis OF Potassic White Mica In

ASYMMETRIC FOLIATION BOUDINS

CHAPTER 4

DATING THE MYLONITIC STAGE IN THE CERRETO VARISCAN BASEMENT: AR/AR LASER ANALYSIS OF POTASSIC WHITE MICA IN ASYMMETRIC FOLIATION BOUDINS

4.1 INTRODUCTION

Asymmetric foliation boudins (or foliation-fishes) are polycrystalline aggregates of mica with a sigmoidal shape (Platt & Vissers, 1980; Passchier & Trouw, 2005; Arslan *et al.*, 2008; Fossen, 2010). They occur in strongly foliated mylonitic rocks and are interpreted to develop during the late stages of the mylonitic event responsible for the development of the foliation (Fossen, 2010). The asymmetry creates a monoclinic microstructure, so they are potential kinematic indicators of the sense of shear (Hanmer, 1986; Passchier & Druguet, 2002; Goscombe & Passchier, 2003; Goscombe *et al.*, 2004).

Compared to other fabric elements defined by micas, e.g. porphyroclasts, mylonitic foliation, shear bands, which are useful to reconstruct the relative timing of deformative and metamorphic events, the asymmetric foliation boudins are less studied.

The importance of the correct interpretation of the microstructure is amplified when this microstructure consists of potassic white mica that can be the target of 40 Ar/ 39 Ar dating (Faure, 1986; McDougall & Harrison, 1999; Dickin, 1995).

Since mylonitic metapelites contain potassic white mica belonging to distinct fabric elements 40 Ar/ 39 Ar dating of potassic white mica is a suitable tool to constrain the age of different stages of the metamorphic evolution. In particular potassic white mica in mylonitic foliation, shear bands and asymmetric boudins is useful to attain information about the age and duration of the shearing stage (Mulch & Costa, 2004; Sanchez *et al.*, 2011).

We present a microstructural and geochronological study on asymmetric foliation boudins occurring in mylonitic micaschist of the Cerreto Pass. These rocks are part of the Variscan basement of the Northern Apennines (Italy). Up to now no data are **Figure of the previous page**: asymmetric foliation boudin in a Cerreto micaschist.

Collaboration: This study is in collaboration with Prof. H.-J. Massonne and Dr. T. Theye from the University of Stuttgart (guidance on microprobe analyses), Dr. A. Montanini and Dr. G. Molli respectively from the University of Parma and University of Pisa (discussion of data), and Dr. G. Di Vicenzo (Ar-Ar analyses).

available for the shearing stage in this crustal sector. ⁴⁰Ar/³⁹Ar dating of white mica in asymmetric foliation boudins and mylonitic foliation reveals a complex pre-Oligocene shearing history associated to the rifting stage that preludes the opening of the Western Tethys.

4.2 GEOLOGICAL FRAMEWORK

The Northern Apennines (Italy) are an Alpine orogenic belt formed during the Oligocene-Miocene collision between the Corsica-Sardinia and the Adria margins (Vai & Martini, 2001).

The Northern Apennine pile consists of three tectonic units (from top to bottom):

- Liguride Units: ophiolites from the Ligurian-Piedmontese ocean;

- Tuscan Units: Mesozoic-Tertiary sedimentary sequences belonging to the continental Adria crust;

- Tuscan Metamorphic Units: Apennine-age metamorphic rocks of the Adria microplate consisting of a Variscan basement and its upper Carboniferous-Tertiary cover.

The Variscan basement outcrops in the Cerreto Pass, Punta Bianca, Tuscan Metamorphic Ridge, Elba Island and it had been drilled also in the Larderello and Pontremoli underground (Fig. 4.1a, b). It shows a polymetamorphism from the Variscan orogenesis to the Alpine one. The Variscan history is best preserved in the Cerreto Pass (metamorphic peak at 550-660°C and 0.8-0.9 GPa; Molli *et al.*, 2002), whereas the Apenninic features can be observed in the Tuscan Metamorphic Units from Alpi Apuane to the Argentario Promontory.

The Variscan basement in the Cerreto Pass consists of medium-grade micaschist and amphibolite, associated with the Calcare cavernoso Formation, a detachment level of Triassic evaporites that allowed thrusting of the Tuscan Unit on the Tuscan Metamorphic Units during the Alpine orogenesis (Vai & Martini, 2001). Mylonitic deformation occurred under amphibolites- and greenschist-facies metamorphism and affected strongly the micaschist structure and composition. Amphibolite bodies, instead, preserve better the pre-shearing history (Molli *et al.*, 2002).

4.2.1 Previous geochronology

Only few radiometric age determinations are available for the Variscan basement of Northern Apennines.

Borsi *et al.* (1967) obtained 275 ± 12 Ma with the Rb/Sr on whole rock method for the Variscan basement outcropping in the Pisani Mountains ("Buti Quartzites and Phyllites Formation"; Schiaffino & Tongiorgi, 1962; Rau & Tongiorgi, 1966; Lo Pò & Braga, 2014). Del Moro *et al.* (1982) used the Rb-Sr and K-Ar methods on biotites, muscovites and horneblende from the Variscan basement in the Larderello subsurface. Their results give a bimodal age distribution of 285 ± 11 and 2.5-3.7 Ma: the former is related to the late Variscan thermal event, whereas the latter is related to the Pliocene magmatism.

K-Ar and ⁴⁰Ar-³⁹Ar analyses have been carried out on white mica separates from Paleozoic to Oligocene metasedimentary rocks present in the Alpi Apuane basement (Kligfield *et al.*,1986). In particular the ages obtained for the Paleozoic basement (80 to 24 Ma) were explained by partial resetting by Oligocene-Miocene metamorphism and/or excess argon.

Molli *et al.* (2002) performed Ar-Ar dating on two amphibolites samples from the Cerreto Pass: the Ar-Ar ages obtained on hornblende belonging to the main foliation (312 and 328 Ma) were related to the collisional stage of the Variscan orogenesis. Hence the Variscan basement in the Cerreto Pass might represent a remnant of the Southern Europe Variscan belt now included in the Apennine stacks.

Musumeci *et al.* (2011) presented geochronological data on detrital zircon and on metamorphic muscovite respectively from the Ortano Porphyroid and the Calamita Schist, two Palaeozoic metamorphic units outcropping in the eastern Elba Island. U-Pb dating in detrital zircon indicated a Middle-Ordovician magmatic origin (470 and 450 Ma) for the protolith of the porphyroids. Ar-Ar in metamorphic muscovite recorded the Late Miocene contact metamorphism (6 Ma) caused by the Porto Azzurro magmatic body (Musumeci *et al.*, 2011).

The mylonitic foliation in the micaschist and amphibolite from the Cerreto Pass, postdating the amphibolite foliation $(312 \pm 4 \text{ Ma} - 328 \pm 5 \text{ Ma})$, has never been dated before. In this contribution we present the first determination of the shearing stage age affecting the Variscan basement of Northern Apennines.



Figure 4.1 Geological framework of the study area: (a) localization of the Cerreto Pass; (b) distribution of the Variscan basement in Northern Apennines; (c) geological map with the Variscan basement outcrops in the Cerreto Pass.

4.3 FIELD DATA

The Variscan basement in the Cerreto Pass is made of mylonitic micaschist and amphibolite (Fig. 4.1c). The micaschist are dominant, whereas the amphibolite outcrop only in the uppermost sector. The contact between micaschist and amphibolite cannot be directly observed because it is covered by debris. Micaschist show in some cases a strong retrogression. Shear bands and mantled porphyroclast commoly indicate a dextral sense of shear. The mean field (mylonitic) foliation dips 180° with an increasing inclination towards east. The stretching lineation in the micaschist sequence has a dip of 200° and an inclination of 20°. Axial surface of anticlinal folds dips about 340° with an inclination of 60° in the easternmost outcrops. Mineralization of pyrite, epidote, K-feldspar, titanite lies on a plane dipping 290° with an inclination of 70°. The whole micaschist-amphibolite complex occurs as a tectonic slice enclosed in Triassic evaporites (Calcare Cavernoso Formation) with a topographic area of about 80,000 m².

4.4 PETROGRAPHY

Petrographic and microstructural analyses were carried out with a polarising optical microscope and a scanning electron microscope Philips 515B at the Dipartimento di Scienze Biologiche, Geologiche ed Ambientali (Università di Bologna). The samples are micaschist with a well-developed mylonitic foliation wrapping around deformed porphyroclasts. Locally, ultramylonitic levels consisting of fine-grained chlorite, epidote porphyroclasts, albite, quartz, titanite have been observed. These levels overprints the mylonitic foliation and have a different orientation. The samples were cut parallel to the mylonitic lineation and perpendicular to the mylonitic foliation in order to observe in thin sections the microstructures with a monoclinic symmetry. One thin section containing well-developed asymmetric foliation boudins (PC23) was studied in detail. PC23 is a medium-grained micaschist with a mylonitic foliation wrapping around white mica and biotite-fishes, garnet, plagioclase and quartz porphyroclasts. C'-type shear bands are also observed (Fig. 4.2). The mylonitic foliation consists of fine-grained white mica, chlorite and biotite and is usually folded (Fig. 4.3a). Ilmenite, zircon, apatite, rutile, titanite, epidote, tourmaline are the accessory minerals. White mica occurs in mica-fishes (Fig. 4.3b), fine-grained grains along the mylonitic foliation, that in some places form asymmetric foliation boudins (Fig. 4.3c). White mica in the mylonitic foliation is often associated with chlorite and biotite. Flattened garnet porphyroclasts (up to 5 mm in length) are extensively fractured. The fractures are

usually filled by chlorite (Fig. 4.3d) that is particularly abundant at the garnet rims. Ilmenite, quartz and zircon inclusions delineate a discontinuous internal schistosity discordant with respect to the external schistosity (mylonitic foliation). Plagioclase and quartz form flattened porphyroclasts with inclusions of white mica, biotite, rutile and apatite. Book-shelf sliding porphyroclasts of plagioclase and dynamically recrystallized quarz subgrains are also present (Fig. 4.3e). Ilmenite is both into garnet and in the matrix with fractured and bended crystals. Ilmenite crystals with rutile cores occur in the matrix; they host quartz, zircon and apatite inclusions. Titanite was found with irregular grains along the mylonitic foliation. Epidote has an irregular internal structure and contains white mica and chlorite inclusions. It occurs also as inclusions in mica-fishes.

Since the mylonitic foliation is the main schistosity in the selected sample, crystals have been distinguished in prekinematic (mica-fishes, bended potassic white mica, plagioclase, quartz and garnet porphyroclasts) and synkinematic (fine-grained white mica and biotite, chlorite, plagioclase, quartz, ilmenite and titanite along the mylonitc fabric). Crystals along the ultramylonitic levels postdates the mylonitic foliation development.



Figure 4.2 Scan and sketch of PC23 thin-section with the main mylonitic microstructures.



4.4.1 Asymmetric foliation boudins

In order to investigate the microstructure of the asymmetric foliation boudins, highmagnification images of these microstructures in PC23 were acquired through a Jeol JSM-6500F thermal Field emission scanning microscopy at INGV (Rome), using an acceleration voltage of 10-15 kV and primary electric current 10-15 nA. An EDS microanalyzer allowed to collect semiquantitative analyses of the mineral composition. Following the indications of Fossen (2010) on the descriptional terminology of boudins, width refers to the maximum axis of the boudins in the extension quadrants, and thickness consists of the measure of the axis in the shortening direction. Asymmetric foliation boudins in the selected sample are characterized by a width of 2-8 mm, a thickness of 0.5-4 mm and an aspect ratio of 2-6. Their area ranges from 1 to 17 mm². The majority of the asymmetric foliation boudins is 3 mm wide, 0.7 mm thick and 1 mm² extended. The average aspect ratio is 3.5. They have commonly lozenge shape, but also irregular shapes have been observed (Fig. 4.4a). They actually consist of aggregates of potassic white mica crystals with different orientations of the cleavage plains (Fig. 4.4b). The borders among different crystals can be recognized in high-resolution and high-magnification BSE images (Fig. 4.4c), while they cannot be easily distinguished in low-magnification BSE images and X-ray maps. Potassic white mica (bright grey in BSE images) is intimately intergrown with small crystals of sodic white mica (dark grey). The latter usually forms flakes with a width of 10-50 µm and a thickness of few µm (Fig. 4.4d). Chlorite form both intergrowths with potassic white mica,. Furthermore, the stretched rims of the boudins consist of chlorite (Fig. 4.4e). Asymmetric foliation boudins contains ilmenite and minor rutile, apatite, zircon, and epidote. Ilmenite delineates a sort of internal schistosity, with a curvature that increases progressive towards the rim of the boudin and connects continuously with the external schistosity (mylonitic foliation in our case). This is a common features of synkinematic microstructure (Zwart, 1962; Zwart & Calon, 1977, Vernon, 2004). C'-shear bands usually separate adjacent asymmetric foliation boudins.



Figure 4.4 Microstructure of an asymmetric foliation boudin: (a) low-magnification complessive view; (b) potassic white mica-sodic white mica intergrowths; (c) several potassic white mica crystals with different orientation of the cleavage plains; (d) 5 μ m-wide sodic white mica flake; (e) chlorite tail in the asymmetric boudin rim. The dotted lines indicate the cleavage plains, whereas the dashed lines correspond to the border between two different crystals. Abbreviations: Chl = chlorite, Ms = potassic white mica, Pg = sodic white mica.

4.5 MINERAL CHEMISTRY AND P-T ESTIMATES

Mineral chemistry analyses of potassic white mica, chlorite, biotite, garnet, plagioclase were collected through a CAMECA SX100 electron microprobe equipped with five wavelength dispersive spectrometers at the Institut für Mineralogie und Kristallchemie (Universität Stuttgart) with an acceleration voltage of 15 kV, a primary electric current of 10-15 nA and counting times for each element of 20 s at the peak and the background. The elements analysed in potassic white mica, chlorite, biotite, garnet, plagioclase are Na, Mg, Al, Si, K, Ca, Ti, Mn, Fe. Ba was analysed also in white mica and plagioclase. Synthetic and natural minerals and pure oxides were used as standards. The PaP correction procedure was applied. Fe, Mg, Ti, Na X-ray maps in mica-fish and potassic white mica along the mylonitic foliation were collected with a counting time of 100 ms, a step width of 2 μ m, an acceleration voltage of 15 kV and a current of 30 nA and were elaborated with XMap software (Bernhardt et al., 1995). Structural formulae were calculated with CalcMin program (Brandelik, 2009). Supplementary Fe, Mg, Ti, Na X-ray maps of potassic white mica in asymmetric foliation boudins were collected with a Jeol JSM-5400 at the University of Bologna using a counting time of 100 ms, a step width of 7 µm, an acceleration voltage of 20 kV and a current of 30 nA.

Potassic white mica shows some chemical variability (Tab. 4.1): mica-fishes have higher contents of Na (0.26-0.55 apfu) and lower contents of Mg (0.08-0.21 apfu) than potassic white mica along the mylonitic foliation (Na=0.04-0.46 apfu; Mg=0.16-0.44 apfu; Fig. 4.5a, b, c); asymmetric foliation boudins show variable composition with Si=6.08-6.38 apfu, Mg=0.10-0.35 apfu, Na=0.09-0.55 apfu (Fig. 4.5e). Na-rich areas can be observed (Fig. 4.5e). Inclusions of potassic white mica in biotite-fishes have Si=6.23-6.32 apfu, Mg=0.17-0.30 apfu, Na=0.32-0.44 apfu (Figs. 4.6a, b). Prekinematic biotite has Al^{IV} content ranging from 2.38 to 2.56 apfu and $XFe=Fe^{2+}/(Fe^{2+}+Mg)=0.47-0.63$. Mylonitic biotite shows roughly the same composition (Al^{IV}=2.48-2.62, XFe=0.55-0.64). Chlorite lying along the mylonitic foliation has Si=5.10-6.04 apfu and XFe=0.53-0.64; Si and XFe in chlorite deriving from the garnet retrocession are respectively 5.31-5.44 apfu and 0.57-0.60. The composition of plagioclase porphyroclasts is Ab₇₉₋₉₅An₅₋₂₁Or₀₋₂ in the core and Ab₇₉₋₈₉An₁₁₋₂₀Or₀₋₁ in the rim. Garnet shows a slight compositional zonation with an increase of almandine and pyrope and a decrease of grossular and spessartine during crystallization: the composition ranges from Alm₆₆Grs₂₄Sps₆Prp₅ in the inner core to Alm₇₃Grs₁₉Sps₁Prp₇ in the rim (Tab. 4.2; Fig. 4.6c).

Table 4.1 Microprobe analyses of potassic white mica calculated assuming 42- (Ca+ Ba) valencies and 4 H. Fe³⁺ is stoichiometric-estimated considering $Fe^{3+} = 42 - sum$ of the valencies (Si, Ti, Cr, Al, Mn, Mg, Fe)-Ba-Ca.

K-white mica													
	inclusions		r	mica-fishes			mylonitic foliation			asymmetric foliation boudins			
Point analyses	Ms03	Ms04	Ms05	Ms17	Ms18	Ms19	Ms28	Ms30	Ms32	Ms15_01	Ms15_02	Ms15_03	
SiO ₂	47.83	47.51	47.66	47.21	47.20	47.49	47.31	48.68	48.10	45.49	46.10	46.25	
TiO ₂	0.52	0.49	0.52	0.37	0.36	0.34	0.23	0.21	0.28	0.27	0.27	0.40	
Al_2O_3	34.58	34.24	33.42	35.72	36.27	36.25	34.69	33.56	33.87	36.38	35.12	35.31	
FeO	1.54	1.40	1.72	1.14	1.08	1.12	1.28	1.90	1.13	1.25	1.20	1.09	
Fe_2O_3	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	
MnO	n.d.	n.d.	n.d.	n.d.	0.01	0.04	n.d.	0.01	n.d.	n.d.	0.02	n.d.	
MgO	1.14	1.19	1.49	0.68	0.65	0.50	0.93	1.34	1.20	0.53	0.83	0.79	
CaO	0.00	0.03	0.03	0.05	0.06	0.04	0.03	0.08	0.02	0.11	0.01	0.06	
Na ₂ O	1.74	1.68	1.45	1.74	1.87	1.82	0.47	0.39	0.88	1.76	0.64	1.41	
K ₂ O	8.07	8.19	8.40	7.96	7.85	8.20	9.95	10.19	9.44	8.92	10.50	9.16	
BaO	0.22	0.21	0.25	0.26	0.24	0.25	0.20	0.07	0.13	0.26	0.20	0.25	
Total	95.63	94.94	94.93	95.13	95.59	96.06	95.07	96.42	95.04	94.97	94.87	94.72	
Si	6.256	6.266	6.301	6.192	6.159	6.182	6.259	6.366	6.344	6.064	6.168	6.162	
Al ^{IV}	1.744	1.734	1.699	1.808	1.841	1.818	1.741	1.634	1.656	1.936	1.832	1.838	
Al ^{VI}	3.586	3.587	3.508	3.715	3.737	3.744	3.668	3.538	3.608	3.779	3.706	3.708	
Ti	0.051	0.049	0.052	0.037	0.035	0.034	0.022	0.020	0.027	0.027	0.027	0.040	
Fe ²⁺	0.168	0.155	0.190	0.125	0.118	0.122	0.141	0.207	0.124	0.140	0.134	0.121	
Fe ³⁺	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	
Mn	n.d.	n.d.	n.d.	n.d.	0.001	0.005	n.d.	0.001	n.d.	n.d.	0.002	n.d.	
Mg	0.222	0.234	0.293	0.134	0.125	0.098	0.183	0.260	0.236	0.105	0.166	0.156	
Ca	0.000	0.004	0.004	0.006	0.009	0.005	0.004	0.012	0.003	0.015	0.001	0.009	
Ba	0.011	0.011	0.013	0.013	0.012	0.013	0.010	0.004	0.007	0.014	0.010	0.013	
Na	0.441	0.429	0.372	0.443	0.473	0.460	0.120	0.098	0.226	0.454	0.165	0.365	
Κ	1.347	1.379	1.416	1.331	1.306	1.362	1.679	1.701	1.587	1.517	1.792	1.557	

The Si content in mylonitic potassic white mica (average Si=3.11 apfu) can provide information about the pressure range (Massonne & Schreyer, 1987). The application of the geothermometer of Coggon & Holland (2002) and Caddick & Thompson (2008) for the selected Si content yields 3.0-6.6 kbar in the range 450-550°C.

In order to obtain a temperature estimate of the mylonitc event, the deformationdependent geometry of quartz is used here as temperature microgauges (Passchier & Trouw, 2005 and reference therein). Quartz forms large crystals with undulose extinction, frequently elongated, and aggregates of dynamically recrystallized small grains. A mantle of fine-grained quartz may surround quartz porphyroclasts with deformation lamellae. The deformational mechanism required for these microstructure is the subgrain rotation recrystallization (Stipp *et al.*, 2002). The observed deformation microstructures of quartz develop at 450-550°C (Law *et al.*, 2013; Law, 2014).

These P-T estimates are in agreement with the upper greenschist-facies mylonitic mineral assemblage.

Table 4.2 Chemical compositions of biotite, chlorite, plagioclase and garnet. The structural formula of biotite is calculated on the basis of 11 oxygens. The structural formula of chlorite is calculated on the basis of 28 oxygen and neglecting Na and Ca. Chlorite is assumed as completely idratated, with 16 H. The structural formula of plagioclase based on 8 oxygen. The garnet structural formula is of garnet normalized on 8 cations and Fe³⁺ is equal to 4 – Al.

	Biotite		Ch	Chlorite		Plagioclase		Garnet		
	prekinematic	mylonitic foliation	mylonitic foliation	garnet retrocession	core	rim	rim	outer core	inner core	
Point analyses	Bt40	Bt61	Chl21	Chl22	P102	P103	Grt01	Grt08	Grt18	
SiO ₂	36.74	36.63	24.34	24.73	64.67	63.37	36.71	36.83	36.72	
TiO ₂	1.57	0.36	0.02	0.04	0.02	0.01	0.07	0.09	0.12	
Al_2O_3	18.58	26.32	21.47	21.02	21.81	22.11	21.28	21.54	21.27	
Cr ₂ O ₃	n.d.	n.d.	n.d.	n.d.	n.d.	n.d.	0.03	0.03	0.03	
FeO ^{tot}			29.01	29.93						
Fe ₂ O ₃	n.c.	n.c.			0.05	0.09	0.01	n.d.	0.40	
FeO	18.02	18.00					33.18	31.69	29.97	
MnO	0.07	0.10	0.24	0.36			0.42	1.07	2.49	
MgO	10.50	6.78	12.98	11.98	n.d.	n.d.	1.71	1.33	1.19	
CaO	0.17	0.10			3.28	3.99	6.54	7.89	8.46	
Na ₂ O	0.17	0.15			10.10	9.52	0.01	n.d.	0.04	
K ₂ O	8.42	6.40			0.05	0.05				
BaO	0.09	0.03			0.00	0.00				
Total	94.35	94.86	88.06	88.04	99.98	99.14	99.96	100.48	100.68	
Si	2.783	2.688	5.223	5.332	2.853	2.825	5.847	5.822	5.785	
Ti	0.089	0.020	0.003	0.007	0.000	0.000	0.009	0.011	0.014	
Al	1.658	2.276	5.430	5.341	1.134	1.161	3.995	4.013	3.949	
Cr	n.c.	n.c.	n.c.	n.c.	n.c.	n.c.	0.003	0.003	0.004	
Fe ³⁺	n.c.	n.c.					0.002	0.000	0.047	
Fe ²⁺	1.142	1.105					4.420	4.190	3.949	
Fe ^{tot}			5.206	5.397						
Mn	0.005	0.006	0.044	0.065			0.057	0.144	0.332	
Mg	1.186	0.741	4.152	3.850			0.406	0.313	0.280	
Ca	0.014	0.008			0.155	0.191	1.115	1.336	1.428	
Na	0.026	0.021			0.864	0.823	0.003	0.001	0.011	
Κ	0.814	0.599			0.003	0.003				
Ba	0.003	0.001								



Figure 4.5 Mineral composition of white mica in different microstructural positions: (a) representative analyses of mica-fish (pink) and white mica along the mylonitic foliation (green); (b) Mg X-ray map of a mica-fish (centre) and mylonitic foliation (lower left angle). Note that the mica-fish has less Mg; (c) Na X-ray map of a mica-fish. Note that the mica-fish is richer in Na; (d) Mg X-ray map of an asymmetric foliation boudin; (e) Na X-ray map of an asymmetric foliation boudin.



Figure 4.6 Chemical diagram for potassic white mica and garnet: (a) Si *vs*. Al plot for potassic white mica; (b) Mg *vs*. Na plot for potassic white mica; (c) almandine *vs*. spessartine plot for garnet.

4.6 Ar-Ar DATING

⁴⁰Ar-³⁹Ar data on sample PC23 were performed at IGG-CNR Pisa through *in situ* and step-heating laserprobe analyses. Rock chips (chip A and chip B; diameter ~ 9 mm) were drilled from a polished \geq 300 µm-thick section and irradiated for 60 hours along with the fluence monitor Fish Canoyn sanidine (FCs) in the TRIGA reactor at the University of Pavia (Italy). Ages are relative to the Fish Canyon Tuff sanidine (28.03 Ma; Jourdan & Renne, 2007) and were calculated using constants recommended by Steiger & Jäger (1977). In situ laser ⁴⁰Ar-³⁹Ar analyses were carried using either a continuous wave diode-pumped Nd-YAG infrared (IR) laser (maximum power 20 W) connected to an external computer-controlled shutter and a pulsed Nd-YAG ultraviolet (UV) laser (frequency quadrupled and Q-switched). Multiple (3-4) shots of the IR laser, operating at ~13W and 20 ms, were focused to produce melt pits ~100 μ m in diameter. The UV laser, operated at 20 Hz and ~1 mJ per pulse, was focused to ~10 µm and repeatedly rastered by a computer-controlled x-y stage over areas of ~ 0.02 mm^2 . ^{40}Ar -³⁹Ar dating was focused on potassic white mica in asymmetric foliation boudins and potassic white mica aligned along the mylonitic foliation. During data acquisition, the sample was observed through a CCD camera coaxial with the laser beam. Incremental heating analysis was performed on a single mica fish (weight 5.1 mg) directly detached from one of the rock chips used for *in situ* analysis. The IR laser beam, homogenized by a beam homogenizer lens which produces a flat power distribution, was defocused to a 2 mm spot size and slowly rastered (at 0.1 mm s⁻¹) by the computer-controlled x-v stage to ensure even heating. Steps were carried out at increasing laser power to complete melting. After cleanup (10 and 15 min), extracted gases were equilibrated via automated valves into a MAP215-50 mass spectrometer equipped with a Balzers SEV217 secondary electron multiplier. Blanks were analyzed every three to four analyses. Data corrected for post-irradiation decay, mass discrimination effects, isotopes derived from interfering neutron reactions and blanks are listed in Tables 4.3 and 4.4. Errors are reported at 2σ and do not include the uncertainty in the J value, which was included in the calculated weighted mean ages or total gas ages. The location of the selected sector for step-heating analysis, the IR laver spots and UV laser pits with the correspondent age is shown in Fig. 4.9.

In chip A, Ar-Ar apparent ages range from 186.6 ± 4.5 Ma to 314.9 ± 8.6 Ma (Fig. 4.7a). The error-weighted mean on the data population of chip A is 189.5 ± 2.8 Ma with a MSWD = 1.38. In chip B, apparent ages ranges between 180.5-374.0 Ma (Fig. 4.7b),

clustering around 191 Ma (error-weighted mean = 191.1 ± 2.0 Ma with MSWD = 1.17). Figures 4.7c, d report the age distribution in asymmetric foliation boudins and along the mylonitic foliation. Step-heating Ar-Ar spectrum is highly discordant (Fig. 8), with a staircase profile especially in low cumulative % of released ³⁹Ar. The apparent ages get systematically old with increasing extraction temperatures. A plateau-like portion can be recognized in the range 178-205 Ma. The total gas is 183.41 ± 1.48 Ma.

Considering the data of both chips, the most representative age is 190.7 ± 1.3 Ma with MSWD = 1.3.

Table		aryses	$(a) \operatorname{cmp} \mathcal{A}, (b) \operatorname{cmp}$	<i>р</i> р .						
No.	description	laser	IR # shots / UV (mm ²)	Age (Ma)	$\pm 2\sigma$	⁴⁰ Ar* %	Ca/K	$\pm 2\sigma$	$^{38}\mathrm{Ar}_{\mathrm{Cl}}\!/^{39}\mathrm{Ar}_{\mathrm{K}}$	$\pm 2\sigma$
J=0.005	5872±0.000047									
Chip A										
92A	asymmetric foliation boudin	UV	200x100	189.1	5.9	92.3	0.11	0.10	0.0012	0.0030
92C	asymmetric foliation boudin	UV	200x100	187.3	4.5	91.5	0.05	0.12	0.0037	0.0026
92D	asymmetric foliation boudin	UV	200x100	226.3	3.1	97.4	0.02	0.05	0.0000	0.0021
92E	asymmetric foliation boudin	UV	200x100	188.6	6.1	99.0	0.03	0.13	0.0022	0.0041
92G	mylonitic foliation	UV	200x100	254	33	92.5	-	-	-	-
92H	asymmetric foliation boudin	UV	200x100	246.3	4.0	99.8	0.00	0.07	0.0001	0.0014
92I	mylonitic foliation	UV	200x100	261.6	3.9	97.8	0.02	0.06	0.0011	0.0021
92L	mylonitic foliation	UV	200x100	246.1	3.7	99.8	0.08	0.06	0.0016	0.0018
92P	mylonitic foliation	IR	3	314.9	8.6	84.3	0.06	0.24	0.0137	0.0034
92Q	mylonitic foliation	IR	3	231	42	89.0	0.10	1.53	0.0143	0.0188
92S	asymmetric foliation boudin	UV	200x100	229.6	2.8	98.7	0.03	0.10	0.0001	0.0012
92T	asymmetric foliation boudin	UV	200x100	192.8	4.0	96.9	0.06	0.13	0.0010	0.0020
92U	asymmetric foliation boudin	UV	200x100	207.9	2.6	97.7	0.04	0.11	0.0023	0.0017
92V	asymmetric foliation boudin	UV	200x100	186.6	4.5	98.1	0.10	0.15	0.0007	0.0019
92W	asymmetric foliation boudin	UV	200x100	192.9	5.7	98.0	-	-	0.0005	0.0014
92X	asymmetric foliation boudin	UV	triangle (b=200, h=400)	250.6	3.0	94.9	0.00	0.11	0.0030	0.0019
92Z	asymmetric foliation boudin	UV	200x100	204.8	3.3	97.9	0.04	0.10	0.0027	0.0017
94A	asymmetric foliation boudin	UV	200x100	209.4	4.4	99.6	0.04	0.15	0.0044	0.0019
No.	description	laser	IR # shots / UV (mm ²)	Age (Ma)	±2σ	⁴⁰ Ar* %	Ca/K	±2σ	$^{38}\mathrm{Ar}_{\mathrm{Cl}}/^{39}\mathrm{Ar}_{\mathrm{K}}$	±2σ
J=0.005	5872±0.000047									
Chip B										
91C	mylonitic foliation	IR	3	189.1	4.0	97.3	0.17	0.12	0.0074	0.0019
91D	mylonitic foliation	IR	3	191.6	1.8	97.3	0.09	0.07	0.0102	0.0012
91E	asymmetric foliation boudin	IR	3	192	30	93.7	0.37	0.79	0.0037	0.0114
91H	mylonitic foliation	IR	4	245.1	6.2	90.2	0.11	0.35	0.0059	0.0040
91I	mylonitic foliation	IR	3	202.4	2.0	94.7	0.17	0.07	0.0086	0.0013
91M	asymmetric foliation boudin	UV	100x100	198	12	96.3	0.07	0.37	0.0047	0.0052
91N	asymmetric foliation boudin	UV	200x100	231.2	3.7	95.5	0.03	0.14	-	-
91Q	asymmetric foliation boudin	UV	200x100	213.4	3.5	95.5	0.084	0.048	0.0017	0.0017
91R	asymmetric foliation boudin	UV	200x100	189.0	4.6	90.0	0.10	0.10	-	-

Table 4 3 40 Ar/ 39 Ar in situ analyses: (a) chin A: (b) chin B

No.	description	laser	IR # shots / UV (mm ²)	Age (Ma)	±2σ	⁴⁰ Ar* %	Ca/K	±2σ	$^{56}Ar_{Cl}/^{59}Ar_{K}$	±2σ
J=0.005	872±0.000047									
Chip B										
91C	mylonitic foliation	IR	3	189.1	4.0	97.3	0.17	0.12	0.0074	0.0019
91D	mylonitic foliation	IR	3	191.6	1.8	97.3	0.09	0.07	0.0102	0.0012
91E	asymmetric foliation boudin	IR	3	192	30	93.7	0.37	0.79	0.0037	0.0114
91H	mylonitic foliation	IR	4	245.1	6.2	90.2	0.11	0.35	0.0059	0.0040
91I	mylonitic foliation	IR	3	202.4	2.0	94.7	0.17	0.07	0.0086	0.0013
91M	asymmetric foliation boudin	UV	100x100	198	12	96.3	0.07	0.37	0.0047	0.0052
91N	asymmetric foliation boudin	UV	200x100	231.2	3.7	95.5	0.03	0.14	-	-
91Q	asymmetric foliation boudin	UV	200x100	213.4	3.5	95.5	0.084	0.048	0.0017	0.0017
91R	asymmetric foliation boudin	UV	200x100	189.0	4.6	90.0	0.10	0.10	-	-
91S	asymmetric foliation boudin	UV	triangle (b=200, h=200)	188.6	5.4	97.6	0.04	0.20	0.0022	0.0030
91V	mylonitic foliation	IR	3	180	11	97.1	0.04	0.42	0.0006	0.0047
91W	mylonitic foliation	IR	3	374	10	65.5	0.33	0.28	0.0815	0.0045
91X	asymmetric foliation boudin	UV	200x100	190.2	5.2	97.4	0.15	0.17	0.0055	0.0032
91Z	asymmetric foliation boudin	UV	200x100	208.7	2.9	92.9	0.11	0.13	0.0087	0.0023
93A	asymmetric foliation boudin	UV	200x100	193.8	3.4	96.7	0.03	0.11	0.0029	0.0013
93B	asymmetric foliation boudin	UV	triangle (b=140, h=300)	191.7	3.3	96.7	0.05	0.11	0.0052	0.0018

No.	laser power (W)	Age (Ma)	$\pm 2\sigma$	⁴⁰ Ar* %	³⁹ Ar _K %	Ca/K	$\pm 2\sigma$	$^{38}\mathrm{Ar}_{\mathrm{Cl}}\!/^{39}\mathrm{Ar}_{\mathrm{K}}$	$\pm 2\sigma$
100F	0.20	25.4	2.8	36.458	0.186	0.107	0.321	0.029	0.002
100G	0.40	51.66	0.99	74.402	0.874	0.099	0.071	0.032	0.001
100H	0.55	99.14	0.69	80.454	1.628	0.043	0.039	0.020	0.001
100I	0.70	130.13	0.64	91.547	2.599	0.041	0.024	0.013	0.001
100K	0.80	148.82	0.68	94.587	3.316	0.043	0.018	0.009	0.001
100L	0.88	160.48	0.76	96.145	3.692	0.032	0.017	0.007	0.001
100M	0.95	171.44	0.74	96.963	4.692	0.032	0.016	0.005	0.001
100O	1.02	178.90	0.79	97.839	4.397	0.027	0.016	0.003	0.001
100P	1.10	181.82	0.78	97.760	6.564	0.027	0.012	0.004	0.001
100Q	1.17	183.75	0.84	98.040	5.771	0.032	0.013	0.003	0.001
100R	1.25	186.47	0.85	98.324	5.962	0.031	0.013	0.003	0.001
100T	1.32	187.05	0.80	98.596	4.256	0.031	0.016	0.003	0.001
100U	1.40	187.93	0.87	98.498	5.946	0.036	0.015	0.003	0.001
100V	1.50	188.49	0.98	98.444	5.305	0.053	0.015	0.003	0.001
100W	1.60	189.62	0.91	98.402	4.734	0.068	0.017	0.004	0.001
100X	1.70	190.39	0.80	98.483	3.373	0.051	0.021	0.003	0.001
100Y	1.85	191.21	0.79	98.444	4.140	0.076	0.019	0.003	0.001
100Z	1.97	192.19	0.83	98.454	2.894	0.097	0.025	0.003	0.001
101C	2.10	193.22	0.86	98.704	2.149	0.058	0.033	0.003	0.001
101D	2.3	194.0	1.0	98.582	1.792	0.052	0.038	0.002	0.001
101E	2.6	194.55	0.83	98.721	4.600	0.051	0.018	0.003	0.001
101F	2.9	197.14	0.79	98.746	6.973	0.039	0.013	0.002	0.001
101G	3.2	200.46	0.86	98.837	3.387	0.028	0.022	0.002	0.001
101H	3.5	201.74	0.82	98.866	3.075	0.032	0.024	0.002	0.001
101I	3.9	204.59	0.94	98.847	2.363	0.055	0.031	0.002	0.001
101K	4.4	203.4	1.4	98.700	1.454	0.045	0.051	0.002	0.001
101L	5.0	204.5	1.0	98.707	0.858	0.056	0.079	0.001	0.001
101M	6.0	208.15	0.92	98.595	1.253	0.066	0.056	0.002	0.001
101P	7.5	212.2	1.1	98.964	0.886	0.049	0.071	0.002	0.001
101Q	9.0	220.6	1.5	98.816	0.456	0.082	0.149	0.002	0.001
101R	12	224.5	1.4	99.054	0.376	0.117	0.171	0.002	0.001
101T	15	253	11	98.169	0.049	1.020	1.412	0.003	0.005

I ADIC T.T $AI/AI SIGD-IICALITIZ ATIALVSUS.$	Table 4.4	40 Ar/ 39 Ar	step-heating	analyses.
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Figure 4.7 (next page) Ages from *in situ* Ar-Ar dating: (a) ages *vs.* cumulative number of analyses in chip A; (b) ages *vs.* cumulative number of analyses in chip B; (c) relative probability plot considering only the point analyses in asymmetric foliation boudins; (d) relative probability plot with the point analyses along the mylonitic foliation. The relative probability distribution was calculated by the Isoplot/Ex program, v. 3.00 (Ludwig, 2003) using the 2σ analytical uncertainty.





Figure 4.8 Step-heating spectrum in an asymmetric foliation boudin.

4.7 DISCUSSION

4.7.1 Dating asymmetric foliation boudins

In situ ⁴⁰Ar/³⁹Ar dating of potassic white mica is texturally-controlled, so it is possible to attribute each age to its microstructural position. A comparison of the age distribution in mylonitc foliation and in asymmetric foliation boudins is presented.

Potassic white mica in asymmetric foliation boudins is characterized by four main ages (principal peaks in the relative probability plot in Fig. 4.7c) around 190 Ma, 210 Ma, 230 Ma, 240 Ma. In the asymmetric foliation boudins of chip B younger ages (~190 Ma) are mostly located in the core of the structure, while the rims of the boudin contain older ages (213.4 ± 3.5 Ma, 231.2 ± 3.7 Ma, 208.7 ± 2.9 Ma). Potassic white mica in mylonitic foliation shows four main age peaks around 190 Ma, 210 Ma, 240 Ma, 260 Ma and two subsidiary peaks at 315 Ma, 370 Ma. In particular mylonitic foliation contains the oldest ages (374.0 ± 10.4 Ma, 314.9 ± 8.6 Ma). Thus potassic white mica in mylonitic foliation shows more scattering in age distribution, with probable prekinematic relics. The preservation of older ages occurs within the mylonitic foliation rather than in asymmetric foliation boudins, which, instead, consist in a significant concentration of white mica coeval with the shearing stage.



Figure 4.9 BSE images of chip A (a) and chip B (b) with the location of the step-heating area (dotted line), the IR laver spots (circles) and UV laser pits (squares, rectangles and triangles). The correspond ages are reported with the 2σ analytical uncertainty.

4.7.2 Interpretation of the ⁴⁰Ar-³⁹Ar ages

A complex age record results from the ⁴⁰Ar-³⁹Ar dating of the ductile shearing microstructures in the mylonitic micaschist of the Cerreto Pass. Asymmetric foliation boudins exhibit compositional and geochronological heterogeneity that is commonly related to local recrystallization of individual potassic white mica grains (Mulch & Costa, 2004). In our case study this is valid also for potassic white mica along the mylonitic foliation.

The staircase-like shape of the step-heating spectrum has been commonly associated to: (i) a partial diffusive loss of radiogenic ⁴⁰Ar from a single mica population during a thermal event (Kelley & Turner, 1991; Harrison et al., 2005); (ii) mixtures of different generation of micas (West & Lux, 1993); (iii) protracted crystallization (Kirschne et al., 1996). In our case study the step-heating spectrum shows a staircase shape, but with two different slopes: the young ages at low extraction temperatures and the old ages at high extraction temperatures are characterized by a pronounced slope; the intermediate ages (from 178.9 ± 0.8 Ma to 204.6 ± 0.9 Ma) at intermediate extraction temperatures resemble a pseudo-plateau shape with a slight staircase variation. These intermediate ages were obtained for the 78% of the ³⁹Ar released. Thus, we propose two different interpretations for the high-slope and low-slope sectors of the step-heating spectrum. The high-slope sectors may be due to the mixing of argon components for the coexistence of different generation of white mica. The low-slope sector resembles the staircase spectra of Kirschne et al. (1996), interpreted as the time span over which mica crystallized and grew. In the Cerreto metapelites, the estimated time of protracted crystallization during ductile deformation is 26 My.

Thus, even if highly-discordant step-heating spectra have occasionally been connected to a partial diffusive loss of radiogenic 40 Ar from a single mica population during a thermal event (Kelley *et al.*, 1991; Harrison *et al.*, 2005), the overall staircase shape of the Cerreto spectrum is presumably due to a mixing of different generations of white mica and to a protracted crystallization.

4.7.3 Tectonic implications

The imprints of different recrystallization stages are preserved in the Cerreto micaschist, allowing the identification of some peculiar geochronological signatures (Tab. 4.5). The oldest *in situ* ages (374 and 315 Ma) are close to the 312, 328, 397 Ma plateau age of the Variscan metamorphism in the amphibolite associated with the micaschist (Molli

et al., 2002), thus they may represent the age of pre-kinematic relics that survived to the shearing deformation. The intermediate ages 210 Ma, 230 Ma, 240 Ma, 260 Ma may be either interpreted as mixed ages or as the record of post-Variscan intracontinental strikeslip tectonics (Massari, 1986; Matte, 2001). Actually, the size of the UV/IR laser spot is bigger (hundreds of µm) than the size of the single muscovite grains (tens to hundreds of µm), so the laser spot can collect the Ar-Ar record of different muscovite generations. On the other hand, exhumation along detachment faults in Permian-Triassic times is also documented in the External Liguride Units representing the transition between the Internal Liguride oceanic domain and the thinned continental margin of the Adria plate (Marroni et al., 1998). In particular, an age of 228 ± 2 Ma (⁴⁰Ar/³⁹Ar on amphibole) has been ascribed to rift-related cooling along this peri-Tethyan distal continental margin. Furthermore, the sedimentary sequence outcropping in Punta Bianca includes a extensional basin deposit associated with alkaline basalts of Middle-Triassic erupted in a rifting setting (Ricci & Serri, 1975; Martini et al., 1986). Thus, the Permian-Triassic ages obtained for the studied sample may testify the involvement of the Cerreto mylonitc rocks in ductile shear zones that mark the passage from the extension consequent of the collapse of the Variscan chain to the onset of the Tethyan-related rifting.

The younger and most frequent age (190.7 Ma) predates the opening of the Western Tethys in late Jurassic, so it may represent another shearing event in a thinned crust or a thermal event that enhanced the partial loss of radiogenic ⁴⁰Ar. A first Tethyan-related rifting stage associated with lithosphere stretching has been documented in Western Alps in Lower Jurassic (Eberli & Froitzheim, 1990; Froitzheim & Manatschal, 1996). Occurrences of shearing in Lower Jurassic is documented in the Adriatic rifted margins (Mohn *et al.*, 2010): mafic rocks of a Permian crust sector records the onset of the rifting from 225-200 Ma and exhumation at seafloor between 170 and 140 Ma (Malenco Unit; Müntener *et al.*, 2000; Villa *et al.*, 2000); in the Southern Alps the shearing has been constrained at 182.0 \pm 1.6 Ma (Pogallo shear zone; Mulch *et al.*, 2002); extensional shearing in the necking zone of the Adria margin in the East-central Alps occurred at 180-205 Ma from middle crustal depth (Eita shear zone; Mohn *et al.*, 2012); Ar-Ar dating in the Adria distal margin of the Alpine Corsica yielded 187.8 \pm 1.4 Ma for the extensional shearing accommodating rift-related crustal thinning (Belli Piani shear zone; Beltrando *et al.*, 2013).

As a whole the geodynamic context of these Adria-derived basement remnants corroborate the thesis of a shearing event at 190 Ma rather than a thermal event.

Age peaks	Interpretations	References
374 - 315 Ma	Variscan metamorphism	Molli et al., 2002
210 - 260 Ma	post-Variscan intracontinental strike- slip tectonics or mixed ages	Ricci & Serri, 1975; Martini et al., 1986; Massari, 1986; Marroni et al., 1998; Matte, 2001
191 Ma	rift-related shearing	Eberli & Froitzheim, 1990; Froitzheim & Manatschal, 1996; Müntener et al., 2000; Villa et al., 2000; Mulch et al., 2002; Mohn et al., 2010; Mohn et al., 2012; Beltrando et al., 2013

Table 4.5 Age peaks in the Cerreto micaschist and respective interpretations and references.

4.7.4 Mica-fish vs. foliation-fish. A cautionary note

In the Cerreto mylonitic rocks what seems a mica-fish at low-magnification may be actually a potassic white mica polycrystalline aggregate with a sigmoidal shape. In this respect, this microstructure should not be considered as a mica-fish, but actually as a fish-shaped asymmetric foliation boudin (foliation fish).

Foliation-fish must not be confused with mica-fish. Mica-fish is a large single crystal of mica with a lozenge shape and monoclinic symmetry (Passchier & Trouw, 2005). Differently from foliation-fish, mica-fish is commonly a prekinematic porphyroclast, thus providing information on the earlier phase(s) with respect to the mylonitic stage.

The common assumption that large (apparent single) potassic white mica crystals in mylonite are prekinematic mica-fishes can led to wrong interpretations of potassic white mica-based geothermobarometry and geochronology.

The composition of potassic white mica is commonly used to obtain thermobaric constrains (e.g. compositional isopleths in phase diagrams). In particular the Si content in potassic white mica provides information on the pressure of metamorphism (Massonne & Schreyer; 1987). If the fish-structure involving potassic white mica is interpreted as mica-fish, the P-T values indicated by potassic white mica are related to the earlier stages of metamorphic evolution. Instead, if the fish-structure is interpreted

as foliation fish, the *P*-*T* constrains provided by potassic white mica regard the synkinematic mylonitc stage.

Radiometric ages of potassic white mica separates are in some case interpreted on the basis of the potassic white mica grain size (e.g. Kligfield *et al.*, 1986; West & Lux, 1993): big crystals of potassic white mica are considered prekinematic relics, while fine-grained potassic white mica is commonly associated with the mylonitic foliation development. This is not always true, because the coarse-grained separates might be foliation-fishes. Conversely, fine-grained minerals may belong to pre-kinematic relics. Thus dating mineral separates is not the recommended technique to obtain information about ages in mylonitic rocks.

4.8 CONCLUSIONS

- In our case study, asymmetric foliation boudin results in a preferential microstructure for obtaining geochronological information about the shearing event because: (i) it shows a more narrow age spectrum, usually related to the shearing stage(s), differently from the mylonitic foliation that may contain prekinematic relics; (ii) white mica in mylonitic foliation is commonly associated with other minerals (e.g. chlorite) that may disturb the analysis; (iii) thanks to its big size, asymmetric foliation boudin is easier to be dated than fine-grained mica along the mylonitic foliation.
- Pre-mylonitic relics preserve the age of the Variscan metamorphism at 374 and 315 Ma.
- 3. The main shearing event in the Cerreto micaschist occurred at 400-500°C and 3-5 kbar. The age (190.7 ± 1.3 Ma) testify the involvement of the Northern Apennine basement in the Tethyan-related rifting stage of the Adria margin, widely documented in the Adria-derived Alpine units.
- 4. Apparent big crystals of white mica in mylonitic rocks are not always prekinematic mica-fishes. They may consist of several fine-grained white-mica grains sinkinematically crystallized during the shearing event.
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Chapter 5

THE ALPINE METAMORPHISM AND THE INFLUENCE OF FERRIC IRON IN THE HEMATITE-RICH METASEDIMENTARY ROCKS FROM THE PISANI MOUNTAINS

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THE ALPINE METAMORPHISM AND THE INFLUENCE OF FERRIC IRON IN THE HEMATITE-RICH METASEDIMENTARY ROCKS FROM THE PISANI MOUNTAINS

5.1 INTRODUCTION

In recent years, the study of the metamorphic evolution of metasedimetary rocks has benefited from the introduction of new activity models of minerals containing important minor elements such as Fe^{3+} and Ti (White *et al.*, 2000, 2002, 2007; Diener *et al.*, 2007; Green et al., 2007; Smye et al., 2010). The development of these models has expanded the ability to extract quantitative *P*-*T* information from metasedimentary rocks modelled in increasingly complex systems, from KFMASHTO (White et al., 2000) to NCKFMASHTO (Diener & Powell, 2010; Boger et al., 2012; Korhonen et al., 2012) and MnNCKFMASHTO (López-Carmona et al., 2013). Although Fe³⁺ is known to have an important control on assemblages, the actual amount of Fe₂O₃ in metasedimentary rocks and minerals is still not able to be quantified accurately. Routine analytical techniques (XRF, EPMA) do not discriminate between FeO and Fe₂O₃. Even when Fe₂O₃ is directly measured in minerals and/or bulk rocks (e.g. by tritation, Mössbauer spectroscopy, synchrotron micro-XANES spectroscopy) its amount is likely to represent an overestimate due to low-temperature oxidation processes, including powder preparation (Diener & Powell, 2010; López-Carmona et al., 2013). Indeed, sample preparation procedures transform part of the Fe^{2+} to Fe^{3+} mainly by contamination, air oxidation and dissolution of the rock powder (Andrade et al., 2002). Milling equipment containing iron or steel may provide an addition of iron to the sample. Besides, the operation of grinding in air provides oxidation of ferrous iron, particularly when the powder is fine and local temperatures are high. Fitton & Gill (1970) demonstrated that only 4 minutes of fine grinding is sufficient to oxidize significantly the powder, with an absolute decrease of FeO by 1 wt. %.

Alternatively, bulk ferric iron can be estimated by using mineral modes and their stoichiometric-based Fe^{3+} calculation, but this method is difficult for some minerals (e.g. mica, amphibole). For example, Fe^{3+} estimation in biotite is uncertain without a complete chemical analysis including H, F and Cl (Braga *et al.*, 2006). Furthermore, other options in dealing with the bulk ferric iron are either to neglect its presence (Braga & Massonne, 2008) or allow a certain amount to stabilize Fe^{3+} -bearing phases in mineral assemblages (e.g., epidote: Braga *et al.*, 2007). Currently, a popular approach is

Figure of the previous page: hematite bands in a phyllite from the Pisani Mountains. This study was published by D. Lo Pò and R. Braga in Journal of Metamorphic Geology , 2014. to investigate phase relations calculating *P*- or *T*-*X* pseudosections, where *X* can be assigned to XFe^{3+} [$Fe^{3+}/(Fe^{3+} + Fe^{2+})$] (Diener & Powell, 2010) or to XFe_2O_3 (e.g.: Phillips *et al.*, 2010; López-Carmona *et al.*, 2013), and to look for correspondence between observations and modelling results.

So far, the effects of Fe₂O₃ have been evaluated in metapelites with mineral associations typical of very high-pressure (López-Carmona et al., 2013) and/or high-temperature conditions (Boger et al., 2012; Korhonen et al., 2012). In the simplified KFMASHTO system, for example, highly oxidized rocks develop garnet- and staurolite-bearing mineral assemblages at higher pressure when compared to the KFMASH system (White et al., 2000). Phase equilibria modelling of high-temperature and high-pressure metapelites in the extended MnNCKFMASHTO system highlights the influence that Fe₂O₃ exerts on phase equilibria (Korhonen et al., 2012; López-Carmona et al., 2013), and on the development of different mineral assemblages in high-grade metapelites with similar bulk-rock composition (Boger et al., 2012). With the exception of the study by White et al. (2000) on a theoretical model composition, "normal" greenschist-facies metasedimentary rocks of intermediate pressure with classical high-variance assemblage of chlorite-chloritoid-white mica have never been investigated. One reason might be the notion that chlorite, chloritoid and white mica do not accommodate high amounts of ferric iron, especially at upper greenschist-facies conditions (Hillier & Velde, 1991; Vidal et al., 1999, 2005).

In this paper we present the results of a thermodynamic modelling involving different assumptions on the bulk Fe₂O₃ content, which allowed us to explore the influence of Fe₂O₃ on the development of greenschist-facies assemblages. We show that oxidized hematite-rich metasedimentary rocks from the Variscan basement of the Northern Apennines occurring in the Pisani Mountains (Italy; Fig. 5.1) are better modelled when considering an effective ferric iron content in whole rock that is significantly different from Fe₂O₃ values obtained by titration. The Variscan basement of the Northern Apennines preserves relics of a Carboniferous polyphase metamorphism (Molli *et al.*, 2002; Franceschelli *et al.*, 2004). Its involvement in the Oligocene-Miocene Alpine collision, which is well recorded by HP/LT assemblages in the metasedimentary rocks now overlying the Variscan basement, is not known in detail. The new *P*-*T* data of this paper suggests that both the Variscan basement and the overlying cover behaved as a coherent tectonic unit during the Oligocene-Miocene collision.

5.2 GEOLOGICAL FRAMEWORK

The Northern Apennines are a thrust and fold belt formed during continental collision between the Corsica-Sardinia (Europe) and Adria (Africa) microplates during the last 30 Ma (Vai & Martini, 2001). The tectonic units are (from top to bottom):

- Liguride Units: ophiolites derived from the Jurassic Ligurian-Piedmontese ocean;
- Tuscan Units: very low-grade metamorphic to non-metamorphic late Triassic-Tertiary sedimentary sequences derived from the Adria crust;
- Tuscan Metamorphic Units: metamorphic sequence belonging to the Adria microplate, consisting of a Variscan pre-Carboniferous basement and Triassic-Oligocene metasedimentary rocks.

In the Northern Apennines the Variscan basement occurs in small and scattered outcrops, which preserve metamorphic and structural Variscan relics and have a strong metamorphic and structural Alpine overprint. The Alpine metamorphism is especially well documented in the Verrucano Formation, a Triassic metasedimentary sequence outcropping from the Alpi Apuane to the Argentario Promontory (the so-called Tuscan metamorphic ridge) and in the Tuscan Archipelago. The Alpine metamorphism is characterized by a first compressional HP-LT phase followed by crustal extension (Franceschelli et al., 1986; Kligfield et al., 1986; Theye et al., 1997; Giorgetti et al., 1998; Jolivet et al., 1998; Franceschelli & Memmi, 1999; Franceschelli et al., 2004). In the first phase (27-20 Ma) SW-NE compressional tectonics produced the main deformation phase (D₁) during wedge thickening and burial of oceanic and continental crust to about 40-55 km depths, with attainment of *P*-*T* conditions typical of blueschistto eclogitic-facies metamorphism. The metamorphic peak (6-8 kbar and 420-560 °C) occurred after D_1 and before the second deformational phase (D₂). The D_2 phase (14-12) Ma) is extensional and produced a rapid exhumation triggered by gravitational collapse. P-T constraints are 300-370 °C and 3-5 kbar.

5.2.1 The Pisani Mountains

In the Pisani Mountains (Fig. 5.1b), the Variscan basement of the Tuscan Metamorphic Units is known as the Buti Quartzites and Phyllites Formation (Schiaffino & Tongiorgi, 1962; Rau & Tongiorgi, 1966; 1974) and it is found in the cores of Alpine anticlines (Conti *et al.* 1991). It consists of albite-rich quartzites and chlorite-muscovite phyllites. The latter exhibits a banded structure due to the presence of hematite-rich layers. Two

different foliations are recognized: S_1 is the main schistosity delineated by phyllosilicates, partially obliterated by S_2 , which is a crenulation schistosity parallel to the axial plane of Alpine anticlines (Rau & Tongiorgi, 1974).

A metasedimentary cover, comprising the metamorphic Triassic silicoclastic succession (Verrucano Formation; Fig. 5.1b) overlies the Variscan basement of the Pisani Mountains. Just a few P-T estimates are available from the Verrucano Formation in this area (Franceschelli et al., 1986; Carosi et al., 1993; Montomoli, 2002), while no P-T constraints are available for the Variscan basement. Franceschelli et al. (1986) proposed two different P-T scenarios for the Verrucano Formation, ~ 400 °C and ~ 4-5 kbar or ~ 450 °C and ~ 9-10 kbar. The first one was determined using the phengite geobarometer calibrated by Velde (1965, 1967); the second one, instead, employed the calibration by Massonne (1981). Carosi et al. (1993) applied the calcite-dolomite geothermometer to marbles belonging to the metasedimentary cover of the Variscan basement, and obtained maximum temperatures of 380±30 °C. They attributed this temperature to the first Alpine metamorphic-deformational phase. In a fluid inclusion study of Alpine veins in the Verrucano formation, Montomoli (2002) obtained P-T conditions of 410 °C and 5 kbar. Because of reequilibration textures in the fluid inclusions, this constraint has to be considered as a retrograde segment of the P-T path, rather than the peak conditions (Montomoli, 2002). Thus, during the first Alpine deformational phase the Verrucano Formation probably went to higher pressures than 5-6 kbar.



Figure 5.1 Regional geological framework of the study area. (a) Location of the Pisani Mountains in the Northern Apennines; (b) geological map of the Pisani Mountains with the sampling sites (modified from Leoni *et al.*, 2009).

5.3 PETROGRAPHY

The selected hematite-rich rocks, which are representative of the Buti Quartzites and Phyllites Formation, were sampled in the Pisani Mountains from small outcrops near the contact with the Verrucano Formation (Fig. 5.1b). Mineral abbreviations are after Siivola & Schmid (2007).

Five thin sections (MP1, MP2, MP3, MP4, and MP4b) of hematite-rich metasedimentary rocks have been studied in detail. The samples are fine-grained phyllites exhibiting a metamorphic banded structure in hand-specimen, constituted of alternating mm-wide hematite rich- and hematite poor-layers (Fig. 5.2a). In particular MP1 is a fine-grained phyllite with two schistosities, delineated by white mica and hematite. The mineral assemblage consists of white mica, chlorite, quartz and hematite. Rutile, tourmaline, zircon are accessory phases. MP2 and MP3 have the same mineral assemblage as MP1 and only show the first schistosity. MP4b and MP4 contain

chloritoid; mineral modes (vol.%) obtained by point-counting techniques are muscovite (37), quartz (37), chlorite (15), hematite (7) and chloritoid (3), while no biotite and feldspar have been found. Rutile, tourmaline, monazite, apatite, zircon and florencite occur as accessory phases.

A polyphase deformation evolution has been reconstructed for the chloritoid-bearing sample. A first deformation phase (D₁) caused the development of a spaced schistosity (S₁), the main foliation, along which muscovite, chlorite, quartz and hematite crystallized. A second deformation phase (D₂), less pervasive than D₁, caused the folding of S₁ and the formation of a muscovite-chlorite -hematite crenulation schistosity (S₂; Fig. 5.2b). Previous studies interpreted the S₂ as an Alpine structure (Rau & Tongiorgi, 1974; Borsi *et al.*, 1967). A post-kinematic event was responsible for the blastesis of randomly oriented chlorite, hematite, muscovite flakes and chloritoid porphyroblasts occurring as single crystals (up to 500 µm) and rosettes (Fig. 5.2c, d).

Chloritoid porphyroblasts are more abundant in white mica- and chlorite-rich layers rather than in quartz-rich layers. They are poikilitic and contain chlorite, rutile, hematite, quartz, zircon, and florencite inclusions. These inclusions define a helicitic structure (Fig. 5.2e), delineating an internal schistosity continuous and concordant with the external one. Hematite can be considered a major mineral due to its modal quantity (7 vol.%). It occurs as fine-grained aggregates in bands parallel to the main schistosity (Fig. 5.2f) and as single crystals. The sample MP4 was chosen for thermodynamic modelling, because of its chloritoid-bearing assemblage that is useful to retrieve the temperature using the chlorite-chloritoid geothermometer (see Geothermobarometry section).



Figure 5.2 Main petrographic and microstructural features of the phyllites of the Buti Formation. (a) Rock slab; (b) crossed polars photomicrograph of the main schistosity (S_1) and the crenulation schistosity (S_2); (c) BSE image showing the chloritoid rosette with zircon and rutile inclusions; (d) enlargment of the previous BSE image showing the location of the analyses of a chlorite-chloritoid pair characterized by equilibrium features; (e) chloritoid porphyroblasts with helicitic structure (photomicrograph with plane polarized light); (f) BSE image showing the hematite bands parallel to S_2 and S_1 and the hematite blasts (Hem₃).

5.4 MINERAL AND WHOLE-ROCK CHEMISTRY

Chemical analyses of potassic white-mica, chlorite and chloritoid in MP4 were obtained with a WDS-CAMECA SX100 electron microprobe (EMP) at the Institut für Mineralogie und Kristallchemie (Universität Stuttgart). Analysis of Na, Mg, Al, Si, K, Ca, Ti, Mn, Fe in white-mica, chlorite and chloritoid was undertaken using an acceleration voltage of 15 kV and a current of 10 nA. For white mica analyses Ba was also measured. Counting times per element were 20 s at the peak and the background. Synthetic and natural minerals and pure oxides were used as standards. The beam diameter was defocused to 5 μ m. The PaP correction method was applied. Point analyses were chosen taking into account the microstructural sites. Mineral structural formulae, including Fe³⁺ calculated through stoichiometric constraints, are reported in Tables 5.1, 5.2 and 5.3.

Potassic white-mica is characterised by Si contents ranging from 6.01 to 6.38 apfu (formula based on 22 oxygens; Table 5.1), with Fe^{2+} and Mg are between 0.14-0.36 apfu and 0.05-0.33 apfu. The calculated average Fe^{3+} is 0.03 apfu. Figure 3a, b shows the variation of Si *vs*. Fe^{2+} and Mg in muscovite from different microstructural sites.

Chlorite is invariably a ripidolite characterized by Si content ranging from 2.50 to 2.73 apfu (Fig. 5.3c). The XFe^{2+} ratio ($Fe^{2+}/(Fe^{2+}+Mg)$) shows limited variations ($XFe^{2+} = 0.52 - 0.57$). On the basis of stoichiometry constraints, no Fe^{3+} was calculated (Table 5.2).

Chloritoid shows only slight chemical variations (Table 5.3). XFe^{2+} is 0.85-0.88, Fe^{3+} and XFe^{3+} are low and 0.04-0.13 apfu and 0.04-0.12 respectively. Mn is 0.01-0.02 apfu.





Table 5.1 Microprobe analyses of potassic white mica. *Footnote:* structural formulae calculated with CalcMin program (Brandelik, 2008) on the basis of 22 O, assuming 42-(Ca+Ba) valencies and 4 H. Fe³⁺ has been stoichiometric-estimated considering $Fe^{3+} = 42$ -sum of the valencies (Si, Ti, Cr, Al, Mn, Mg, Fe)-Ba- Ca.

Generation	Kwm_1						Kwm	1 2				
Analyses	Ms14	Ms39	Ms41	Ms57	Ms58	Ms64	Ms	31 M	ls33 N	Is34	Ms44	Ms45
SiO ₂	45.63	47.36	45.90	46.31	46.49	44.66	45.	.61 45	5.15 4	5.69	44.97	44.18
TiO_2	0.05	0.46	0.55	1.81	0.03	0.08	0.	.08 ().11	0.07	0.15	0.15
Al ₂ O ₃	35.18	35.32	34.38	35.10	35.69	34.12	35.	.02 34	4.14 3	5.42	34.00	34.47
FeO	2.02	1.63	2.52	2.52	2.08	1.81	3.	.08 2	2.64	2.68	2.78	2.72
Fe ₂ O ₃	0.00	0.00	0.00	0.00	0.00	3.47	0.	.00 0).61	0.00	0.93	0.94
MnO	0.01	0.00	0.02	0.00	0.00	0.04	0.	.04 ().08	0.05	0.02	0.05
MgO	0.41	1.11	0.91	0.43	0.40	0.59	0.	.36 ().69	0.34	0.64	0.30
CaO	0.09	0.07	0.04	0.06	0.05	0.06	0.	.04 ().04	0.02	0.13	0.04
Na ₂ O	1.11	1.41	1.29	1.01	1.62	1.01	1.	.04 ().86	0.87	1.08	1.16
K ₂ O	8.54	8.53	8.41	8.95	8.79	8.63	9.	.43 9	9.46	9.70	9.04	9.24
BaO	0.08	0.40	0.36	0.19	0.12	0.09	0.	.17 ().16	0.15	0.13	0.15
Total	93.12	96.27	94.38	96.38	95.26	94.55	94	85 93	3.94 9	4.98	93.85	93.39
Si	6.15	6.18	6.14	6.07	6.16	6.01	6.	.13 6	5.12	6.12	6.11	6.05
Al ^{IV}	1.85	1.82	1.86	1.93	1 84	1 99	1	87 1	1.88	1.88	1.89	1.95
Al ^{VI}	3 74	3.62	3 57	3 50	3 74	3 41	3	67	3 58	3 70	3 54	3.61
Ti	0.01	0.04	0.06	0.18	0.00	0.01	0	01 (0.01	0.01	0.02	0.01
Fe ²⁺	0.01	0.18	0.28	0.10	0.00	0.01	0.	35 () 30	0.30	0.32	0.01
Fe ³⁺	0.23	0.10	0.20	0.20	0.25	0.20	0.) 06	0.00	0.02	0.51
Mn	0.00	0.00	0.00	0.00	0.00	0.00	0.	.00 (1.00	0.00	0.09	0.10
Ma	0.00	0.00	0.00	0.00	0.00	0.00	0.	.00 (1.01	0.01	0.00	0.01
Ca	0.08	0.22	0.13	0.08	0.08	0.12	0.	01 ().14	0.07	0.15	0.00
Ca	0.01	0.01	0.01	0.01	0.01	0.01	0.	.01 (0.01	0.00	0.02	0.01
Da N-	0.00	0.02	0.02	0.01	0.01	0.00	0.	.01 ().01	0.01	0.01	0.01
Na	0.29	0.30	0.55	0.20	0.41	0.20	0.	.27 ().25	1.66	0.28	0.51
К	1.47	1.42	1.44	1.50	1.49	1.48	1.	.02 1	1.04	1.00	1.57	1.01
K-white mica												
Generation	Kw	m ₃										
Point analyses	Ms	03 Ms	04 Ms	s07 Ms	s09 Ma	s20 Ms21	Ms25	Ms28	Ms29	Ms9	2 Ms	s93
SiO_2												
T'O	46.	22 46.	12 48	.74 47	.46 47	.37 47.47	47.46	48.12	47.32	46.6	5 46	.43
110_{2}	46. 0.	22 46. 74 0.	12 48 06 0	.74 47 .03 0	.46 47 .05 0	.37 47.47 .04 0.28	47.46 0.13	48.12 0.06	47.32 0.24	46.6	5 46 5 0	.43 .06
Al_2O_3	46. 0. 34.	22 46. 74 0. 65 34.	12 48 06 0 71 36	.74 47 .03 0 .98 36	.46 47 .05 0 .25 35	.37 47.47 .04 0.28 .43 31.56	47.46 0.13 34.01	48.12 0.06 36.52	47.32 0.24 33.04	46.6 0.0 35.5	5 46 5 0 2 35	.43 .06 .12
Al_2O_3 FeO	46. 0. 34. 1.	22 46. 74 0. 65 34. 28 2.	12 48 06 0 71 36 00 1	.74 47 .03 0 .98 36 .96 1	.46 47 .05 0 .25 35 .86 2	.3747.47.040.28.4331.56.333.24	47.46 0.13 34.01 2.34	48.12 0.06 36.52 1.92	47.32 0.24 33.04 2.34	46.6 0.0 35.5 1.8	5 46 5 0 2 35 6 1	.43 .06 .12 .73
HO_2 Al_2O_3 FeO Fe_2O_3	46. 0. 34. 1. 0.	22 46. 74 0. 65 34. 28 2. 00 0.	12 48 06 0 71 36 00 1 00 0	.74 47 .03 0 .98 36 .96 1 .00 0	.46 47 .05 0 .25 35 .86 2 .00 0	.3747.47.040.28.4331.56.333.24.000.00	47.46 0.13 34.01 2.34 0.00	48.12 0.06 36.52 1.92 0.00	47.32 0.24 33.04 2.34 0.00	46.6 0.0 35.5 1.8 0.0	5 46 5 0 2 35 6 1 0 0	.43 .06 .12 .73 .00
FeO Fe_2O_3 FeO Fe_2O_3 FeO	46. 0. 34. 1. 0. 0.	22 46. 74 0. 65 34. 28 2. 00 0. 00 0.	12 48 06 0 71 36 00 1 00 0 02 0	.74 47 .03 0 .98 36 .96 1 .00 0 .03 0	.46 47 .05 0 .25 35 .86 2 .00 0 .02 0	.37 47.47 .04 0.28 .43 31.56 .33 3.24 .00 0.00 .02 0.00	47.46 0.13 34.01 2.34 0.00 0.00	48.12 0.06 36.52 1.92 0.00 0.00	47.32 0.24 33.04 2.34 0.00 0.00	46.6 0.0 35.5 1.8 0.0 0.0	5 46 5 0 2 35 6 1 0 0 0 0	.43 .06 .12 .73 .00
Al ₂ O ₃ FeO Fe ₂ O ₃ MnO MgO	46. 0. 34. 1. 0. 0.	22 46. 74 0. 65 34. 28 2. 00 0. 00 0. 66 0.	12 48 06 0 71 36 00 1 00 0 02 0 39 0	.74 47 .03 0 .98 36 .96 1 .00 0 .03 0 .27 0	.46 47 .05 0 .25 35 .86 2 .00 0 .02 0 .40 0	.37 47.47 .04 0.28 .43 31.56 .33 3.24 .00 0.00 .02 0.00 .50 1.65	47.46 0.13 34.01 2.34 0.00 0.00 0.84	48.12 0.06 36.52 1.92 0.00 0.00 0.28	47.32 0.24 33.04 2.34 0.00 0.00 1.05	46.6 0.0 35.5 1.8 0.0 0.0 0.0	5 46 5 0 2 35 6 1 0 0 0 0 0 0 0 0 0 0	.43 .06 .12 .73 .00 .00 .46
Al ₂ O ₃ FeO Fe ₂ O ₃ MnO MgO CaO	46. 0. 34. 1. 0. 0. 0. 0.	22 46. 74 0. 65 34. 28 2. 00 0. 00 0. 66 0. 00 0.	12 48 06 0 71 36 00 1 00 0 02 0 39 0 08 0	.74 47 .03 0 .98 36 .96 1 .00 0 .03 0 .03 0 .03 0 .03 0 .03 0 .07 0	.46 47 .05 0 .25 35 .86 2 .00 0 .02 0 .40 0 .08 0	.3747.47.040.28.4331.56.333.24.000.00.020.00.501.65.080.01	47.46 0.13 34.01 2.34 0.00 0.00 0.84 0.04	48.12 0.06 36.52 1.92 0.00 0.00 0.28 0.04	47.32 0.24 33.04 2.34 0.00 0.00 1.05 0.06	46.6. 0.0. 35.5 1.8 0.0 0.0 0.0 0.4 0.4	5 46 5 0 2 35 6 1 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	.43 .06 .12 .73 .00 .00 .46 .02
Al ₂ O ₃ FeO Fe ₂ O ₃ MnO MgO CaO Na ₂ O	46. 0. 34. 1. 0. 0. 0. 0. 0. 0.	22 46. 74 0. 65 34. 28 2. 00 0. 00 0. 66 0. 00 0. 66 0. 00 0. 66 0. 00 0. 60 1.	12 48 06 0 71 36 00 1 00 0 02 0 39 0 08 0 05 0	.74 47 .03 0 .98 36 .96 1 .00 0 .03 0 .03 0 .03 0 .03 0 .03 0 .03 0 .07 0 .90 1	.46 47 .05 0 .25 35 .86 2 .00 0 .02 0 .40 0 .08 0 .87 0	.37 47.47 .04 0.28 .43 31.56 .33 3.24 .00 0.00 .02 0.00 .50 1.65 .08 0.01 .71 0.38	47.46 0.13 34.01 2.34 0.00 0.00 0.84 0.04 0.96	48.12 0.06 36.52 1.92 0.00 0.00 0.28 0.04 1.68	47.32 0.24 33.04 2.34 0.00 0.00 1.05 0.06 0.48	46.6. 0.0. 35.5. 1.8 0.00 0.00 0.4 0.00 0.4 0.00 0.8	5 46 5 0 2 35 6 1 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 9 0	.43 .06 .12 .73 .00 .00 .46 .02 .93
HO_2 Al_2O_3 FeO Fe_2O_3 MnO MgO CaO Na_2O K_2O	46. 0. 34. 1. 0. 0. 0. 0. 0. 0. 9.	22 46. 74 0. 65 34. 28 2. 00 0. 00 0. 66 0. 00 0. 66 0. 00 1. 91 8.	12 48 06 0 71 36 00 1 00 0 02 0 39 0 08 0 05 0 84 8	.74 47 .03 0 .98 36 .96 1 .00 0 .03 0 .03 0 .03 0 .07 0 .90 1 .97 6	.46 47 .05 0 .25 35 .86 2 .00 0 .02 0 .02 0 .03 0 .04 0 .05 0 .08 0 .87 0 .95 8	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	47.46 0.13 34.01 2.34 0.00 0.00 0.84 0.04 0.96 9.02	48.12 0.06 36.52 1.92 0.00 0.00 0.28 0.04 1.68 7.75	47.32 0.24 33.04 2.34 0.00 0.00 1.05 0.06 0.48 9.88	46.6 0.0 35.5 1.8 0.0 0.0 0.0 0.4 0.4 0.0 0.8 9.0	5 46 5 0 2 35 6 1 0 0	.43 .06 .12 .73 .00 .00 .46 .02 .93 .89
HO ₂ Al ₂ O ₃ FeO Fe ₂ O ₃ MnO MgO CaO Na ₂ O K ₂ O BaO	46. 0. 34. 1. 0. 0. 0. 0. 0. 0. 9. 0.	22 46. 74 0. 65 34. 28 2. 00 0. 00 0. 66 0. 00 0. 66 1. 91 8. 28 0.	12 48 06 0 71 36 00 1 00 0 02 0 39 0 005 0 84 8 12 0	.74 47 .03 0 .98 36 .96 1 .00 0 .03 0 .03 0 .27 0 .07 0 .90 1 .97 6 .12 0	.46 47 .05 0 .25 35 .86 2 .00 0 .02 0 .040 0 .08 0 .87 0 .95 8 .07 0	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	47.46 0.13 34.01 2.34 0.00 0.00 0.84 0.04 0.96 9.02 0.16	48.12 0.06 36.52 1.92 0.00 0.00 0.28 0.04 1.68 7.75 0.11	47.32 0.24 33.04 2.34 0.00 0.00 1.05 0.06 0.48 9.88 0.16	46.6. 0.0. 35.5. 1.8 0.00 0.00 0.4 0.0. 0.8 9.0 0.2	5 46 5 0 2 35 6 1 0 0 00 0 05 0 99 0 33 8 1 0	.43 .06 .12 .73 .00 .00 .46 .02 .93 .89 .13
HO_2 Al_2O_3 FeO Fe_2O_3 MnO MgO CaO Na_2O K_2O BaO Total	46. 0. 34. 1. 0. 0. 0. 0. 0. 9. 0. 94.	22 46. 74 0. 65 34. 28 2. 00 0. 00 0. 66 0. 00 0. 66 0. 90 8. 28 0. 34 93.	12 48 06 0 71 36 00 1 00 0 02 0 39 0 05 0 84 8 12 0 38 98	.74 47 .03 0 .98 36 .96 1 .00 0 .03 0 .27 0 .07 0 .99 1 .997 6 .12 0 .06 95	.46 47 .05 0 .25 35 .86 2 .00 0 .02 0 .040 0 .08 0 .87 0 .95 8 .07 0 .00 95	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	47.46 0.13 34.01 2.34 0.00 0.00 0.84 0.04 0.96 9.02 0.16 94.96	48.12 0.06 36.52 1.92 0.00 0.28 0.04 1.68 7.75 0.11 96.47	47.32 0.24 33.04 2.34 0.00 0.00 1.05 0.06 0.48 9.88 0.16 94.57	46.6. 0.0. 35.5: 1.8 0.00 0.00 0.44 0.00 0.44 0.00 0.88 9.00 0.2 94.6	5 46 5 0 2 35 6 1 0 0 0 0 0 0 0 0 0 0 0 0 3 8 1 0 6 93	.43 .06 .12 .73 .00 .00 .46 .02 .93 .89 .13 .79
HO_2 AI_2O_3 FeO Fe_2O_3 MnO MgO CaO Na_2O K_2O BaO $Total$	46. 0. 34. 1. 0. 0. 0. 0. 0. 9. 0. 94.	22 46. 74 0. 65 34. 28 2. 00 0. 00 0. 66 0. 00 0. 66 0. 90 0. 91 8. 28 0. 34 93.	12 48 06 0 71 36 00 1 00 0 02 0 39 0 005 0 84 8 12 0 38 98	.74 47 .03 0 .98 36 .96 1 .00 0 .03 0 .27 0 .07 0 .990 1 .997 6 .12 0 .06 95	.46 47 .05 0 .25 35 .86 2 .00 0 .02 0 .40 0 .40 0 .87 0 .95 8 .07 0 .00 95	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	$\begin{array}{c} 47.46\\ 0.13\\ 34.01\\ 2.34\\ 0.00\\ 0.00\\ 0.84\\ 0.04\\ 0.96\\ 9.02\\ 0.16\\ 94.96\end{array}$	48.12 0.06 36.52 1.92 0.00 0.28 0.04 1.68 7.75 0.11 96.47	47.32 0.24 33.04 2.34 0.00 1.05 0.06 0.48 9.88 0.16 94.57	46.6 0.0 35.5 1.8 0.0 0.0 0.0 0.4 0.0 0.8 9.0 0.2 94.6	5 46 5 0 2 35 6 1 0 0 0 0 0 0 5 0 5 0 9 0 3 8 1 0 6 93	.43 .06 .12 .73 .00 .00 .46 .02 .93 .89 .13 .79
HO_2 AI_2O_3 FeO Fe_2O_3 MnO MgO CaO Na_2O K_2O BaO $Total$ Si	46. 0. 34. 1. 0. 0. 0. 0. 0. 9. 0. 94. 6.	22 46. 74 0. 65 34. 28 2. 00 0. 00 0. 66 0. 00 0. 66 0. 91 8. 28 0. 34 93. 19 6.	12 48 06 0 71 36 00 1 00 0 02 0 39 0 005 0 84 8 12 0 38 98 21 6	.74 47 .03 0 .98 36 .96 1 .00 0 .03 0 .27 0 .07 0 .99 1 .97 6 .12 0 .06 95 .20 6	.46 47 .05 0 .25 35 .86 2 .00 0 .02 0 .40 0 .08 0 .95 8 .07 0 .18 6	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	47.46 0.13 34.01 2.34 0.00 0.00 0.84 0.04 0.96 9.02 0.16 94.96 6.29	48.12 0.06 36.52 1.92 0.00 0.28 0.04 1.68 7.75 0.11 96.47 6.20	47.32 0.24 33.04 2.34 0.00 0.00 1.05 0.06 0.48 9.88 0.16 94.57 6.33	46.6 0.0 35.5 1.8 0.0 0.0 0.4 0.0 0.8 9.0 0.2 94.6 6.1	$\begin{array}{cccccccccccccccccccccccccccccccccccc$.43 .06 .12 .73 .00 .00 .46 .02 .93 .89 .13 .79 .20
HO_{2} $AI_{2}O_{3}$ FeO $Fe_{2}O_{3}$ MnO MgO CaO $Na_{2}O$ $K_{2}O$ BaO $Total$ Si AI^{IV}	46. 0. 34. 1. 0. 0. 0. 0. 9. 0. 94. 6.	22 46. 74 0. 65 34. 28 2. 00 0. 60 0. 00 0. 66 0. 00 0. 66 0. 91 8. 28 0. 34 93. 19 6. 81 1.	12 48 06 0 71 36 00 1 00 0 00 0 02 0 39 0 005 0 84 8 12 0 38 98 21 6 79 1	.74 47 .03 0 .98 36 .96 1 .00 0 .03 0 .27 0 .07 0 .99 1 .97 6 .12 0 .06 95 .20 6 .80 1	.46 47 .05 0 .25 35 .86 2 .00 0 .02 0 .03 0 .040 0 .08 0 .08 0 .095 8 .007 0 .00 95 .118 6 .82 1	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	47.46 0.13 34.01 2.34 0.00 0.00 0.84 0.04 0.96 9.02 0.16 94.96 6.29 1.71	48.12 0.06 36.52 1.92 0.00 0.00 0.28 0.04 1.68 7.75 0.11 96.47 6.20 1.80	47.32 0.24 33.04 2.34 0.00 0.00 1.05 0.06 0.48 9.88 0.16 94.57 6.33 1.67	46.6 0.0 35.5 1.8 0.0 0.0 0.4 0.0 0.4 0.0 0.2 94.6 6.1 1.8	$\begin{array}{cccccccccccccccccccccccccccccccccccc$.43 .06 .12 .73 .00 .00 .46 .02 .93 .89 .13 .79 .20 .80
HO_{2} $Al_{2}O_{3}$ FeO $Fe_{2}O_{3}$ MnO MgO CaO $Na_{2}O$ $K_{2}O$ BaO $Total$ Si Al^{IV} Al^{VI}	46. 0. 34. 1. 0. 0. 0. 0. 9. 0. 94. 6. 1. 3.	22 46. 74 0. 65 34. 28 2. 00 0. 66 0. 00 0. 66 0. 00 0. 66 0. 91 8. 28 0. 34 93. 19 6. 81 1. 65 3.	12 48 06 0 71 36 00 1 00 0 00 0 02 0 339 0 005 0 84 8 12 0 38 98 21 6 779 1 72 3	.74 47 .03 0 .98 36 .96 1 .00 0 .03 0 .227 0 .07 0 .990 1 .997 6 .12 0 .06 95 .20 6 .80 1 .75 3	.46 47 .05 0 .25 35 .86 2 .00 0 .02 0 .40 0 .08 0 .95 8 .07 0 .00 95 .18 6 .82 1 .74 3	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	47.46 0.13 34.01 2.34 0.00 0.00 0.84 0.04 0.96 9.02 0.16 94.96 6.29 1.71 3.60	48.12 0.06 36.52 1.92 0.00 0.28 0.04 1.68 7.75 0.11 96.47 6.20 1.80 3.75	47.32 0.24 33.04 2.34 0.00 0.00 1.05 0.06 0.48 9.88 0.16 94.57 6.33 1.67 3.54	46.6 0.0 35.5 1.8 0.0 0.0 0.4 0.0 0.4 0.0 0.2 94.6 6.1 1.8 3.7	$\begin{array}{cccccccccccccccccccccccccccccccccccc$.43 .06 .12 .73 .00 .00 .46 .02 .93 .89 .13 .79 .20 .80 .73
$\begin{array}{c} \text{I1O}_2\\ \text{Al}_2\text{O}_3\\ \text{FeO}\\ \text{FeO}\\ \text{Fe}_2\text{O}_3\\ \text{MnO}\\ \text{MgO}\\ \text{CaO}\\ \text{Na}_2\text{O}\\ \text{K}_2\text{O}\\ \text{BaO}\\ \text{Total}\\ \end{array}$	46. 0. 34. 1. 0. 0. 0. 0. 0. 9. 0. 94. 6. 1. 3. 0.	22 46. 74 0. 65 34. 28 2. 00 0. 66 0. 00 0. 66 0. 00 0. 66 0. 91 8. 28 0. 34 93. 19 6. 81 1. 65 3. 07 0.	12 48 06 0 71 36 00 1 00 0 02 0 339 0 005 0 84 8 12 0 38 98 21 6 779 1 772 3 01 0	$\begin{array}{rrrr} .74 & 47 \\ .03 & 0 \\ .98 & 36 \\ .96 & 1 \\ .00 & 0 \\ .03 & 0 \\ .27 & 0 \\ .07 & 0 \\ .07 & 0 \\ .07 & 0 \\ .07 & 0 \\ .07 & 0 \\ .07 & 0 \\ .07 & 0 \\ .00 & 95 \\ .00 & .0 \\ .00 & 0 \\ .0$.46 47 .05 0 .25 35 .86 2 .00 0 .02 0 .40 0 .08 0 .08 0 .07 0 .00 95 .18 6 .82 1 .74 3 .00 0	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	$\begin{array}{c} 47.46\\ 0.13\\ 34.01\\ 2.34\\ 0.00\\ 0.00\\ 0.84\\ 0.04\\ 0.96\\ 9.02\\ 0.16\\ 94.96\\ \hline 6.29\\ 1.71\\ 3.60\\ 0.01\\ \end{array}$	48.12 0.06 36.52 1.92 0.00 0.28 0.04 1.68 7.75 0.11 96.47 6.20 1.80 3.75 0.01	47.32 0.24 33.04 2.34 0.00 1.05 0.06 0.48 9.88 0.16 94.57 6.33 1.67 3.54 0.02	46.6 0.0 35.5 1.8 0.0 0.0 0.4 0.0 0.4 0.0 0.2 94.6 6.1 1.8 3.7 0.0	$\begin{array}{cccccccccccccccccccccccccccccccccccc$.43 .06 .12 .73 .00 .00 .46 .02 .93 .89 .13 .79 .20 .80 .73 .01
$\begin{array}{c} \text{I1O}_2\\ \text{Al}_2\text{O}_3\\ \text{FeO}\\ \text{FeO}\\ \text{Fe}_2\text{O}_3\\ \text{MnO}\\ \text{MgO}\\ \text{CaO}\\ \text{MgO}\\ \text{CaO}\\ \text{Na}_2\text{O}\\ \text{K}_2\text{O}\\ \text{BaO}\\ \text{Total}\\ \end{array}$	46. 0. 34. 1. 0. 0. 0. 0. 0. 9. 0. 94. 6. 1. 3. 0. 0.	22 46. 74 0. 65 34. 28 2. 00 0. 00 0. 66 0. 00 0. 66 0. 00 0. 66 0. 91 8. 28 0. 34 93. 19 6. 81 1. 65 3. 07 0. 14 0.	12 48 06 0 71 36 00 1 00 0 02 0 39 0 005 0 84 8 12 0 38 98 21 6 779 1 72 3 01 0 23 0	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} 47.46\\ 0.13\\ 34.01\\ 2.34\\ 0.00\\ 0.00\\ 0.84\\ 0.04\\ 0.96\\ 9.02\\ 0.16\\ 94.96\\ 6.29\\ 1.71\\ 3.60\\ 0.01\\ 0.26\end{array}$	48.12 0.06 36.52 1.92 0.00 0.28 0.04 1.68 7.75 0.11 96.47 6.20 1.80 3.75 0.01 0.21	47.32 0.24 33.04 2.34 0.00 1.05 0.06 0.48 9.88 0.16 94.57 6.33 1.67 3.54 0.02 0.26	46.6 0.0 35.5 1.8 0.0 0.0 0.4 0.0 0.4 0.0 0.2 94.6 6.1 1.8 3.7 0.0 0.2 94.6	$\begin{array}{cccccccccccccccccccccccccccccccccccc$.43 .06 .12 .73 .00 .00 .00 .00 .00 .01 .13 .79 .20 .80 .73 .01 .19
	46. 0. 34. 1. 0. 0. 0. 0. 0. 9. 0. 94. 6. 1. 3. 0. 0. 0.	22 46. 74 0. 65 34. 28 2. 00 0. 66 0. 00 0. 66 0. 00 0. 66 0. 90 0. 60 1. 91 8. 28 0. 34 93. 19 6. 81 1. 65 3. 07 0. 14 0. 00 0.	12 48 06 0 71 36 00 1 00 0 02 0 339 0 005 0 84 8 12 0 38 98 21 6 779 1 72 3 01 0 23 0 00 0	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} 47.46\\ 0.13\\ 34.01\\ 2.34\\ 0.00\\ 0.00\\ 0.84\\ 0.04\\ 0.96\\ 9.02\\ 0.16\\ 94.96\\ 6.29\\ 1.71\\ 3.60\\ 0.01\\ 0.26\\ 0.00\\ \end{array}$	48.12 0.06 36.52 1.92 0.00 0.28 0.04 1.68 7.75 0.11 96.47 6.20 1.80 3.75 0.01 0.21 0.00	47.32 0.24 33.04 2.34 0.00 1.05 0.06 0.48 9.88 0.16 94.57 6.33 1.67 3.54 0.02 0.26 0.00	46.6 0.0 35.5 1.8 0.0 0.0 0.4 0.0 0.4 0.0 0.2 94.6 6.1 1.8 3.7 0.0 0.2 94.6 0.1 0.2 94.6 0.1 0.0 0.2 94.6 0.0 0.2 94.6 0.0 0.2 0.0 0.0 0.2 0.0 0.0 0.0	$\begin{array}{cccccccccccccccccccccccccccccccccccc$.43 .06 .12 .73 .00 .00 .00 .00 .00 .00 .00 .00 .01 .19 .00
	46. 0. 34. 1. 0. 0. 0. 0. 9. 0. 94. 6. 1. 3. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0.	22 46. 74 0. 65 34. 28 2. 00 0. 66 0. 00 0. 66 0. 00 0. 66 0. 90 0. 60 1. 91 8. 28 0. 34 93. 19 6. 81 1. 65 3. 07 0. 14 0. 00 0. 00 0.	12 48 06 0 71 36 00 1 00 0 02 0 39 0 05 0 84 8 12 0 38 98 21 6 779 1 72 3 01 0 23 0 000 0	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} 47.46\\ 0.13\\ 34.01\\ 2.34\\ 0.00\\ 0.00\\ 0.84\\ 0.04\\ 0.96\\ 9.02\\ 0.16\\ 94.96\\ 6.29\\ 1.71\\ 3.60\\ 0.01\\ 0.26\\ 0.00\\ 0.00\\ 0.00\\ \end{array}$	48.12 0.06 36.52 1.92 0.00 0.28 0.04 1.68 7.75 0.11 96.47 6.20 1.80 3.75 0.01 0.21 0.00 0.00	47.32 0.24 33.04 2.34 0.00 1.05 0.06 0.48 9.88 0.16 94.57 6.33 1.67 3.54 0.02 0.26 0.00 0.00	46.6 0.0 35.5 1.8 0.0 0.0 0.4 0.0 0.4 0.0 0.2 94.6 6.1 1.8 3.7 0.0 0.2 94.6 0.1 0.2 94.6 0.1 0.2 94.6 0.0 0.2 94.6 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0	$\begin{array}{cccccccccccccccccccccccccccccccccccc$.43 .06 .12 .73 .00 .00 .00 .00 .00 .00 .00 .00 .01 .19 .00 .00
	46. 0. 34. 1. 0. 0. 0. 0. 0. 9. 0. 94. 6. 1. 3. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0.	22 46. 74 0. 65 34. 28 2. 00 0. 66 0. 00 0. 66 0. 00 0. 60 1. 91 8. 28 0. 34 93. 19 6. 81 1. 65 3. 07 0. 14 0. 00 0. 00 0.	12 48 06 0 71 36 00 1 00 0 02 0 339 0 005 0 84 8 12 0 38 98 21 6 779 1 72 3 01 0 223 0 000 0 000 0	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} 47.46\\ 0.13\\ 34.01\\ 2.34\\ 0.00\\ 0.00\\ 0.84\\ 0.04\\ 0.96\\ 9.02\\ 0.16\\ 94.96\\ 6.29\\ 1.71\\ 3.60\\ 0.01\\ 0.26\\ 0.00\\ 0.00\\ 0.17\\ \end{array}$	48.12 0.06 36.52 1.92 0.00 0.28 0.04 1.68 7.75 0.11 96.47 6.20 1.80 3.75 0.01 0.21 0.00 0.00 0.00	47.32 0.24 33.04 2.34 0.00 0.00 1.05 0.06 0.48 9.88 0.16 94.57 6.33 1.67 3.54 0.02 0.26 0.00 0.00 0.00	46.6 0.0 35.5 1.8 0.0 0.0 0.4 0.0 0.4 0.0 0.2 94.6 6.1 1.8 3.7 0.0 0.2 94.6 0.2 94.6 0.0 0.2 94.6 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0	$\begin{array}{cccccccccccccccccccccccccccccccccccc$.43 .06 .12 .73 .00 .00 .00 .00 .00 .00 .00 .00 .01 .19 .00 .00 .09
HO_2 Al_2O_3 FeO Fe_2O_3 MnO MgO CaO Na_2O K_2O BaO Total Si Al^{IV} Al^{VI} Ti Fe^{2^+} Fe^{3^+} Mn Mg Ca	46. 0. 34. 1. 0. 0. 0. 0. 0. 9. 0. 94. 94. 6. 1. 3. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0.	22 46. 74 0. 65 34. 28 2. 00 0. 66 0. 000 0. 66 1. 91 8. 28 0. 34 93. 19 6. 81 1. 65 3. 007 0. 14 0. 000 0. 13 0.	12 48 06 0 71 36 00 1 00 0 02 0 39 0 08 0 05 0 84 8 12 0 38 98 21 6 79 1 72 3 01 0 223 0 000 0 000 0 000 0 001 0	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} 47.46\\ 0.13\\ 34.01\\ 2.34\\ 0.00\\ 0.00\\ 0.84\\ 0.04\\ 0.96\\ 9.02\\ 0.16\\ 94.96\\ 6.29\\ 1.71\\ 3.60\\ 0.01\\ 0.26\\ 0.00\\ 0.00\\ 0.17\\ 0.01\\ \end{array}$	$\begin{array}{c} 48.12\\ 0.06\\ 36.52\\ 1.92\\ 0.00\\ 0.28\\ 0.04\\ 1.68\\ 7.75\\ 0.11\\ 96.47\\ 6.20\\ 1.80\\ 3.75\\ 0.01\\ 0.21\\ 0.00\\ 0.00\\ 0.05\\ 0.00\\ 0.05\\ 0.00\\ \end{array}$	47.32 0.24 33.04 2.34 0.00 0.00 1.05 0.06 0.48 9.88 0.16 94.57 6.33 1.67 3.54 0.02 0.26 0.00 0.00 0.01 0.01	46.6 0.0 35.5 1.8 0.0 0.0 0.4 0.0 0.4 0.0 0.2 94.6 6.1 1.8 3.7 0.0 0.2 94.6 0.2 94.6 0.0 0.2 94.6 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0	$\begin{array}{cccccccccccccccccccccccccccccccccccc$.43 .06 .12 .73 .00 .00 .00 .00 .00 .00 .00 .00 .46 .02 .93 .89 .13 .79 .20 .80 .73 .01 .19 .00 .09 .00
	46. 0. 34. 1. 0. 0. 0. 0. 0. 9. 0. 94. 94. 6. 1. 3. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0	22 46. 74 0. 65 34. 28 2. 00 0. 66 0. 00 0. 66 1. 91 8. 28 0. 34 93. 19 6. 81 1. 65 3. 07 0. 14 0. 00 0. 13 0. 00 0.	12 48 06 0 71 36 00 1 00 0 02 0 39 0 005 0 84 8 12 0 38 98 21 6 79 1 72 3 01 0 023 0 000 0 000 0 001 0 001 0	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} 47.46\\ 0.13\\ 34.01\\ 2.34\\ 0.00\\ 0.00\\ 0.84\\ 0.04\\ 0.96\\ 9.02\\ 0.16\\ 94.96\\ 6.29\\ 1.71\\ 3.60\\ 0.01\\ 0.26\\ 0.00\\ 0.00\\ 0.17\\ 0.01\\ 0.01\\ 0.01\\ \end{array}$	48.12 0.06 36.52 1.92 0.00 0.28 0.04 1.68 7.75 0.11 96.47 6.20 1.80 3.75 0.01 0.21 0.00 0.00 0.05 0.00	47.32 0.24 33.04 2.34 0.00 0.00 1.05 0.06 0.48 9.88 0.16 94.57 6.33 1.67 3.54 0.02 0.26 0.00 0.00 0.21 0.01	46.6 0.0 35.5 1.8 0.0 0.0 0.4 0.0 0.4 0.0 0.2 94.6 6.1 1.8 3.7 0.0 0.2 94.6 0.2 94.6 0.0 0.2 94.6 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0	$\begin{array}{cccccccccccccccccccccccccccccccccccc$.43 .06 .12 .73 .00 .00 .00 .00 .00 .00 .00 .00 .46 .02 .93 .89 .13 .79 .20 .80 .73 .01 .19 .00 .09 .00 .01
	46. 0. 34. 1. 0. 0. 0. 0. 0. 9. 0. 94. 6. 1. 3. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0	22 46. 74 0. 65 34. 28 2. 00 0. 66 0. 00 0. 66 0. 00 0. 66 1. 91 8. 28 0. 34 93. 19 6. 81 1. 65 3. 07 0. 14 0. 00 0. 13 0. 00 0. 01 0. 02 0.	12 48 06 0 71 36 00 1 00 0 02 0 39 0 005 0 84 8 12 0 38 98 21 6 79 1 72 3 01 0 023 0 000 0 000 0 001 0 011 0 023 0 038 0 000 0 000 0 001 0 011 0 027 0	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	47.46 0.13 34.01 2.34 0.00 0.00 0.84 0.04 0.96 9.02 0.16 94.96 6.29 1.71 3.60 0.01 0.26 0.00 0.00 0.00 0.17 0.01 0.01 0.25	48.12 0.06 36.52 1.92 0.00 0.28 0.04 1.68 7.75 0.11 96.47 6.20 1.80 3.75 0.01 0.21 0.00 0.00 0.00 0.05 0.00 0.01 0.42	47.32 0.24 33.04 2.34 0.00 0.00 1.05 0.06 0.48 9.88 0.16 94.57 6.33 1.67 3.54 0.02 0.26 0.00 0.00 0.21 0.01 0.01	46.6 0.0 35.5 1.8 0.0 0.0 0.4 0.0 0.4 0.0 0.2 94.6 6.1 1.8 3.7 0.0 0.2 94.6 0.2 94.6 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 0.0 0.0 0.0 0.2 0.0 0.0	$\begin{array}{cccccccccccccccccccccccccccccccccccc$.43 .06 .12 .73 .00 .00 .00 .00 .00 .46 .02 .93 .89 .13 .79 .20 .80 .73 .01 .19 .00 .00 .01 .24
$I10_{2}$ $AI_{2}O_{3}$ FeO $Fe_{2}O_{3}$ MnO MgO CaO $Na_{2}O$ $K_{2}O$ BaO $Total$ Si AI^{IV} AI^{VI} Ti Fe^{2+} Fe^{3+} Mn Mg Ca Ba Na K	46. 0. 34. 1. 0. 0. 0. 0. 0. 9. 0. 94. 6. 1. 3. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0	22 46. 74 0. 65 34. 28 2. 00 0. 66 0. 00 0. 66 0. 00 0. 66 0. 90 0. 60 1. 91 8. 28 0. 34 93. 19 6. 81 1. 65 3. 07 0. 14 0. 00 0. 13 0. 00 0. 113 0. 001 0. 16 0.	12 48 06 0 71 36 00 1 00 0 02 0 39 0 05 0 84 8 12 0 38 98 21 6 79 1 72 3 01 0 23 0 000 0 000 0 001 0 27 0	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} 47.46\\ 0.13\\ 34.01\\ 2.34\\ 0.00\\ 0.00\\ 0.84\\ 0.04\\ 0.96\\ 9.02\\ 0.16\\ 94.96\\ 6.29\\ 1.71\\ 3.60\\ 0.01\\ 0.26\\ 0.00\\ 0.00\\ 0.17\\ 0.01\\ 0.25\\ 1.53\\ \end{array}$	48.12 0.06 36.52 1.92 0.00 0.28 0.04 1.68 7.75 0.11 96.47 6.20 1.80 3.75 0.01 0.21 0.00 0.00 0.00 0.05 0.00 0.01 0.42	47.32 0.24 33.04 2.34 0.00 0.00 1.05 0.06 0.48 9.88 0.16 94.57 6.33 1.67 3.54 0.02 0.26 0.00 0.00 0.21 0.01 0.01 0.01 2.54	46.6. 0.0. 35.5: 1.8 0.00 0.4 0.00 0.4 0.00 0.2 94.6 6.1 ⁴ 1.8 3.7 ⁷ 0.00 0.2 94.6 0.2 94.6 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.2 94.6 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0	$\begin{array}{cccccccccccccccccccccccccccccccccccc$.43 .06 .12 .73 .00 .00 .00 .00 .00 .00 .46 .02 .93 .89 .13 .79 .20 .80 .73 .01 .19 .00 .00 .01 .24 .52

Chlorite											
Microstructure	Chl alor	ng S ₁					Chlalor	ng S ₂			
Analyses	Chl19	Chl24	Chl31	Chl39	Chl47	Chl48	Chl32	Chl36	Chl37	Chl43	Chl44
SiO ₂	23.61	27.05	26.37	24.59	26.15	25.30	25.92	23.06	23.88	24.18	23.90
TiO ₂	0.01	0.07	0.04	0.08	0.03	0.01	0.07	0.13	0.21	0.01	0.02
Al_2O_3	23.80	26.27	25.81	24.56	25.90	24.93	25.48	23.46	23.50	24.10	24.41
FeO	27.14	25.36	25.27	25.90	24.13	24.72	26.16	25.92	24.89	26.39	25.48
MnO	0.16	0.10	0.09	0.13	0.11	0.07	0.13	0.09	0.15	0.08	0.15
MgO	12.46	10.72	10.65	12.12	11.04	11.23	11.61	11.77	11.48	12.69	13.00
Total	87.18	89.57	88.23	87.38	87.36	86.26	89.37	84.43	84.11	87.45	86.96
Si	2.53	2.73	2.71	2.59	2.70	2.67	2.66	2.54	2.61	2.56	2.54
Ti	n.c.	0.01	n.c.	0.01	n.c.	n.c.	0.01	0.01	0.02	n.c.	n.c.
Al	3.00	3.13	3.13	3.05	3.15	3.10	3.08	3.04	3.02	3.01	3.06
Fe ²⁺	2.43	2.14	2.17	2.28	2.08	2.18	2.24	2.39	2.27	2.34	2.26
Mn	0.02	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01	0.01
Mg	1.99	1.61	1.63	1.90	1.70	1.76	1.77	1.93	1.87	2.00	2.06
XFe ²⁺	0.55	0.57	0.57	0.55	0.55	0.55	0.56	0.55	0.55	0.54	0.52
Microstructure Chl randomly oriented											
Analyses	Chl03	Chl04	ChlC	08a Ch	113 Chl	16 Chl18					
SiO ₂	23.04	24.28	24.	.18 24	.46 23.	59 24.03					
TiO ₂	0.01	0.05	0.	.03 0	.05 n	.d. 0.04					
Al_2O_3	23.41	23.70	24.	.19 24	.29 23.	17 23.48					
FeO	26.65	26.52	26.	.44 25	.54 23.	44 26.20					
MnO	0.09	0.12	0.	.10 0	.04 0.	12 0.12					

 Table 5.2 Microprobe analyses of chlorite.

Footnote: structural formulae calculated with AX program of T. Holland, on the basis of 14 O. Chlorite is assumed as completely hydrated. Stoichiometric-estimated Fe^{3+} , assuming cation sum equal to 10 for 14 oxygen, is negligible.

MgO

Total

Si

Ti

Al

 Fe^{2+}

Mn

Mg

XFe²⁺

12.76

85.96

2.50

n.c.

3.00

2.42

0.01

2.06

0.54

12.44

87.11

2.59

n.c.

2.98

2.36

0.01

1.98

0.54

11.88

86.82

2.58

n.c.

3.04

2.36

0.01

1.89

0.56

12.28

86.66

2.60

n.c.

3.04

2.27

n.c.

1.94

0.54

12.57

82.89

2.61

n.c.

3.02

2.17

0.01

2.07

0.51

12.61

86.48

2.58

n.c.

2.97

2.35

0.01

2.01

0.54

 Table 5.3 Microprobe analyses of chloritoid.

Footnote: structural formulae calculated with CalcMin program (Brandelik, 2009) on the basis of 6 O. Chloritoid is assumed as completely hydrated, with 4 H. Fe³⁺ has been stoichiometric-estimated with the following assumption: $Fe^{3+}=(2-Ti-0.5*AI)/0.5$.

Chloritoid									
Analyses	Cld06	Cld07	Cld08	Cld12	Cld14	Cld17	Cld18	Cld22	Cld25
SiO ₂	23.23	23.55	22.75	23.24	23.50	23.38	22.33	23.02	23.41
TiO_2	0.03	0.05	0.06	n.d.	0.01	n.d.	0.13	0.02	n.d.
Al_2O_3	39.72	40.66	38.40	39.90	40.15	39.43	37.88	39.72	38.60
Cr_2O_3	0.02	0.01	0.03	0.02	0.02	0.01	0.06	0.02	0.02
FeOtot	25.72	26.02	25.92	27.82	25.47	25.34	27.82	25.47	25.34
MnO	0.39	0.45	0.41	0.42	0.42	0.46	0.38	0.43	0.40
MgO	2.21	2.04	2.41	2.07	2.36	2.10	1.89	2.24	2.21
Total	91.31	92.77	89.96	93.47	91.93	90.71	90.49	90.91	89.97
Si	0.98	0.98	0.98	0.97	0.98	0.99	0.97	0.98	1.00
Ti	n.c.								
Al	1.98	1.99	1.95	1.96	1.98	1.97	1.94	1.99	1.95
Cr	n.c.								
Fe ³⁺	0.06	0.05	0.09	0.10	0.05	0.04	0.13	0.06	0.04
Fe ²⁺	0.85	0.86	0.85	0.87	0.84	0.86	0.88	0.85	0.87
Mn	0.01	0.02	0.01	0.01	0.02	0.02	0.01	0.02	0.01
Mg	0.14	0.13	0.15	0.13	0.15	0.13	0.12	0.14	0.14
XFe ²⁺	0.86	0.87	0.85	0.87	0.85	0.87	0.88	0.86	0.86

Bulk-rock chemistry (Table 5.4) of the hematite-rich phyllite MP4 was obtained through WD-XRF analysis on pressed powder pellets, using a Philips PW1480 spectrometer at the Department of Biological, Geological and Environmental Sciences of Bologna University. Loss on ignition (L.O.I.) was measured with standard gravimetric procedures, after heating the powder at 950 °C for about 12 h.

The FeO content was measured by titration at the Service d'Analyse des Roches et des Mineraux of Nancy. The procedure involves an attack with HF + H_2SO_4 in a teflon beaker on a hot plate and a direct volumetric titration by $K_2Cr_2O_7$ after cooling, and resulted 0.93 ± 0.05 wt. %. The Fe₂O₃ amount content of the rock is 7.00 wt. %.

Whole-rock composition, recalculated excluding L.O.I., was used to classify the protolith, assuming isochemical metamorphism. Normalization diagram of the bulk rock chemistry to PAAS (Post-Archean Australian Shale; Taylor & McLennan, 1985) and to UCC (Upper Continental Crust; McLennan, 2001) indicates depletion in CaO and Na₂O for the MP4 protholith, in agreement with the lack of modal plagioclase and garnet. In

the Fe₂O₃/K₂O *vs.* SiO₂/Al₂O₃ (Herron, 1988) diagram the selected sample plots in the wacke field.

Table 5.4 Whole-rock composition of the hematite-rich metapelite MP4. *Footnote:* (a) bulk rock composition of MP4 by XRF; (b) bulk rock composition of MP4 by XRF with FeO by titration and Fe₂O₃ calculated as [Fe₂O₃^{tot}-(FeO*1.11)]; (c) compositions used for thermodynamic modelling. $XFe_2O_3 = Fe_2O_3/Fe_2O_3^{tot}$

		XRF	Thermodynamic modelling (c)								
	XRF (a)	and titration (b)	MnKNFMASHT system with XRF bulk composition (FeOtot = FeO) (Fig. 5.4)	MnKNFMASHTO system with FeO measured by titration (Fig. 5.5)	MnKNFMASHTO system with Fe ₂ O ₃ estimated by thermodynamic modelling (Fig. 5.8)						
	wt. %	wt. %	wt. %	wt. %	wt. %						
SiO ₂	74.87	74.96	76.80	76.25	76.39						
TiO_2	0.42	0.42	0.43	0.43	0.43						
Al_2O_3	11.59	11.60	11.89	11.80	11.82						
Fe_2O_3	8.03	7.00	-	7.12	5.31						
FeO	-	0.93	7.41	0.95	2.59						
MnO	0.05	0.05	0.05	0.05	0.05						
MgO	0.69	0.69	0.71	0.70	0.70						
CaO	0.12	0.12	-	-	-						
Na ₂ O	0.33	0.33	0.34	0.34	0.34						
K_2O	2.32	2.32	2.38	2.36	2.36						
P_2O_5	0.09	0.09	-	-	-						
L.O.I.	1.48	1.48	in excess	in excess	in excess						
Total	99.99	100.00	100.01	100.00	99.99						
XFe ₂ O ₃		0.87		0.87	0.65						

5.5 GEOTHERMOBAROMETRY

5.5.1 Chlorite-chloritoid thermometry

The chlorite-chloritoid Fe/Mg exchange thermometer of Vidal *et al.* (1999) was used to determine the temperature of the post-kinematic event recognized in MP4. This thermometer is suitable because it was empirically calibrated for greenschist- to blueschist-facies conditions (300-550 °C) and pressure does not have to be known. Only pairs of point analyses of coexisting minerals with no reaction and/or disequilibrium structures such as chloritoid overgrown by chlorite or with MnO > 1 wt.% or $(Cr_2O_3+MnO+CaO+Na_2O+K_2O) > 2$ wt. % were used with this thermometer.

From an initial dataset comprising 30 chlorite analysis and 20 chloritoid analyses, 8 chlorite-chloritoid pairs have been selected because they fully satisfied the equilibrium (Fig. 5.2d) and compositional criteria. The derived temperature is 475 ± 15 °C.

5.5.2 Phase diagram calculations

Thermodynamic modelling (Perple_X, Connolly 1990; Connolly & Petrini, 2002) was performed to reproduce measured mineral compositions and the observed assemblage, represented by randomly oriented muscovite and chlorite ($XFe^{2+} = 0.54$), chloritoid (XFe²⁺ = 0.86), quartz, hematite and rutile. The thermodynamic dataset of Holland & Powell (1998, with the 2002 upgrade) and the compensated Redlich Kwong (CORK) fluid equation of state by Holland & Powell (1991, 1998) were used. SiO₂ and H₂O are in excess. Input compositions are expressed in wt.%. Solid solution models in the MnKNFMASHT system pseudosection with FeO_{tot} = FeO are white-mica (Holland & Powell, 1998), chloritoid (White et al., 2000), garnet (Holland & Powell, 1998), biotite (Tajcmanová et al., 2009), chlorite (Holland et al., 1998), carpholite (Massonne & Willner, 2008), feldspar (Fuhrman & Lindsley, 1988) and ilmenite (ideal ilmenitegeikielite-pyrophanite solid solution; http://www.perplex.ethz.ch/PerpleX_solution_model_glossary.html). In the MnKNFMASHTO system pseudosections the available solution models involving Fe³⁺ end-members for white-mica, chloritoid, carpholite (Smye et al., 2010), garnet (White et al., 2005), biotite (Tajcmanová et al., 2009), hematite and magnetite (Andersen & Lindsley, 1988) were used. There is no solid solution model with Fe^{3+} for chlorite, so the solid solution model by Holland et al. (1998) was used. The feldspar model by Fuhrman & Lindsley (1988) was also used. Pure phases are rutile, paragonite kyanite, sillimanite, pyrophyllite and stilpnomelane.

P-T pseudosection in the system MnKNFMASHT with FeO_{tot} = FeO

Neglecting ferric iron in the input composition allows us to investigate if the choice of not considering Fe₂O₃ in an obviously Fe₂O₃-bearing rock can reproduce, reasonably enough, the observed mineral assemblage and compositions. The whole-rock XRF composition was assumed to be the equilibrium volume, recalculated excluding Ca and P₂O₅, which are contained in apatite, and L.O.I. According to this procedure FeO_{tot} = FeO = 7.41 wt. %. FeO_{tot} was calculated from the conversion of XRF Fe₂O₃. The considered *P*-*T* range is 4-15 kbar and 300-550 °C respectively. Muscovite and chloritoid are stable in each field of the pseudosection (Fig. 5.4). Garnet is stable above 500 °C and 6 kbar. Biotite is stable at temperatures above 500 °C and pressures below 8 kbar. Carpholite appears at 15 kbar and 300 °C. Rutile breaks down above 425-500 °C, replaced by ilmenite, and appears again at 525 °C and pressures higher than 13 kbar.

In this *P*-*T* range, the calculated XFe^{2+} for chlorite and chloritoid are 0.57-0.85 and 0.84-0.98, respectively. The observed XFe^{2+} chlorite composition ($XFe^{2+} = 0.54$) is close to the lower end of the calculated range ($XFe^{2+} = 0.57-0.85$). The observed chloritoid composition ($XFe^{2+}=0.86$) plots in a garnet-bearing field, in disagreement with the observed mineral assemblage. As a consequence the input bulk composition without Fe_2O_3 is inappropriate to reproduce the observed mineral compositions.



Figure 5.4 *P*-*T* pseudosection with $FeO_{tot} = FeO$

P-T pseudosection in the system MnKNFMASHTO with FeO estimated by titration

A *P*-*T* pseudosection was performed using the XRF bulk rock composition and FeO and Fe₂O₃ determined by titration (FeO = 0.93 wt. %; Fe₂O₃ = 7.00 wt. %). The pressure and temperature ranges are as in the previous modelling.

Chloritoid is stable in the whole *P*-*T* range, except at high pressures and low temperatures and low pressures and high temperatures (Fig. 5.5). The garnet stability field occurs below 7 kbar and above 525 °C. Carpholite-out curve is at 350-500 °C, depending on pressure. Hematite and rutile are stable over the entire *P*-*T* grid, accompanied by magnetite above 500 °C and below 8 kbar.

Calculated chlorite XFe^{2+} isopleths are between 0.15 and 0.49, while measured chlorite $XFe^{2+} = 0.54$ is not reproduced. Besides this discrepancy, measured XFe^{2+} isopleths of chloritoid plot in fields that include carpholite, which is in disagreement with the observed mineral assemblage.



Figure 5.5 *P-T* pseudosection with FeO determined by titration.

P-X pseudosection in the system MnKNFMASHTO at fixed temperature

In order to calculate the effective Fe₂O₃ content during metamorphism a P-XFe₂O₃ (XFe₂O₃ = Fe₂O₃/Fe₂O₃^{tot}) pseudosection was performed at a fixed T = 475 °C (Fig. 5.6), derived from the chloritoid-chlorite geothermometer (Vidal *et al.*, 1999). This approach is the same as used by Boger *et al.* (2012) and López-Carmona *et al.* (2013). Stability fields of minerals are reported in Fig. 5.7. Chlorite breaks down at 12 kbar and XFe₂O₃ = 0.8, substituted by carpholite. Biotite is stable only at pressures below 3 kbar and in weakly oxidized rocks (XFe₂O₃ < 0.4). Chloritoid is present over the full range at XFe₂O₃ < 0.95, and is then replaced by garnet, which is stable in extremely oxidized rocks (XFe₂O₃ > 0.95).



Figure 5.6 *P*-*X* pseudosection at 475°C.

Ilmenite is stable below 12 kbar and $XFe_2O_3 < 0.35$. Hematite appears at $XFe_2O_3 > 0.3$ at pressures of 15 kbar and at $XFe_2O_3 > 0.5$ in the lower pressure region. Rutile is stable at 10 kbar at $XFe_2O_3 = 0$, and the pressure of its appearance decreases with increasing XFe_2O_3 . Magnetite is stable in the range of $XFe_2O_3 = 0.02-0.75$.

The $P-XFe_2O_3$ field corresponding to the observed mineral assemblage (muscovite + chlorite + chloritoid + quartz + hematite + rutile) is in the range $XFe_2O_3 = 0.60-0.85$ and

3-12 kbar, delimitated above 10-12 kbar by paragonite-bearing assemblages, by the magnetite in-curve at $XFe_2O_3 = 0.60-0.75$, and by the crystallization of kyanite at $XFe_2O_3 > 0.8$. Isopleths related to XFe^{2+} in chlorite and XFe^{2+} in chloritoid intersect in this field at P = 9-10 kbar and $XFe_2O_3 = 0.65$, equal to FeO = 2.6 wt.% and Fe₂O₃ = 5.3 wt.% (effective Fe₂O₃ content).

P-T pseudosection in the system MnKNFMASHTO with the effective Fe_2O_3 content in rock

Considering the bulk XFe_2O_3 determined from the previously *P*-*X* pseudosection ($XFe_2O_3 = 0.65$), a *P*-*T* pseudosection in the system MnKNFMASHTO was calculated (Fig. 5.8). The field corresponding to the observed mineral assemblage is bounded to high pressure by the paragonite-in, and to high temperature by the magnetite-in curve. The observed mineral assemblage (muscovite + chlorite + chloritoid + quartz + hematite + rutile) and XFe^{2+} values ($XFe^{2+} = 0.54$ for chlorite and $XFe^{2+} = 0.86$ for chloritoid) are successfully reproduced at 475 °C and 9-10 kbar.





Figure 5.8

P-T pseudosection with the bulk rock chemistry calculated in the previous P-X pseudosection.

5.6 DISCUSSION

Calculations of *P*- or *T*-*X*Fe₂O₃ diagrams are good means to evaluate the influence of effective Fe₂O₃ contents in bulk-rock compositions, thus avoiding an overestimation of the ferric iron due to low-temperature processes and/or sample preparation (Boger *et al.*, 2012, López-Carmona *et al.*, 2013). This approach can provide useful results, but the uncertainties involved in thermodynamic modelling must not be undervalued (Powell & Holland, 2008): the wrong choice of the equilibration volume, the inappropriate assessment of the chemical system, the uncertainties related to the activity–composition (*a*–*x*) models are some factors that can introduce significant mistakes in the *P*-*T*-*X*Fe₂O₃ constraints.

Another approach to calculate effective Fe_2O_3 contents in bulk-rock compositions is to combine phase proportions with measured phase compositions. This method overcomes the problem of oxidation during sample preparation. However, it requires actual determination of Fe^{3+} in minerals, as calculation by stoichiometry can introduce significant errors especially when dealing with mica and amphibole. This problem coupled with the uncertainties related to the determination of the mineral modes may produce a bulk composition that fails to reproduce the mineralogical constraints.

5.6.1 Ferric iron influence on low-grade metasedimentary rocks

The thermodynamic results can be used to examine the behaviour of minerals in lowgrade Fe_2O_3 -rich metasedimentary rocks. Figure 9 provides a comparison of two different bulk rock Fe_2O_3 compositions. The typical greenschist-facies assemblage of white mica + chlorite + chloritoid + quartz develops in Fe_2O_3 -free compositions, with garnet and biotite making an appearance starting from about 500 °C as expected in Alpoor metapelites. Rutile and ilmenite are the dominant Ti-oxides, the latter replacing the former at temperatures towards the greenschist- to amphibolite-facies transition.

In Fe₂O₃-rich compositions, garnet-bearing assemblages are shifted, especially at intermediate- to high-pressure conditions, towards higher temperatures than in the "all-bulk-iron-as-FeO" case. But it must be noted that the considered MnNKFMASHTO system does not contain Ca, which has the effect of not stabilizing the ugrandite group end-members of garnet. The calculated garnet composition in an $XFe_2O_3=0.65$ bulk composition is a spessartine with a low ferric iron content (0.005 apfu) and its mode is very low, around 0.02 vol.%.

Biotite is not stable in the $XFe_2O_3=0.65$ bulk composition even at 550 °C, which may be related to the activity-composition model for the Fe³⁺-bearing end-members of biotite. The thermodynamic modelling of the hematite-rich metasedimentary rock MP4 applied the Fe³⁺-Ti-biotite model of Tajcmanová *et al.* (2009), who warned against the use of this extended biotite thermodynamic model in Fe₂O₃-rich compositions. Here, however, the choice of the Ti-Fe³⁺-biotite model correctly reproduced the observed biotite-free assemblage. The biotite-free assemblage could also be the consequence of a lack of alkali-feldspar that is involved in the production of biotite during prograde chlorite breakdown (chlorite + K-feldspar = muscovite + biotite + quartz + H₂O; Spear, 1993). A further explanation is that hematite acted as a sink for ferric iron and no ferric iron was left for the formation of biotite, which is the major Fe³⁺ silicate container in metasedimentary rocks (Guidotti & Dyar, 1991). This hypothesis was tested by performing a pseudosection (not shown) in which hematite and magnetite were excluded from the calculation: without these ferric iron phases, biotite occurs in the the mineral assemblages at *X*Fe₂O₃ contents of 0.65.

Carpholite is another important phase that indicates HP-LT metamorphism in metasedimentary rocks, and its occurrence is long known in the Northern Apennines silicoclastic units of the Triassic Verrucano Formation (Theye *et al.*, 1997). A bulk-rock composition with $XFe_2O_3 = 0.65$ enlarges the carpholite stability towards lower pressures and higher temperatures conditions. At 300 °C the carpholite-in curve moves from 15 kbar in reduced rock compositions to 10 kbar in oxidized bulk compositions. The minimum pressures for the occurrence of carpholite in Fe₂O₃-rich metasedimentary rocks is close to the minimum pressures of 6-8 kbar estimated for carpholite-bearing metapelitic assemblages (Oberhänsli *et al.*, 2001).

Rutile and hematite are the stable oxides within bulk-rock composition with $XFe_2O_3 = 0.65$. Magnetite appear above 450 °C at low pressures whereas magnetite-bearing assemblages need temperatures exceeding 500 °C at high pressures.



5.6.2 Geological implications for the Northern Apennines

Many studies devoted to the Triassic Verrucano from the Tuscan Metamorphic Units of the Northern Apennines document HP-LT Oligocene-Miocene metamorphism (Theye *et al.*, 1997; Giorgetti *et al.*, 1998; Jolivet *et al.*, 1998; Franceschelli & Memmi, 1999; Franceschelli *et al.*, 2004). This metamorphic signature, defined by pressures ranging from 8 to 15 kbar and temperatures of 300-560 °C, is associated to subduction and crustal thickening connected to the Corsica-Sardinia and Adria collision (Jolivet *et al.* 1998). In this geodynamic setting, the role of the Variscan basement is generally overlooked, probably because less thermobarometric constraints on its Alpine evolution are available. Our study confirms the involvement of Variscan basement rocks in the Northern Apennine Oligocene-Miocene orogeny (Conti *et al.*, 1991) and helps to clarify the contrasting peak pressure estimates available for the overlying Verrucano of the

Pisani Mountains (4-5 kbar *vs.* 9-10 kbar: Franceschelli *et al.*, 1981). The Alpine peak metamorphism recorded in the Verrucano from Pisani Mountains is similar to the peak *P-T* values (T=300-560 °C; P=8-15 kbar) found elsewhere in the Verrucano cropping out in the Tuscan Archipelago and southern Tuscany. Hence, both the Variscan basement and its sedimentary cover resided at depths of around 25-30 km after the crustal thickening stage, in a collisional tectonic setting characterized by a geotherm of around 50 °C kbar⁻¹.

5.7 CONCLUSIONS

The Pisani Mountains hematite-rich phyllites case study offers the opportunity to achieve P-T information by taking into account its effective ferric iron content by P-XFe₂O₃.

The main results are:

- 1. Underestimating (or neglegting) ferric iron in oxide-rich phyllites may result in a thermodynamic modelling that reproduces neither the observed mineral assemblage nor the measured mineral compositions. Quantifying Fe_2O_3 is possible through a *P*-*T*-*X*Fe₂O₃ approach whereas the titration method may overestimate the bulk Fe_2O_3 .
- 2. The introduction of ferric iron has the effect of increasing the MgO/(MgO+FeO) ratio in bulk rock chemistry and this in turn affects silicate mineral compositions (i.e. chloritoid and chlorite XFe^{2+}) and the Fe–Mg exchange thermobarometry.
- 3. In Fe₂O₃-rich (and CaO-poor) bulk compositions garnet and carpholite, two important thermobarometric indicators, occur at higher temperatures than in Fe₂O₃-poor bulk compositions. Furthermore, carpholite is predicted at lower pressure in the MnNKFMASHTO system than in the MnNKFMASHT system. Our modelling in Fe₂O₃-rich rocks indicates that the formation of biotite, the typical Fe³⁺-bearing silicate in metasedimentary rocks, can be controlled by the stability of Fe³⁺-oxides such as hematite that is the major sink for Fe³⁺ in the present case study. Alternative explanations include the poor knowledge of the activity-composition model for the Fe³⁺-bearing end-member of biotite, or the lack of precursor K-feldspar. Thus the stability of Fe³⁺-oxides may have the effect of changing the coexisting silicate mineralogy.
- 4. Hematite-rich phyllites of the Buti formation (Pisani Mountains, Northern Apennines) reached 475 °C and 9-10 kbar during the metamorphic event

presumably occurring after nappe-stacking in the Alpine continental collision between Adria and Corsica-Sardinia margins. These new *P-T* values agree with the peak pressures recorded by the Verrucano during the Alpine evolution. Hence, both the Variscan basement and its cover behaved as a coherent tectonic unit during the assembly of the Northern Apennines belt.

5.8 ACKNOWLEDGEMENTS

We are grateful to H. J. Massonne and T. Theye for access to the microprobe at Stuttgart. G. Gasparotto and E. Dinelli provided additional analytical support. Friendly discussion with G. Molli and A. Montanini helped us to move our first steps into the Northern Apennine geology. Comments by A. J. R. White and P. Tropper helped to improve the manuscript. We are also thankful to D. Robinson for editorial handling and additional comments that clarified many point in a previous manuscript. The University of Bologna financed this study (Marco Polo and RFO).

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CHAPTER 6

CONCLUSIONS

6.1 SUMMARY

The thesis deals with the metamorphic evolution of the Northern Apennines Variscan basement. Only few data (comprising *P-T* estimates obtained through classical thermobarometry) were available for this basement because of (i) complex polymetamorphism (Variscan metamorphism and Alpine overprint); (ii) scarce and rare outcrops, that hamper any regional correlation.

The PhD activity aimed to unravel a complete metamorphic history, including the pre-Alpine stages, the possible correlations with other Variscan basement, the evolution between the Variscan and the Alpine orogenesis, and the Alpine peak conditions.

The study of the Variscan basement of Northern Apennines allowed also to examine some general petrological issues, such as the possibility to combine thermodynamic modelling and microstructural analysis to reconstruct a polyphasic metamorphic evolution, the strategies to link petrology and geochronology, the breakdown reaction of monazite in medium-grade metapelites, the geochronology of mylonitic microstructures, and the influence of ferric iron in greenschist-facies metasediments. The Variscan basement occurrences suitable to investigate the above-mentioned questions were located at the Cerreto Pass, in the Pontremoli area and in the Pisani Mountains. The research was carried out with field work, sampling, petrography and microstructural analysis, determination of the bulk rock- and mineral composition, thermodynamic modelling, classical geothermobarometry, monazite chemical dating, and Ar-Ar dating of muscovite. The thesis has been structured according to a chronological order: Variscan orogeny, post-Variscan evolution, Jurassic shearing and Alpine metamorphism. The results are shown in the following paragraphs.

6.2 PETROLOGICAL FINDINGS

1. In the Cerreto micaschists the Mg-rich cores within deformed prekinematic potassic white mica record the pre-peak stage. Garnet rim composition is, instead, related to the peak stage. The late paragenesis and the composition of

Figure of the previous page: View of the Northern Apennines from the Cerreto Pass. The mountains in the background are the Apuane Alps.

potassic white mica and chlorite defining the mylonitic foliation provide information about the retrograde shearing stage (chapter 2). Thus, an integrated approach between microstructural analysis and thermodynamic modelling allows to unravel a polyphase P-T-D evolution.

- 2. The application of thermodynamic modelling and U-Th-Pb chemical dating of monazite on micaschists from the Pontremoli well (chapter 3) allowed to investigate the link between monazite age dating and petrology in medium-grade metapelites. Coronitic microstructures of apatite, allanite, epidote around partially dissolved monazite grains are a clear breakdown microstructure. Besides, the high-Th and -U contents in monazite suggests that the monazite was inherited and became unstable undergoing partial or total dissolution. The breakdown reaction of monazite generated apatite + Th-silicate, allanite, and epidote coronae during prograde metamorphism at 500-600°C and 5-7 kbar.
- 3. Large fish-shaped muscovite crystals in mylonites are not always prekinematic mica-fishes. They can consist of a polycrystalline aggregate of fine-grained muscovite. In this case the fish-shaped microstructure is an asymmetric foliation boudin and muscovite is synkinematic with respect to the mylonitic event (chapter 4). A comparison between the ⁴⁰Ar-³⁹Ar ages of potassic white mica along the mylonitic foliation and within asymmetric foliation boudins in the Cerreto mylonites reveals that mylonitic foliation provides more scatter ages due to prekinematic relics, while asymmetric foliation boudins ages are related only to the mylonitization. Hence, in our case study, asymmetric foliation boudins is a preferential microstructure for obtaining geochronological information about the shearing event.
- 4. The long-standing problem of the uncertainties of the bulk FeO/Fe₂O₃ ratio when performing thermodynamic modelling has been tackled in the chapter 5. In the Pisani Mountains hematite-rich phyllites, neglecting the presence of bulk Fe₂O₃ may lead to incorrect *P*-*T* estimations. Through *P*-*T*-*X*Fe₂O₃ phase diagrams it is possible to quantify the amount of ferric iron during metamorphism. This amount can be different from the ferric iron quantity resulting from the titration method, which is related to the present-day ferric iron in bulk-rock composition. The inclusion of Fe³⁺ in thermodynamic modelling increases the whole-rock MgO/(MgO+FeO) ratio, which in turn affects

chloritoid and chlorite XFe^{2+} in chloritoid-bearing phyllites from the Pisani Mountains. The presence of ferric iron in bulk-rock composition influences the stability of the main rock-forming silicates: in Fe₂O₃-rich compositions garnetand carpholite-in curves shift towards higher temperatures with respect to the Fe₂O₃-free systems. The presence of a ferric-iron oxide (hematite) prevents the formation of biotite from the mineral assemblage even at temperatures approaching 550 °C. Rutile and hematite and magnetite are the stable oxides in ferric iron-rich bulk-rock compositions.

6.3 REGIONAL RECONSTRUCTIONS

The combination of the metamorphic evolution reconstructions for the Cerreto Pass mylonites, the Pontremoli micaschist and the Monti Pisani phyllites allows to depict a long and complex metamorphic history (Figs. 6.1, 6.2).

Carboniferous

The closure of the Rheic Ocean (360 Ma) was followed by the Variscan orogeny with the collision of Gondwana and Laurussia and the formation of Pangea at 300 Ma (Matte, 2001; von Raumer *et al.*, 2009).

In the Cerreto Pass micaschist, the Variscan thickening generated a high-pressure stage occurring at 430-480°C and 11-13.5 kbar. The Variscan peak (600 °C at 9-10 kbar) postdates this stage and was attained at 312-328 Ma (Molli *et al.*, 2002), thanks to correlation with the associated amphibolites.

Some analogy can be established among the tectono-metamorphic Variscan evolution in Northern Apennines and that described for other fragments of the Southern Variscan belt (e.g. Sardinia).

Late Carboniferous- MiddleTriassic

The Late-Carboniferous tectonics dismembered the Variscan chain through intracontinental strike slip shear zones (Arthaud & Matte, 1977; Doglioni 1984; Massari, 1986; Ziegler, 1986; Handy & Zingg, 1991; Rau, 1993; Schmid, 1993; Matte, 2001; Padovano *et al.*, 2012). This tectonic regime is linked to a megashear system affecting the Variscan belt with alternating transpressive and transtensive conditions (Rau, 1990). These megashears caused the displacement of the Sardinian, the Tuscan

and the Calabrian Variscan fragments about 150 km westward away from the "stable" Europe (Rau, 1990 and reference therein).

The Pontremoli metapelites record a metamorphic history younger than 294 ± 5 Ma (U-Th-Pb chemical age on inherited monazite). The replacement of inherited monazite by apatite, allanite and epidote occurred during a prograde evolution at 500-600°C and 5-7 kbar. These metamorphic peak conditions predate a stage characterised by a pressure increase and concomitant cooling to 520°C at 8 kbar, followed by a nearly isothermal cooling to 500°C at 2 kbar. This anticlockwise *P*-*T* path is different from the available metamorphic histories of the pre-Mesozoic basement of the Northern Apennines. An Oligocene-Miocene metamorphic history cannot be completely ruled out. However, the comparison with the few adjacent basement slivers and the shape of the *P*-*T* path here derived support a post-Variscan stage characterised by transpressive (counterclockwise - T_{max} before P_{max}) to transtensive (near isothermal decompression) tectonics. This reconstruction may be related to the Late Carboniferous-Middle Triassic crustal scale shear zone.

The strike-slip shearing affected probably also the Cerreto mylonites, that show some ⁴⁰Ar-³⁹Ar age determinations around 210 and 260 Ma.

Late Triassic-Jurassic

Starting from Late Triassic a distensive tectonic regime predominated. The extensional tectonics led to the final separation between the Variscan Sardinia and the Tuscan Variscan basements. The Variscan basement of Northern Apennines became the western divergent margin of the Adria promontory. Along this separation stretching and thinning occurred (Arthaud & Matte, 1977; Rau, 1990). In the meanwhile, tilted and imbricate blocks formed in the Tuscan Adria, such as in the Southern Alps (Schmidt *et al.*, 1987; Handy, 1987). The drifting stage that led to the formation of the Ligurian-Piedmont ocean, a branch of the Western Tethys, occurred in Middle-Late Jurassic.

In this framework the retrograde evolution in the Cerreto mylonites took place. This stage occurred at 190.7 ± 1.3 Ma (Ar/Ar isotopic age of mylonitic muscovite) and was accompanied by a strong ductile deformation at $450-500^{\circ}$ C and 6-7.5 kbar (thermodynamic modelling). Thus, the exhumation of the Cerreto mylonites was probably connected with the Ligurian-Piedmont ocean-related rifting stage of the Adria margin.

Cretaceous-Eocene

The convergence between Europe (Corsica-Sardinia) and Africa (Adria) led to the subduction of the Ligurian-Piedmont ocean. Eclogitic- to blueschist-metamorphism affected the Liguride ophiolites at 49-40 Ma (Federico *et al.*, 2004). The Variscan basement of Northern Apennines was not involved in the Alpine collision yet. This is confirmed by the lack of Cretaceous-Eocene geochronological data.

Late Oligocene-Early Miocene

The collision between the Corsica-Sardinia block and the Adria microplate caused the underthrusting of the Adria margin (Vai & Martini, 2001). Some sectors of the Variscan basement were affected by the Alpine metamorphism. The compressional structures of Northern Apennines formed in these times (27 Ma; Kligfield *et al.*, 1986). High-pressure greenschist-facies metamorphism affected the Massa Unit in the Apuane Alps (6-8 kbar and 400-500°C; Molli *et al.*, 200, 2002) and the Buti Formation phyllites from the Pisani Mountains (475 °C and 9-10 kbar; Lo Pò & Braga, 2014).

High-pressure lower-temperature metamorphism (350-420°C and 6-12 kbar) affected, instead, the Tuscan Metamorphic Units in Southern Tuscany, where it is extensively recorded by the Verrucano Group metasediments (Theye *et al.*, 1997; Giorgietti *et al.*, 1998; Jolivet *et al.*, 1998).



Figure 6.1. *P-T* paths for the Variscan basement determined for the Cerreto Pass, in the Pontremoli well and in the Pisani Mountains.



Figure 6.2

Tectonic sketches drawn in collaboration with dr. G. Molli reporting the main events recorded in the Variscan basement of Northern Apennines.

6.3.1 Open questions

Some regional points need to be examined in future researches:

- The mafic amphibolites in the Cerreto Pass exhibit a peak metamorphism at 650°C and 8 kbar, calculated using classical geothermobarometry. Our *P-T* estimates from thermodynamic modelling for the associated micaschist is 580-600°C and 9-10 kbar. The differences can be due to the different geothermobarometric methods or, alternatively, these discrepancies are true and need a refinement of the geological interpretations. Thus, it would be interesting to use thermodynamic modelling also for the amphibolites in order to verify if the peak conditions changes.
- A debate is still open about the derivation of the Cerreto metamorphic rocks. They have been attributed in the past to the Tuscan domain (Pandeli *et al.*, 2005) or to a more external domain (Umbrian or Padan domain; Reutter *et al.*, 1983; Vai & Cocozza, 1986). A comparative study on the Cerreto micaschist and the Micaschist Complex in the Larderello subsurface is needed to establish if the two belong to the same crustal sector.
- The reconstructed metamorphic evolution for the Pontremoli micaschists postdates the chemical monazite age 294 ± 5 Ma. Further dating can more precisely constrain the recognized metamorphic stages. In this respect, Ar-Ar dating on Na-rich muscovite and Mg-rich muscovite may provide information about the age of the thermal peak and of the pressure peak, respectively.

6.4 REFERENCES

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SAMPLE INFORMATION

SAMPLE	LOCATION	UNIT	LITHOLOGY	OPTICAL MICROSCOPE	SEM EDS	EMP WDS	XRF	GEO THERMOBAROMETRY	GEO CHRONOLOGY
PB8	Punta Bianca	Phyllitic and quartzitic complex	phyllite	Х					
PB9	Punta Bianca	Phyllitic and quartzitic complex	phyllite	х			х		
PB9b	Punta Bianca	Phyllitic and quartzitic	phyllite	х	Х	х	х		
MP1	Pisani Mountains	Phyllitic and quartzitic complex	phyllite	х					
MP2	Pisani Mountains	Phyllitic and quartzitic	phyllite	Х					
MP3	Pisani Mountains	Phyllitic and quartzitic complex	phyllite	х					
MP4	Pisani Mountains	Phyllitic and quartzitic	chloritoid-bearing phyllite	Х			х		
MP4b	Pisani Mountains	Phyllitic and quartzitic complex	chloritoid-bearing phyllite	Х	Х	Х	х	x	
CER10a	Cerreto Pass	Micaschist Complex	mylonitic micaschist	Х			х		
CER12	Cerreto Pass	Micaschist Complex	mylonitic micaschist	х			x		
CER36-1a	Cerreto Pass	Micaschist Complex	mylonitic micaschist	Х			х		
CER8	Cerreto Pass	Micaschist Complex	garnet-bearing mylonitic micaschist	х	Х	Х	х	х	
CER8 chip	Cerreto Pass	Micaschist Complex	garnet-bearing mylonitic			х			
CER9	Cerreto Pass	Micaschist Complex	mylonitic micaschist	х			x		
CER41-3	Cerreto Pass	Micaschist Complex	garnet-bearing amphibolite	Х			х		
CER40-7	Cerreto Pass	Micaschist Complex	amphibolite	Х			х		
PC1	Cerreto Pass	Micaschist Complex	garnet-bearing mylonitic	X					
PC1b	Cerreto Pass	Micaschist Complex	garnet-bearing mylonitic micaschist	Х	Х		Х	х	

SAMPLE	LOCATION	UNIT	LITHOLOGY	OPTICAL MICROSCOPE	SEM EDS	EMP WDS	XRF	GEO THERMOBAROMETRY	GEO CHRONOLOGY
PC1 chip	Cerreto Pass	Micaschist Complex	garnet-bearing mylonitic			х			
PC2	Cerreto Pass	Micaschist Complex	garnet-bearing mylonitic micaschist	х					
PC2b	Cerreto Pass	Micaschist Complex	garnet-bearing mylonitic micaschist	х	Х				
PC4	Cerreto Pass	Micaschist Complex	garnet-bearing mylonitic	Х					
PC5	Cerreto Pass	Micaschist Complex	ultramylonite	Х					
PC6	Cerreto Pass	Micaschist Complex	mylonitic biotite-micaschist	х					
PC7b	Cerreto Pass	Micaschist Complex	mylonitic biotite-micaschist	х					
PC11	Cerreto Pass	Micaschist Complex	mylonitic micaschist	х					
PC12	Cerreto Pass	Micaschist Complex	mylonitic biotite-micaschist	х					
PC13	Cerreto Pass	Micaschist Complex	ultramylonite	х					
PC14	Cerreto Pass	Micaschist Complex	mylonitic biotite-micaschist	х					
PC15	Cerreto Pass	Micaschist Complex	mylonitic micaschist	х					
PC16	Cerreto Pass	Micaschist Complex	mylonitic micaschist	х					
PC17	Cerreto Pass	Micaschist Complex	mylonitic biotite-micaschist	х					
PC18	Cerreto Pass	Micaschist Complex	mylonitic micaschist	Х					
PC19	Cerreto Pass	Micaschist Complex	green amphibole-amphibolite	Х					
PC20	Cerreto Pass	Micaschist Complex	ultramylonite	х					
PC21	Cerreto Pass	Micaschist Complex	mylonitic micaschist	х	х				
PC22	Cerreto Pass	Micaschist Complex	mylonitic micaschist	Х	х				
PC23	Cerreto Pass	Micaschist Complex	garnet-bearing mylonitic micaschist	Х	Х	х	x	х	

SAMPLE	LOCATION	UNIT	LITHOLOGY	OPTICAL MICROSCOPE	SEM EDS	EMP WDS	XRF	GEO THERMOBAROMETRY	GEO CHRONOLOGY
PC23 chip	Cerreto Pass	Micaschist Complex	garnet-bearing mylonitic			Х			Х
PC24	Cerreto Pass	Micaschist Complex	garnet-bearing mylonitic micaschist	х	Х				
PC24b	Cerreto Pass	Micaschist Complex	garnet-bearing mylonitic	Х					
PC25	Cerreto Pass	Micaschist Complex	mylonitic micaschist	х					
3118	Pontremoli 1 well	Micaschist Complex	phyllite	Х					
AG7	Pontremoli 1 well	Micaschist Complex	garnet-bearing micaschist	Х	х		х		
Pontremoli 8	Pontremoli 1 well	Micaschist Complex	garnet-bearing micaschist	Х	х				
Pontremoli a	Pontremoli 1 well	Micaschist Complex	phyllite	Х					
AG6a	Pontremoli 1 well	Micaschist Complex	phyllite	Х					
AG6 AG6 chip	Pontremoli 1 well Pontremoli 1 well	Micaschist Complex Micaschist Complex	phyllite phyllite	х		х	Х		
Pontremoli 7	Pontremoli 1 well	Micaschist Complex	garnet-bearing micaschist	Х	х	х		Х	х
AG8	Pontremoli 1 well	Micaschist Complex	garnet-bearing micaschist	Х	х				
AP1	Apuane Alps	Phyllitic and quartzitic complex	phyllite	х	х	Х	Х		
AP2	Apuane Alps	Phyllitic and quartzitic	phyllite	Х	х		х		
AP3	Apuane Alps	Phyllitic and quartzitic complex	porphyric schist	х	х		х		
AP4	Apuane Alps	Phyllitic and quartzitic	porphyroid	Х	х		х		
AP5	Apuane Alps	Phyllitic and quartzitic complex	metabasite	Х	х		х		
AP6	Apuane Alps	Phyllitic and quartzitic	phyllite	Х	х		Х		
AP7	Apuane Alps	Phyllitic and quartzitic complex	phyllite	Х	х		х		

The complete bulk-rock and mineral composition dataset can be requested sending a mail to deborah.lopo@unibo.it.



EMP COMPOSITION AND AGE OF MONAZITE

Sample	Pontremoli7	1					
Location	Pontremoli	1 well					
Analysis	Mnz01_01	Mnz01_04	Mnz04_01	Mnz02_05	Mnz04_04	Mnz01_11	Mnz01_01a
SiO ₂	0.66	1.56	0.58	0.72	0.40	1.46	0.78
P_2O_5	29.07	29.72	29.34	29.40	28.13	29.99	27.44
SO_3	0.04	0.02	0.03	0.02	0.02	0.04	0.03
CaO	1.53	2.54	1.54	1.33	4.41	1.50	1.52
Y_2O_3	0.07	0.16	0.05	0.03	0.15	0.09	0.09
La_2O_3	13.72	10.00	14.56	15.31	13.62	13.53	14.01
Ce_2O_3	27.73	24.96	28.64	29.41	29.20	26.55	27.83
Pr_2O_3	3.01	2.50	3.05	3.06	2.24	2.96	2.93
Nd_2O_3	11.98	11.00	11.95	11.59	12.21	11.53	11.82
Sm_2O_3	2.17	1.66	2.11	1.60	0.86	1.93	2.01
Gd_2O_3	1.66	0.97	1.54	0.84	1.60	1.27	1.53
Dy_2O_3	0.23	0.22	0.20	0.12	0.00	0.19	0.25
PbO	0.13	0.05	0.10	0.09	0.02	0.08	0.12
ThO ₂	8.20	10.35	6.89	7.32	3.15	10.93	8.26
UO_2	0.45	0.29	0.42	0.38	0.45	0.26	0.49
Tot	100.65	96.00	100.99	101.22	96.46	102.31	99.11
Si	0.026	0.061	0.023	0.028	0.016	0.056	0.032
Р	0.969	0.989	0.972	0.971	0.958	0.968	0.945
S	0.001	0.001	0.001	0.001	0.001	0.001	0.001
Ca	0.064	0.107	0.065	0.056	0.190	0.061	0.066
Y	0.002	0.003	0.001	0.001	0.003	0.002	0.002
La	0.199	0.145	0.210	0.220	0.202	0.190	0.210
Ce	0.400	0.359	0.410	0.420	0.430	0.371	0.415
Pr	0.043	0.036	0.043	0.044	0.033	0.041	0.043
Nd	0.169	0.155	0.167	0.161	0.175	0.157	0.172
Sm	0.030	0.022	0.028	0.022	0.012	0.025	0.028
Gd	0.022	0.013	0.020	0.011	0.021	0.016	0.021
Dy	0.003	0.003	0.003	0.002	0.000	0.002	0.003
Pb	0.001	0.001	0.001	0.001	0.000	0.001	0.001
Th	0.074	0.093	0.061	0.065	0.029	0.095	0.076
U	0.004	0.003	0.004	0.003	0.004	0.002	0.004
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AGE	311.21	103.61	289.59	247.70	91.75	170.42	292.17
error	6.51	4.80	7.21	0.25	10.68	4.99	6.24

Sample	Pontremoli7					
Location	Pontremoli 1	well				
Analysis	Mnz01_02a	Mnz01_04a	Mnz01_05a	Mnz01_06a	Mnz01_07a	Mnz01_08a
SiO ₂	0.81	0.80	0.79	0.87	0.96	0.91
P_2O_5	27.30	26.28	26.75	30.17	29.76	29.80
SO ₃	0.02	0.04	0.03	0.03	0.03	0.06
CaO	1.58	1.78	1.80	1.75	1.86	1.88
Y_2O_3	0.09	0.08	0.05	0.06	0.06	0.05
La_2O_3	13.42	13.72	13.37	13.39	13.36	13.84
Ce_2O_3	27.53	26.80	26.66	26.88	26.87	26.70
Pr_2O_3	2.65	2.79	2.94	2.95	2.94	2.93
Nd_2O_3	11.48	11.12	11.55	11.80	11.65	11.33
Sm_2O_3	2.05	1.97	2.11	2.11	2.05	2.04
Gd_2O_3	1.57	1.50	1.50	1.53	1.53	1.46
Dy_2O_3	0.27	0.21	0.21	0.22	0.21	0.21
PbO	0.13	0.14	0.13	0.14	0.15	0.15
ThO_2	9.08	9.62	9.62	9.86	10.24	10.17
UO_2	0.48	0.50	0.50	0.47	0.50	0.52
Tot	98.46	97.35	98.02	102.24	102.18	102.05
Si	0.033	0.033	0.033	0.033	0.037	0.035
Р	0.946	0.931	0.937	0.979	0.970	0.972
S	0.001	0.001	0.001	0.001	0.001	0.002
Ca	0.069	0.080	0.080	0.072	0.077	0.077
Y	0.002	0.002	0.001	0.001	0.001	0.001
La	0.203	0.212	0.204	0.189	0.190	0.197
Ce	0.413	0.411	0.404	0.377	0.379	0.376
Pr	0.040	0.043	0.044	0.041	0.041	0.041
Nd	0.168	0.166	0.171	0.161	0.160	0.156
Sm	0.029	0.028	0.030	0.028	0.027	0.027
Gd	0.021	0.021	0.021	0.019	0.020	0.019
Dy	0.004	0.003	0.003	0.003	0.003	0.003
Pb	0.001	0.002	0.001	0.001	0.002	0.002
Th	0.085	0.092	0.091	0.086	0.090	0.089
U	0.004	0.005	0.005	0.004	0.004	0.004
		_	_	_	_	_
AGE	293.63	288.20	281.49	294.97	295.74	293.88
statistical error	5.88	5.58	5.70	5.66	5.51	5.49

Sample	Pontremoli7					
Location	Pontremoli 1	well				
Analysis	Mnz01_09a	Mnz02_01a	Mnz02_03a	Mnz02_04a	Mnz02_05a	Mnz02_06a
SiO_2	0.88	0.81	0.81	0.58	0.63	0.75
P_2O_5	30.10	30.00	29.48	29.98	30.33	27.16
SO_3	0.03	0.03	0.03	0.03	0.02	0.04
CaO	1.70	1.36	1.60	1.47	1.51	1.25
Y_2O_3	0.07	0.06	0.05	0.01	0.03	0.06
La_2O_3	13.73	14.81	14.10	15.39	14.86	14.12
Ce_2O_3	27.40	28.06	28.69	29.16	28.70	27.71
Pr_2O_3	3.01	3.10	3.04	3.04	3.06	2.73
Nd_2O_3	11.97	12.00	12.36	11.63	11.87	11.83
Sm_2O_3	2.12	2.07	2.07	1.54	1.90	1.98
Gd_2O_3	1.47	1.50	1.41	0.79	1.18	1.43
Dy_2O_3	0.21	0.20	0.20	0.11	0.17	0.28
PbO	0.11	0.10	0.11	0.11	0.12	0.13
ThO_2	9.49	6.66	8.26	7.44	7.69	7.08
UO_2	0.40	0.30	0.31	0.42	0.47	0.48
Tot	102.72	101.05	102.54	101.70	102.55	97.03
Si	0.034	0.031	0.031	0.022	0.024	0.031
Р	0.975	0.981	0.965	0.981	0.982	0.952
S	0.001	0.001	0.001	0.001	0.001	0.001
Ca	0.070	0.056	0.066	0.061	0.062	0.056
Y	0.002	0.001	0.001	0.000	0.001	0.001
La	0.194	0.211	0.201	0.219	0.210	0.215
Ce	0.384	0.397	0.406	0.412	0.402	0.420
Pr	0.042	0.044	0.043	0.043	0.043	0.041
Nd	0.164	0.166	0.171	0.161	0.162	0.175
Sm	0.028	0.027	0.028	0.021	0.025	0.028
Gd	0.019	0.019	0.018	0.010	0.015	0.020
Dy	0.003	0.002	0.002	0.001	0.002	0.004
Pb	0.001	0.001	0.001	0.001	0.001	0.001
Th	0.083	0.059	0.073	0.065	0.067	0.067
U	0.003	0.003	0.003	0.004	0.004	0.004
AGE	248.71	296.20	274.78	292.51	311.87	362.33
statistical error	5.64	7.68	6.50	6.81	6.69	7.19

Sample	Pontremoli7					
Location	Pontremoli 1	well				
Analysis	Mnz02_09a	Mnz03_06a	Mnz04_01a	Mnz04_03a	Mnz04_04a	Mnz04_05a
SiO ₂	0.71	1.42	0.54	0.52	0.49	0.55
P_2O_5	29.96	27.93	30.10	30.75	30.51	26.70
SO_3	0.02	0.03	0.03	0.03	0.02	0.01
CaO	1.42	1.15	1.57	1.50	1.40	1.82
Y_2O_3	0.04	0.40	0.06	0.09	0.09	0.09
La_2O_3	15.14	14.39	14.86	14.77	14.69	14.51
Ce_2O_3	29.34	29.72	29.14	28.94	29.02	28.95
Pr_2O_3	3.08	3.26	3.13	3.08	3.09	3.06
Nd_2O_3	11.50	13.23	12.08	12.05	12.19	12.26
Sm_2O_3	1.50	2.19	2.15	2.13	2.15	2.04
Gd_2O_3	0.86	1.32	1.60	1.60	1.59	1.40
Dy_2O_3	0.10	0.38	0.21	0.24	0.27	0.26
PbO	0.09	0.04	0.10	0.09	0.09	0.07
ThO ₂	7.69	4.30	6.66	6.24	5.87	5.29
UO_2	0.39	0.04	0.43	0.42	0.43	0.33
Tot	101.84	99.82	102.65	102.45	101.88	97.33
Si	0.027	0.057	0.021	0.020	0.019	0.023
Р	0.978	0.939	0.978	0.991	0.990	0.939
S	0.001	0.001	0.001	0.001	0.001	0.000
Ca	0.059	0.049	0.064	0.061	0.057	0.081
Y	0.001	0.008	0.001	0.002	0.002	0.002
La	0.215	0.211	0.210	0.207	0.208	0.222
Ce	0.414	0.432	0.409	0.403	0.407	0.440
Pr	0.043	0.047	0.044	0.043	0.043	0.046
Nd	0.158	0.188	0.166	0.164	0.167	0.182
Sm	0.020	0.030	0.028	0.028	0.028	0.029
Gd	0.011	0.017	0.020	0.020	0.020	0.019
Dy	0.001	0.005	0.003	0.003	0.003	0.003
Pb	0.001	0.000	0.001	0.001	0.001	0.00
Th	0.067	0.039	0.058	0.054	0.051	0.050
U	0.003	0.000	0.004	0.004	0.004	0.003
AGE	249.86	189.74	293.63	291.96	307.32	277.9
statistical error	6.50	11.96	7.35	7.70	8.01	8.6

Sample	Pontremoli7			
Location	Pontremoli 1	well		
Analysis	Mnz04_06a	Mnz04_07a	Mnz04_08a	Mnz04_02a
SiO ₂	0.52	0.50	0.48	0.61
P_2O_5	30.68	30.77	30.44	29.96
SO ₃	0.02	0.01	0.03	0.04
CaO	1.50	1.58	1.58	1.67
Y_2O_3	0.10	0.10	0.10	0.05
La_2O_3	14.48	14.67	14.89	14.97
Ce_2O_3	29.13	29.10	29.58	29.20
Pr_2O_3	3.16	3.20	3.16	3.10
Nd_2O_3	12.17	12.26	12.44	12.33
Sm_2O_3	2.12	2.14	2.15	2.14
Gd_2O_3	1.55	1.55	1.57	1.57
Dy_2O_3	0.25	0.27	0.26	0.20
PbO	0.09	0.10	0.09	0.11
ThO ₂	5.99	5.76	5.60	7.01
UO_2	0.40	0.43	0.43	0.42
Tot	102.16	102.43	102.80	103.38
Si	0.020	0.019	0.018	0.023
Р	0.991	0.991	0.983	0.970
S	0.001	0.000	0.001	0.001
Ca	0.061	0.064	0.064	0.068
Y	0.002	0.002	0.002	0.001
La	0.204	0.206	0.209	0.211
Ce	0.407	0.405	0.413	0.409
Pr	0.044	0.044	0.044	0.043
Nd	0.166	0.167	0.169	0.168
Sm	0.028	0.028	0.028	0.028
Gd	0.020	0.020	0.020	0.020
Dy	0.003	0.003	0.003	0.002
Pb	0.001	0.001	0.001	0.001
Th	0.052	0.050	0.049	0.061
U	0.003	0.004	0.004	0.004
AGE	295.57	315.43	301.51	311.04
statistical error	7.99	8.18	8.27	33.11

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ACKNOWLEDGMENTS

At the beginning of the PhD I didn't know that there a Variscan basement in Northern Apennine existed. At the end of this adventure I know that it really exists and it is also quite complicated to deal with.

My challenging study has benefited from the help of numerous people.

First of all my supervisor, Roberto Braga, a scientist who considers his job as his hobby. He is fully and madly enthusiastic when handling rocks. I already know that whatever I write about him he will make fun of me. Anyway I have to thank him for many reasons. First he accepted to be my supervisor even if he didn't know anything about me. Furthermore he modified his approach to the coordination of my activity when it was necessary. Moreover he has been for me an example of how a researcher driven by passion and fervour works. I really enjoyed our friendly discussions about metamorphic petrology. I hope to have paid you back for the time and the patience you dedicated me.

I would also like to thank Professor G. Molli (University of Pisa) and Professor A. Montanini (University of Parma) who helped me to move my first steps into the Northern Apennine geology. They followed my progresses through these 3 years and accepted to be part of the reading committee of my thesis. Their suggestions are among the lines of the entire thesis.

I'd like to thank also Dr. G. Di Vincenzo (CNR-Pisa) who introduced me to the Ar-Ar dating world.

I am grateful to Professor H. J. Massonne and Dr. T. Theye (University of Stuttgart) for access to the microprobe and for the useful discussions about my samples. I thank them also for providing me the possibility to work in Stuttgart for 6 months.

I thank also Alan Santoro, who shared with me the experience of tick bites during the fieldwork in the Cerreto Pass.

Professor G. M. Bargossi, G. Gasparotto and E. Dinelli (University of Bologna) provided additional analytical support and interesting ideas.

Thanks also to the PhD commission, that supervised the evolution of my research year after year.

I am thankful to my two thesis reviewers, Professor A. Willner (University of Stuttgart) and Professor G. Godard (Institut de Physique du Globe de Paris), who carefully read my work and suggested me improvements for a better product.

I thank the University of Bologna which financed this study.

I thank also the SIMP (Italian Society of Mineralogy and Petrology) which offered me a grant for participating in the 29th Himalaya-Karakoram-Tibet workshop (Lucca, Italy). The Marco Polo project (University of Bologna) provided the founds for my research period in Stuttgart.

In Stuttgart I had a great time with the Institute team, Elena Spano, Salvatore Iaccarino, Li Botao, Sergio Baggio. Thank you guys for many great dinners all around Stuttgart. You contributed to make me feel very good during those six months.

During my stay in Bologna I met new friends. Erika Brattich, Giorgia Cinelli, Rossella Raimondo, Bibiana Förster, Maurizio Sansonetti, Valerio Funari, Gloria Acciari, Daniele Redi, Claudia Piscionieri, Alberto Previti and others deserve special thanks for encouraging me. Guys, we shared pleasant pieces of life!

A special mention goes to my friends from Sicily: Alessia Gugliotta, Catia Tarsia, Federica Mazzeo, Federica Zangara, Daniele Maglitto, Alessandra Arena, Manuele Carciotto and others. Your help has been invaluable even if now a thousand of km separates us.

My family supported me despite the long distance.

I finish my acknowledgments with a big thank to Mr. Opitz, who taught me to enjoy the sunshine everyday.

CURRICULUM VITAE



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